

# On-line Monitoring Development in Support of the Nuclear Fuel Cycle

Dr. Amanda Lines and Dr. Samuel Bryan

Pacific Northwest National Laboratory, USA

31 July 2024

# Some Housekeeping Items



Listen through your computer

Please select the “mic and speakers” radio button on the right-hand audio and pane display



Technical Difficulties

Search the Go To Webinars Support:  
<https://support.goto.com/webinar>



To ask a question

Select the “Questions” pane on your screen and type in your question



Share with others or watch it again

A video/audio recording of the webinar and the slide deck will be made available at [www.gen-4.org](http://www.gen-4.org)



Please take the survey

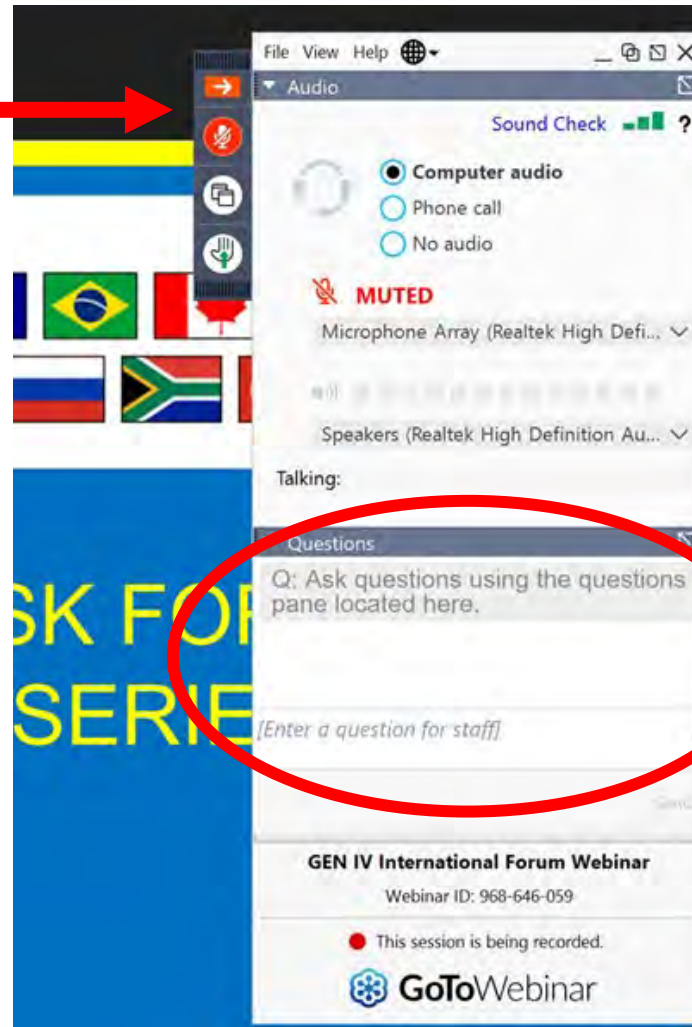
A brief online survey will follow the webinar.

# To Ask a Question

Click Here



To Open the  
Go To  
Webinar  
Control Panel



Write question in bottom box  
and hit "Send"

# On-line Monitoring Development in Support of the Nuclear Fuel Cycle

Dr. Amanda Lines and Dr. Samuel Bryan  
Pacific Northwest National Laboratory, USA  
31 July 2024

## Meet the Presenters

**Dr. Amanda Lines** is a Chief Scientist at Pacific Northwest National Laboratory with experience in the design and deployment of on-line monitoring and sensor technology. Dr. Lines graduated from Washington State University with her Ph.D. in analytical chemistry after earning her undergraduate degrees in chemistry and chemical engineering from Purdue University. Her work primarily focusses on the development of optical sensors for highly harsh environments, such as those common to nuclear materials processing, as well as the application of chemical data science tools to enable advanced, automated data output.



**Dr. Samuel A. Bryan** is a Laboratory Fellow in Nuclear Chemistry and Engineering at the Pacific Northwest National Laboratory (PNNL). Dr. Bryan joined PNNL in 1990, and has over 35 years' experience in optical spectroscopy, electrochemistry, and separations science. Bryan's research interests involve the design and development of spectroscopic and spectroelectrochemical sensors for the measurement of actinides, lanthanides, fission products and transition metal complexes in aqueous and molten salt media. He also serves as an Adjoint Faculty member in the Department of Chemistry at Washington State University.

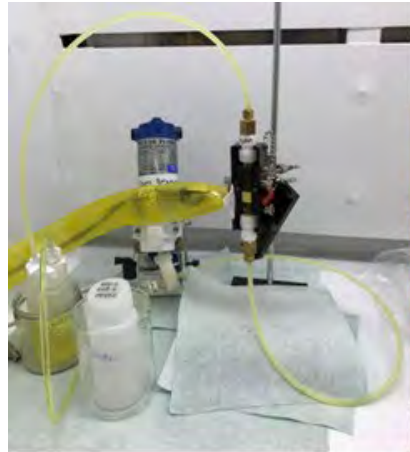


# The Benefits of *In Situ*, Real-time Monitoring

- Gaining unapparelled insight into fundamentals
- Improving the route of designing and optimizing new processes
- Enabling informed scale up
- Supporting better, faster, safer, and more cost-effective deployment of nuclear material processing



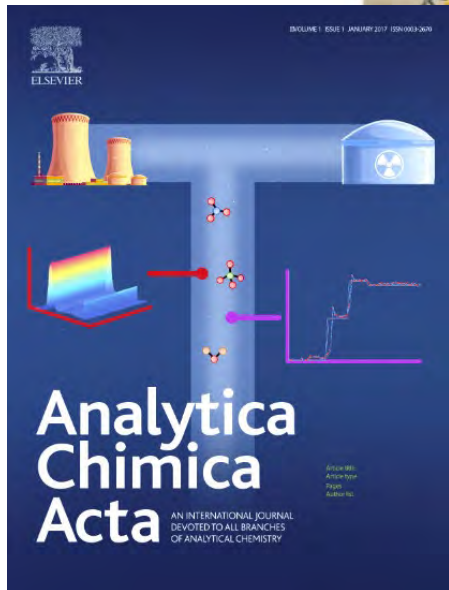
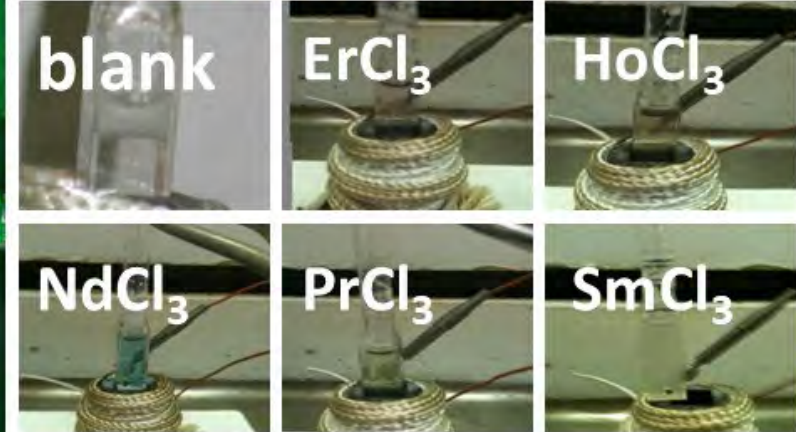
# How is the Fuel Cycle Benefited?



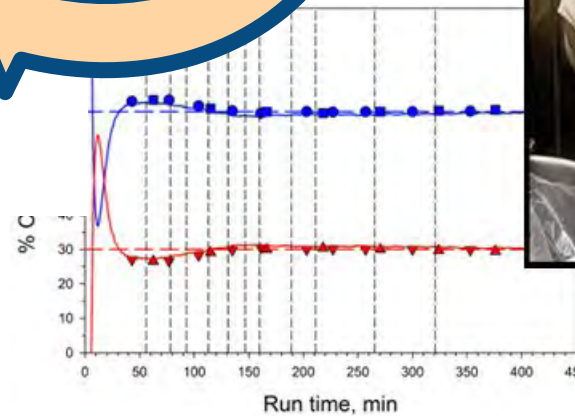
## FUEL PRODUCTION



## REACTOR OPERATION



## WASTE PROCESSING

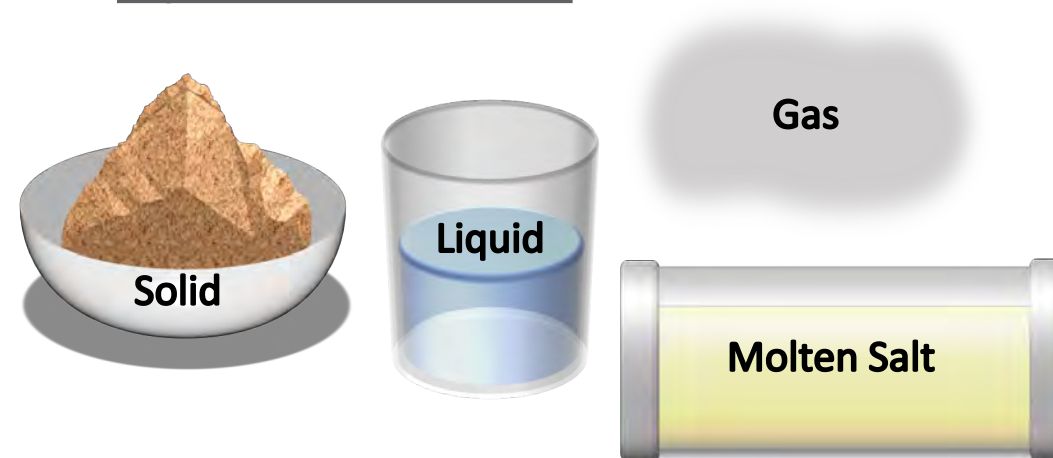


## FUEL RECYCLE

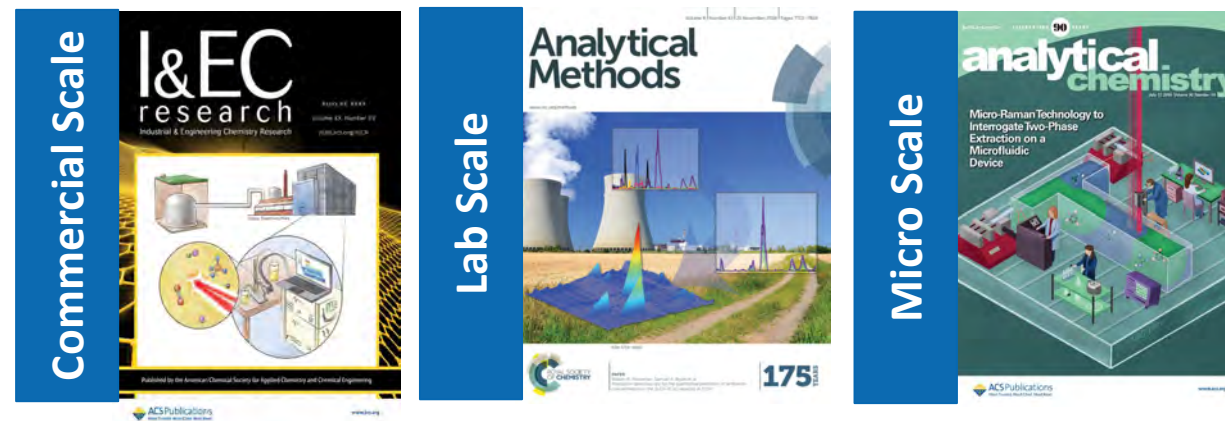
# Many Tools and Sensors are Available: Focus on Optical Spectroscopy for Chemical Composition Analysis

- Provides chemical information
  - Identification and quantification
  - Oxidation state
    - Essential information for control of systems
  - Molecular and elemental species
    - Essential information to control system behavior
- Highly mature technology
- Simplistic integration
- Versatile

## System matrix



## System scale





# Optical spectroscopy comes in many forms, allowing for analysis of a huge range of chemical targets

## Raman spectroscopy

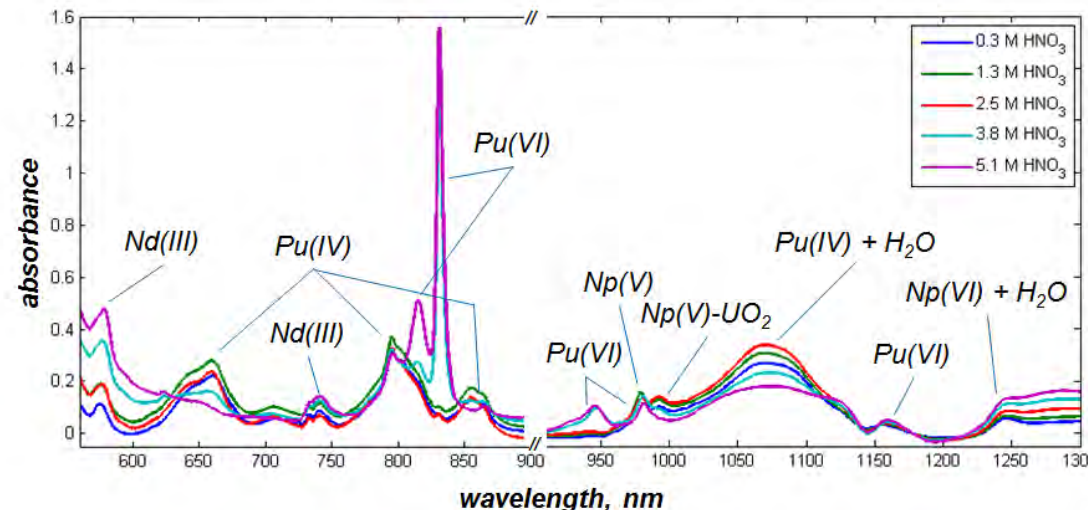
- Actinide oxide ions ( $\text{UO}_2^{2+}$ )
- Organics: solvent components and complexants
- Inorganic oxo-anions ( $\text{NO}_3^-$ ,  $\text{CO}_3^{2-}$ ,  $\text{OH}^-$ ,  $\text{SO}_4^{2-}$ )
- Water, acid ( $\text{H}^+$ ), base ( $\text{OH}^-$ )
- pH of weak acid buffer systems

## UV-vis-NIR absorption

- Actinides and lanthanides in multiple oxidation states
  - Pu (III/IV/VI)
  - Np (III/IV/V/VI)
- Various metal-ligand complexes

## Several other options

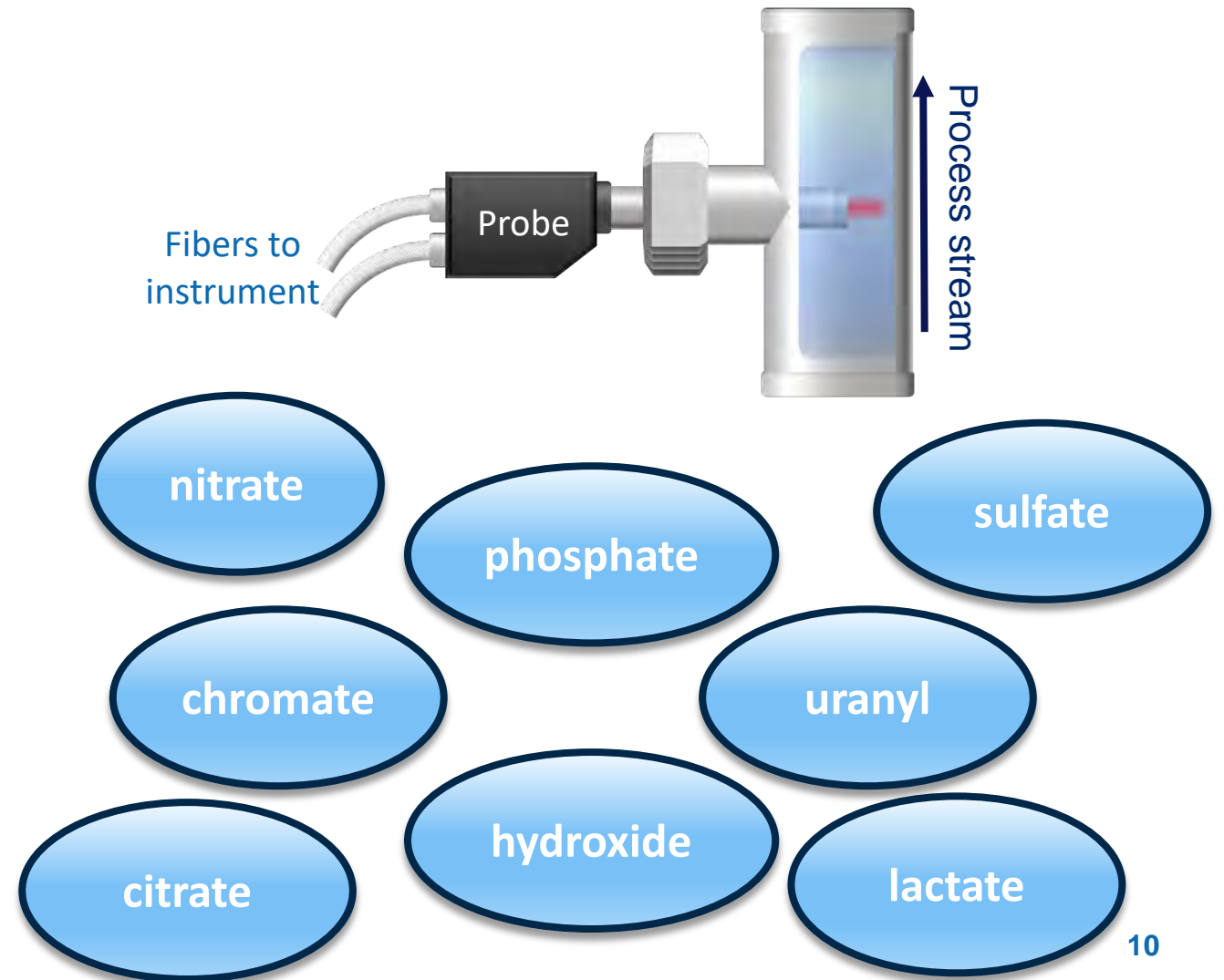
- FTIR
  - Organic complexants
- Light scatter
  - turbidity
- Optical density
  - Formation of complexes



Numerous, versatile tools available to capture fingerprints of huge range of fission products/species of interest to the fuel cycle

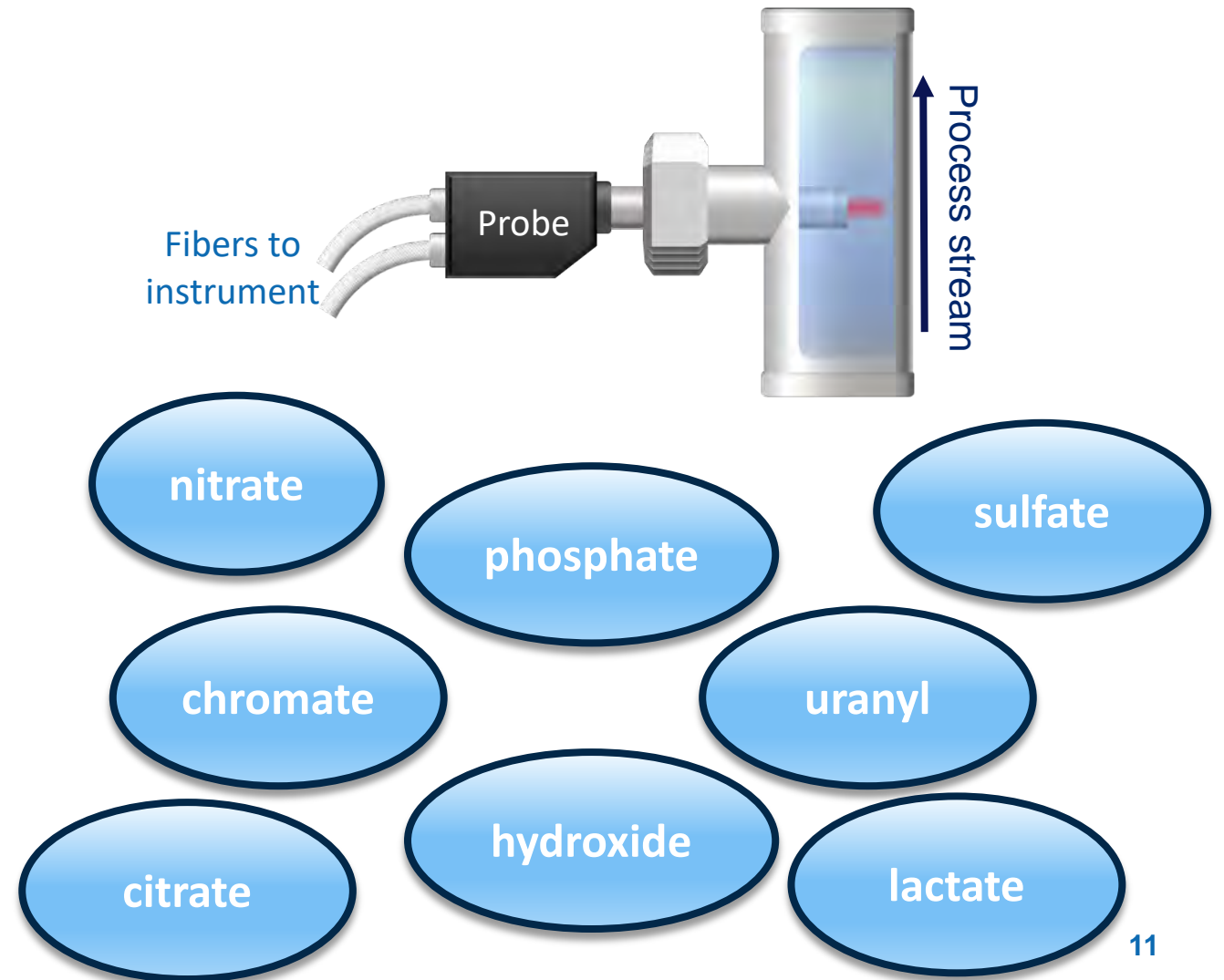
# The Greatest Strength of Optical Spectroscopy

- A wide range of optical tools are available, and can provide pathways to characterize almost any analytical target
- For examples such as Raman and UV-vis absorbance, numerous metal, organic, etc. species can be characterized with a single in-process probe



# The Greatest **Weakness** of Optical Spectroscopy?

- A wide range of optical tools are available, and can provide pathways to characterize almost any analytical target
- For examples such as Raman and UV-vis absorbance, numerous metal, organic, etc. species can be characterized with a single in-process probe



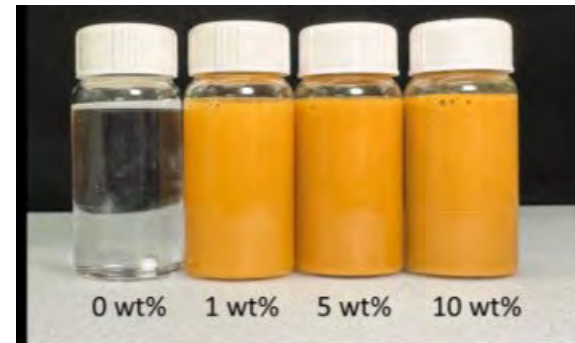
# Is Chemical Complexity the Only Challenge We Face in Applications to the Nuclear Fuel Cycle?

## Environments that damage sensors

- Highly corrosive or chemically destructive environments
- High temperatures (molten salts)
- High radiation

Example window material

0 Rad      1.7E8 Rad

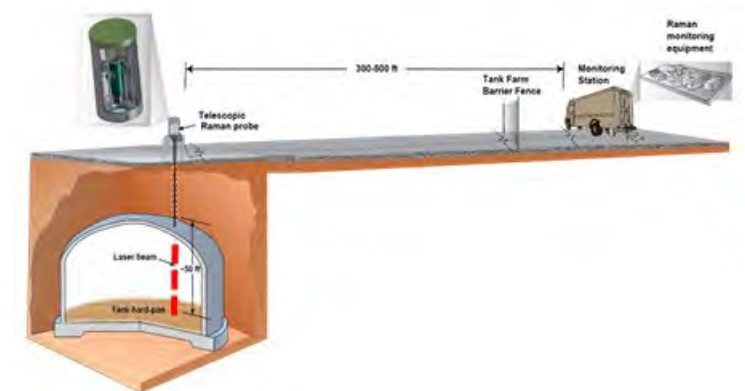


## Chemistry or process conditions that complicate analysis

- Variable turbidity
- Highly sensitive matrix effects (e.g., speciation in variable pH processes)
- Confounded signals

## Real processes where frequent calibrations or probe maintenance is not realistic

- Processes that cannot be accessed due to hazards

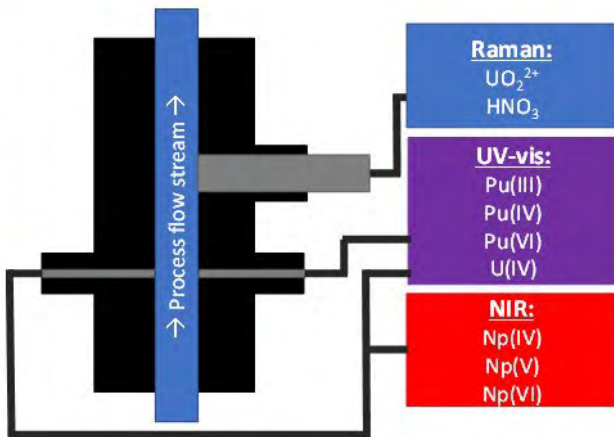


# The Two Prongs of Focus

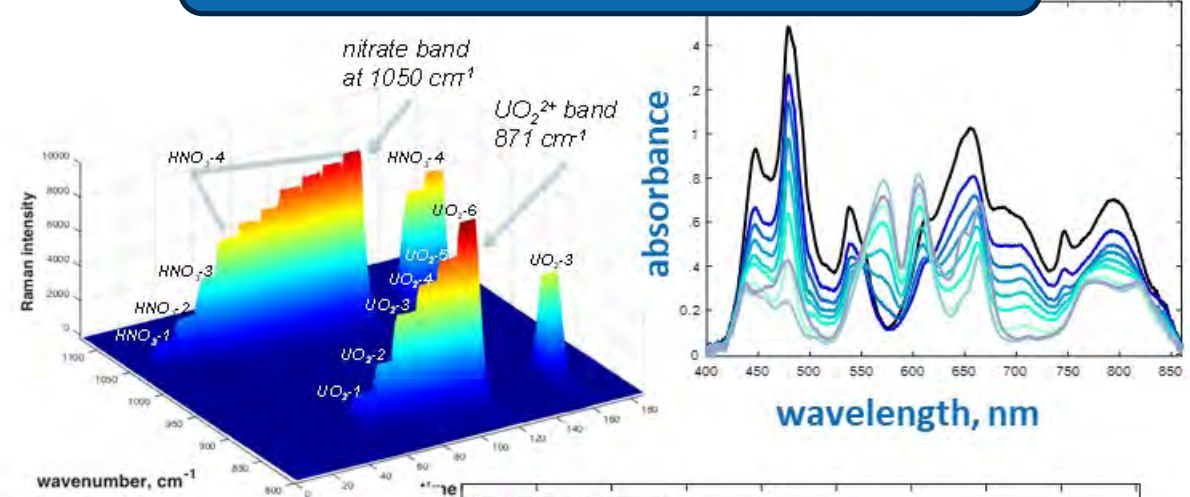
## Sensor Hardening and Design



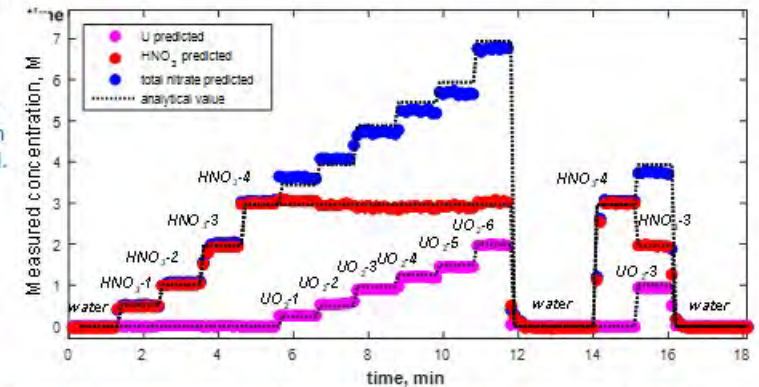
405 nm 532 nm 671 nm



## Making Sensors Smart



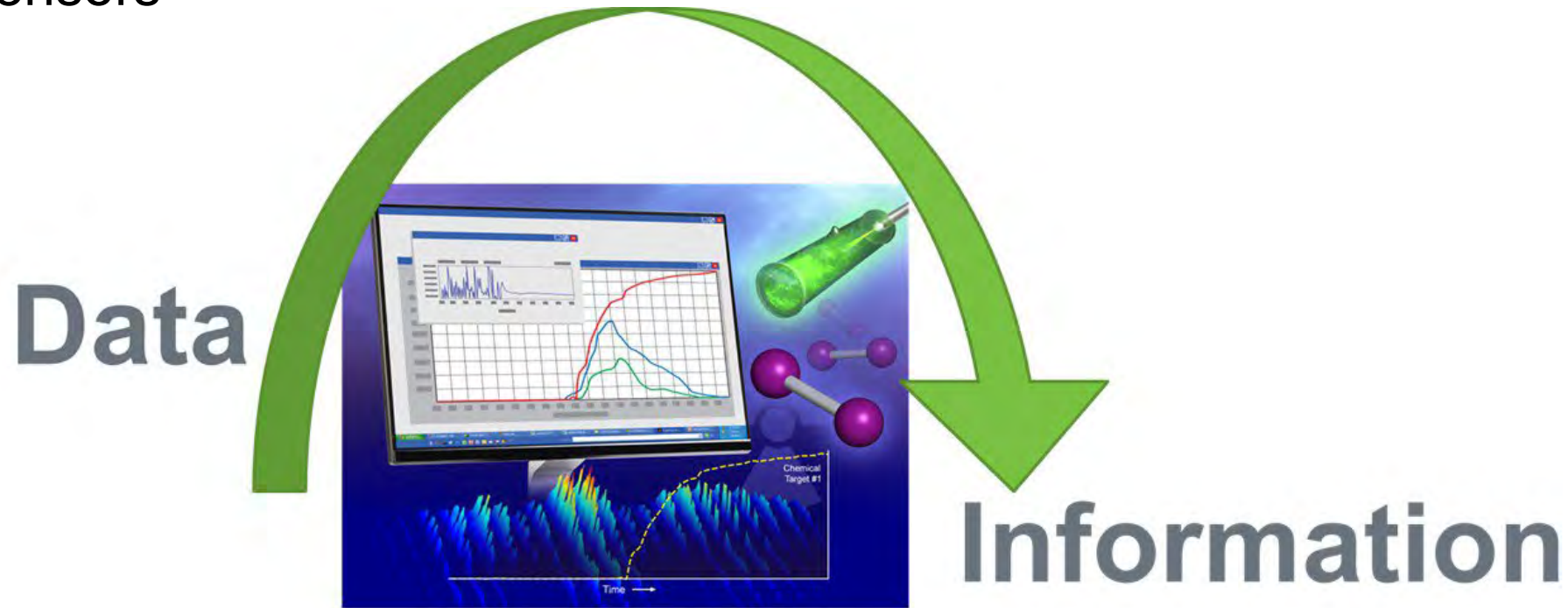
Lines, A.M., S.R. Adami, S.I. Sinkov, G.J. Lumetta, and S.A. Bryan. 2017. "Multivariate analysis for quantification of plutonium (IV) in nitric acid based on absorption spectra." *Anal. Chem.*, 89(14):9354-9359, DOI: 10.1021/acs.analchem.7b02161.



Lines, Bello, Clark, Bryan. Multivariate analysis to quantify species in the presence of direct interferences: micro-Raman analysis of  $HNO_3$  in microfluidic environments. *Anal. Chem.* 2018

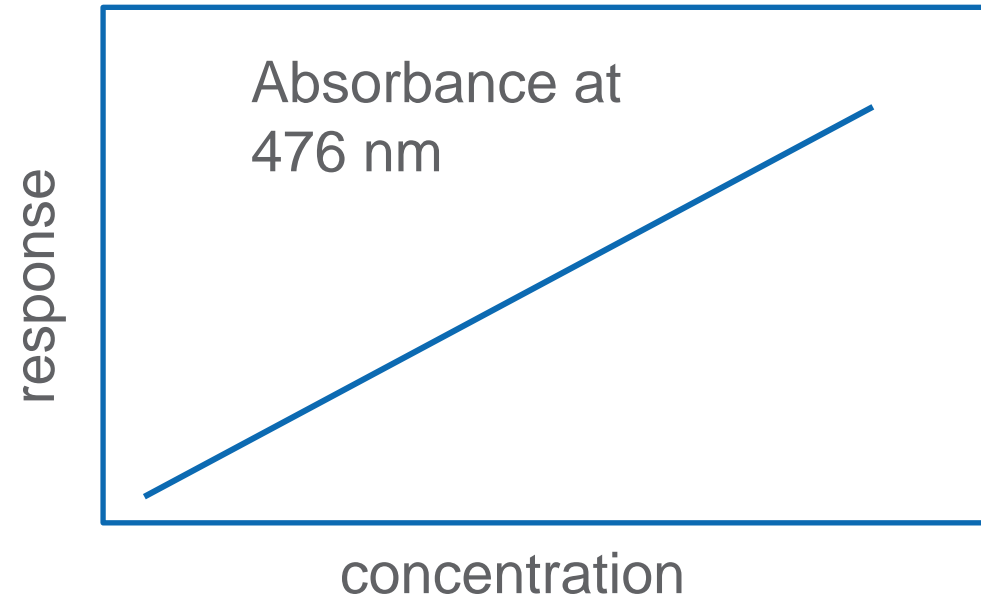
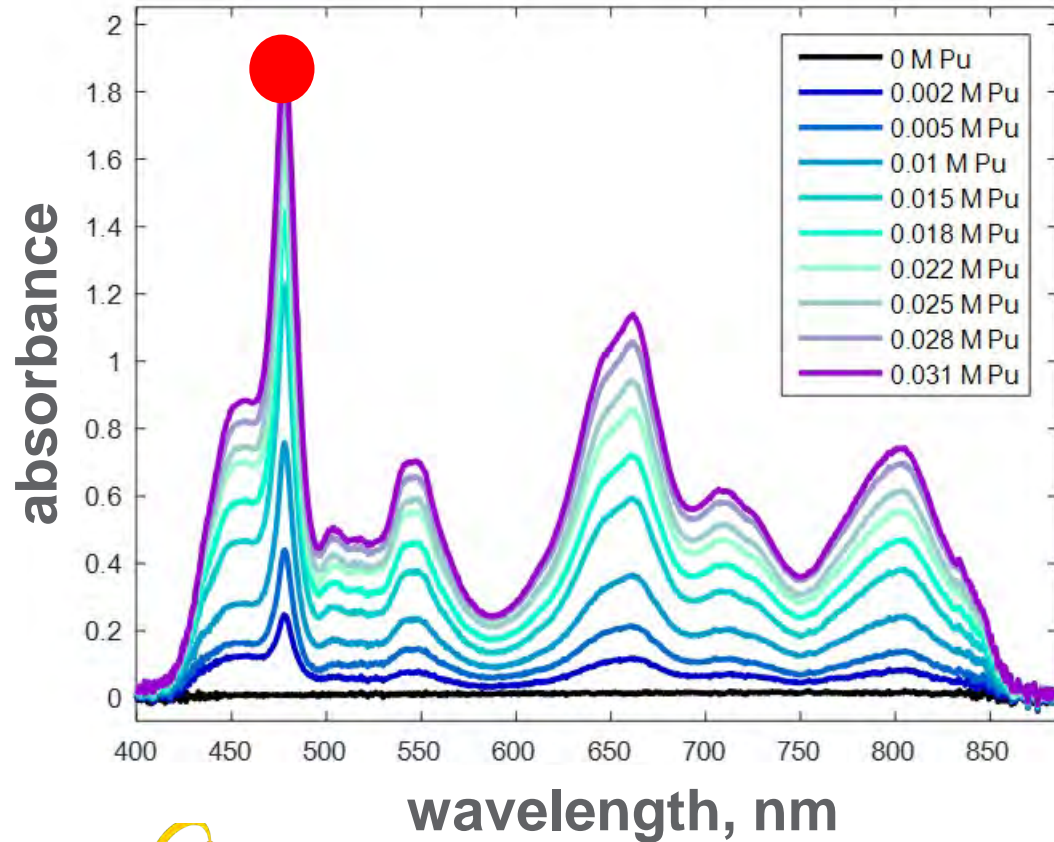
# Why We Need Advanced Tools for Data Analysis

Integrating advanced tools for spectral analysis improves not only accuracy, but also speed of obtaining results, BUT most importantly expands the applicability of our sensors

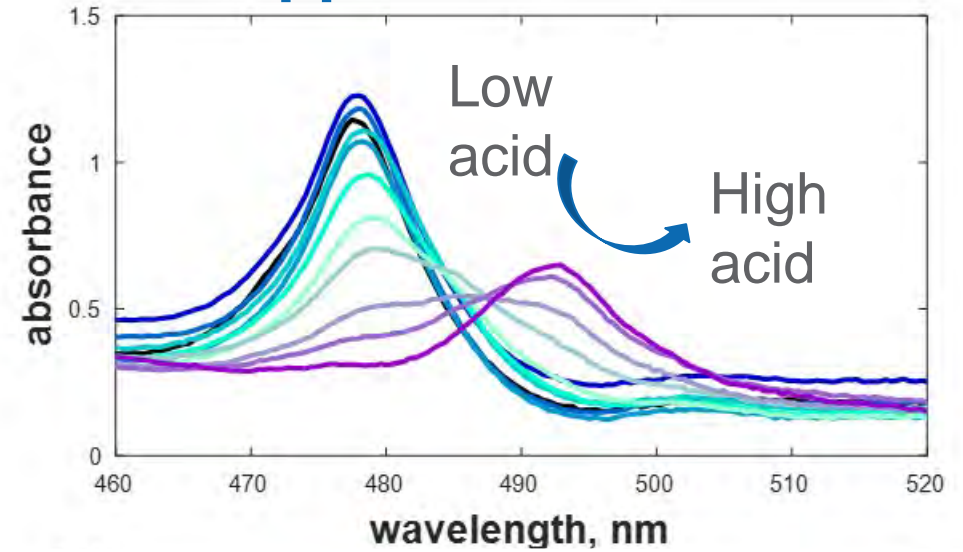
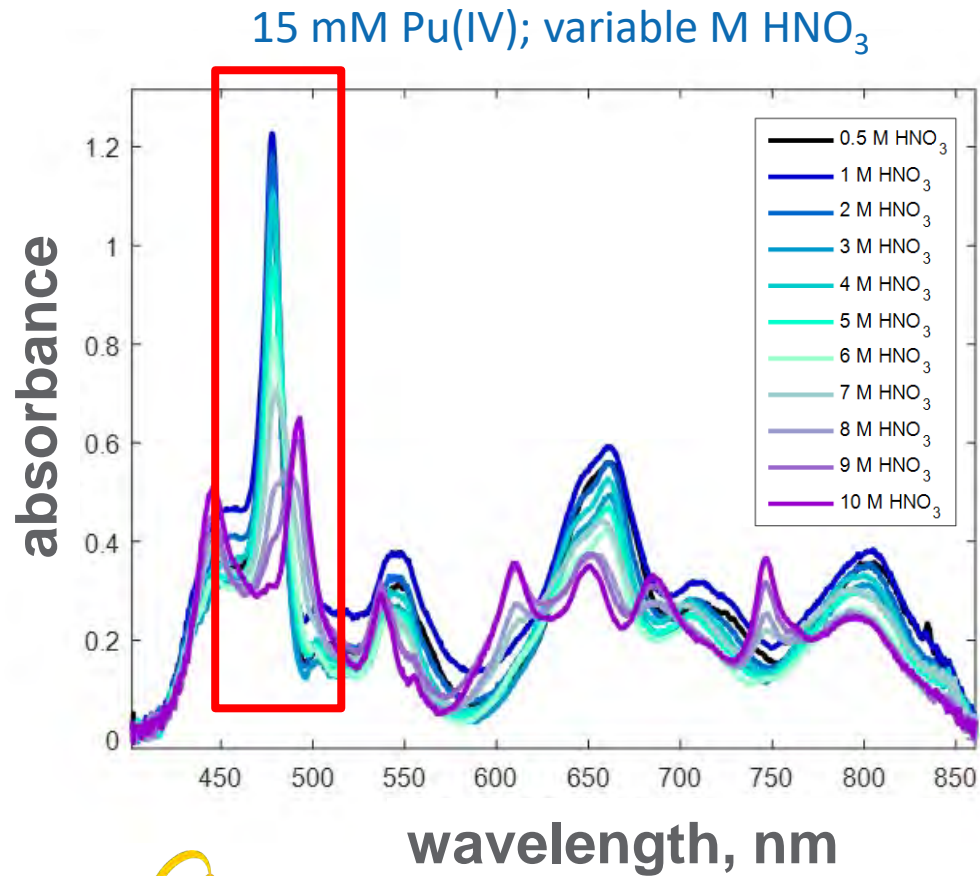


# How Chemometric Modeling Works: Graphical Example of Chemical Data Science Application

variable Pu(IV); 1 M HNO<sub>3</sub>



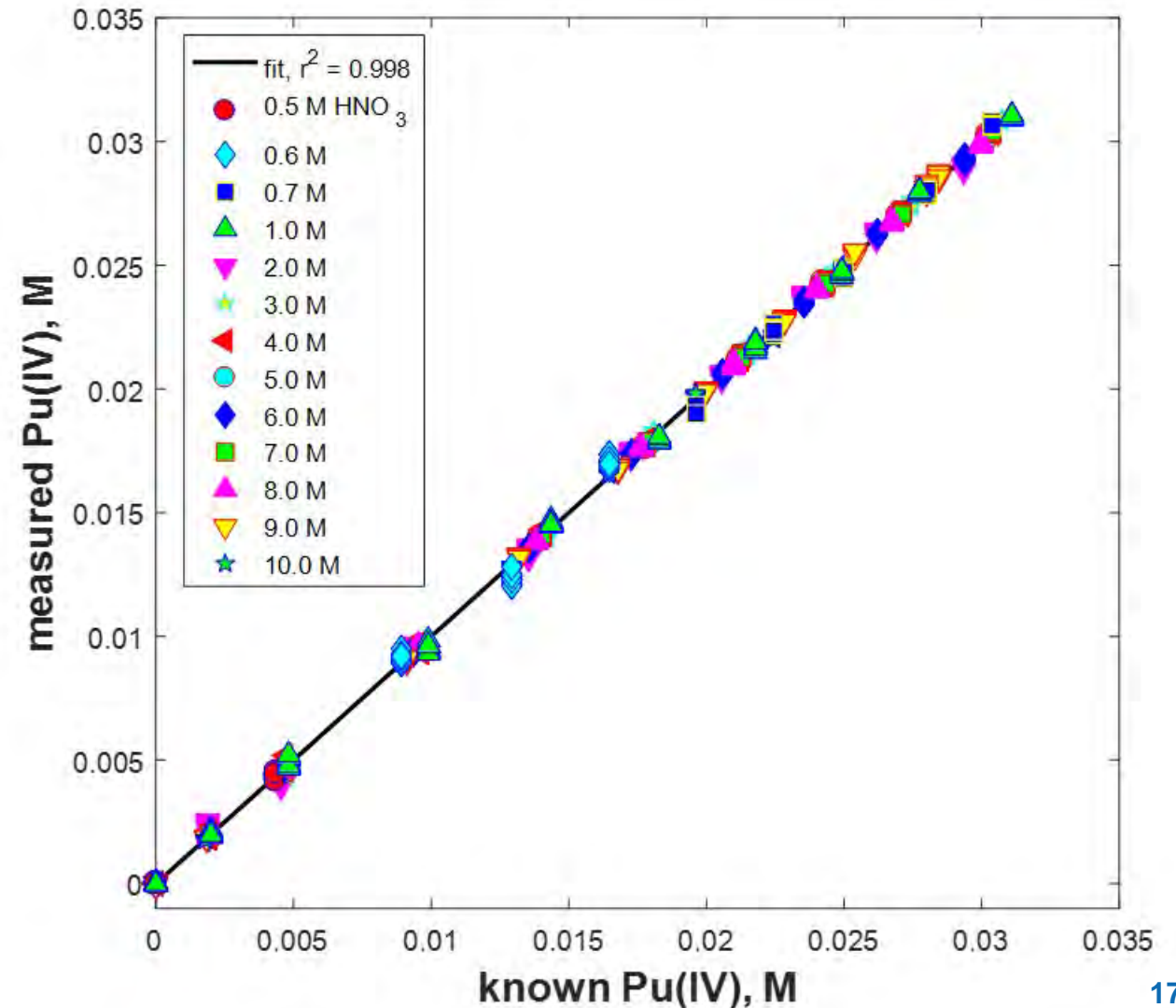
# How Chemometric Modeling Works: Graphical Example of Chemical Data Science Application





# How Chemometric Modeling Works: Graphical Example of Chemical Data Science Application

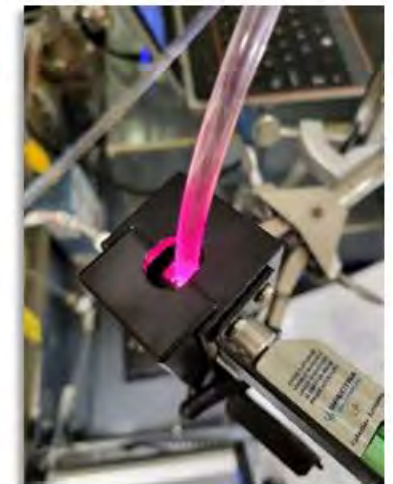
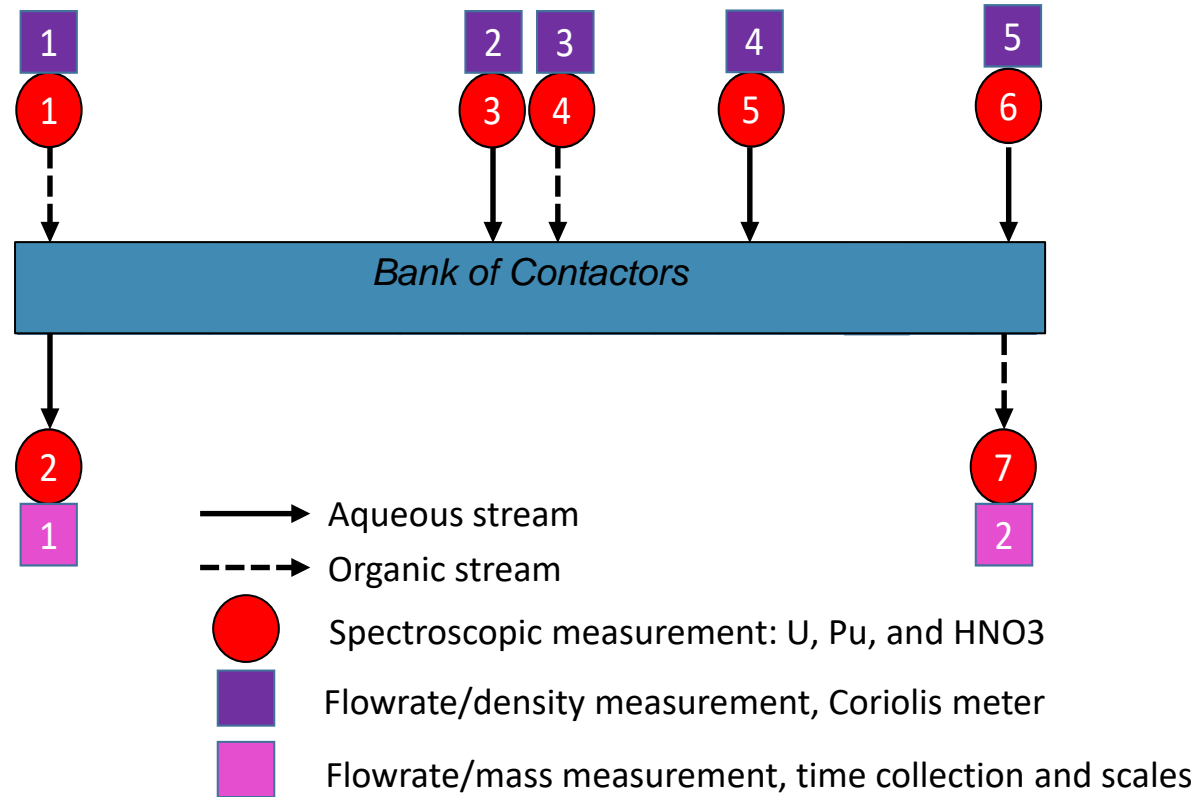
- Highly accurate quantification despite wide range in Pu and HNO<sub>3</sub> concentrations
- Multiple applications beyond quantification as shown here



## Exploring Challenging Applications and Demonstrations: both in sensor design and analytical tool development

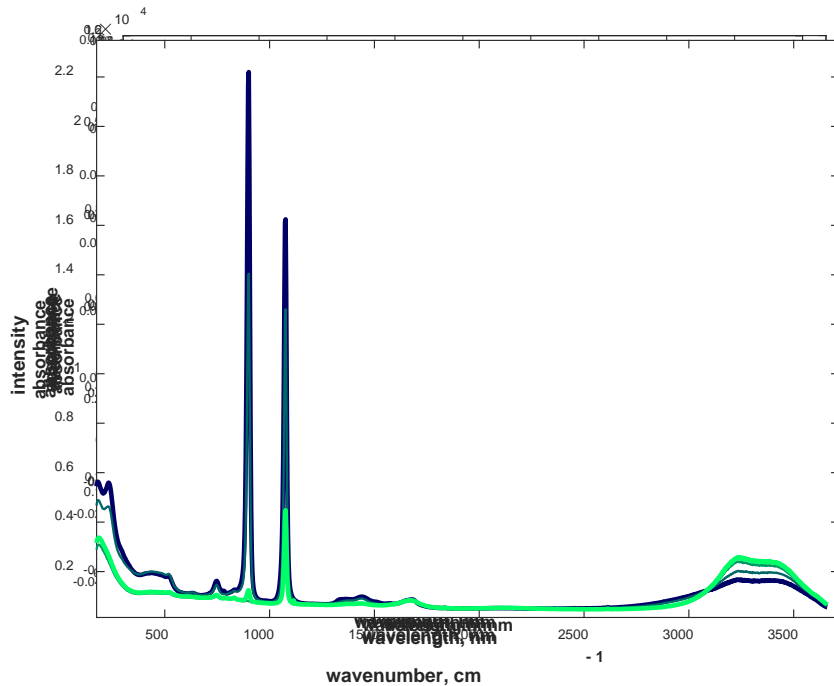
- Highly complex chemical compositions (many interferences to optical fingerprints or other matrix effects)
- Extremely harsh environments that impact sensors and therefore impact optical outputs
- Highly variable processes that require robust approaches to data analysis
- Demonstrations of real-time process control (manual and automated)

# CoDCon Application: Used Nuclear Fuel Recycle



# CoDCon Application: Used Nuclear Fuel Recycle

## Raman

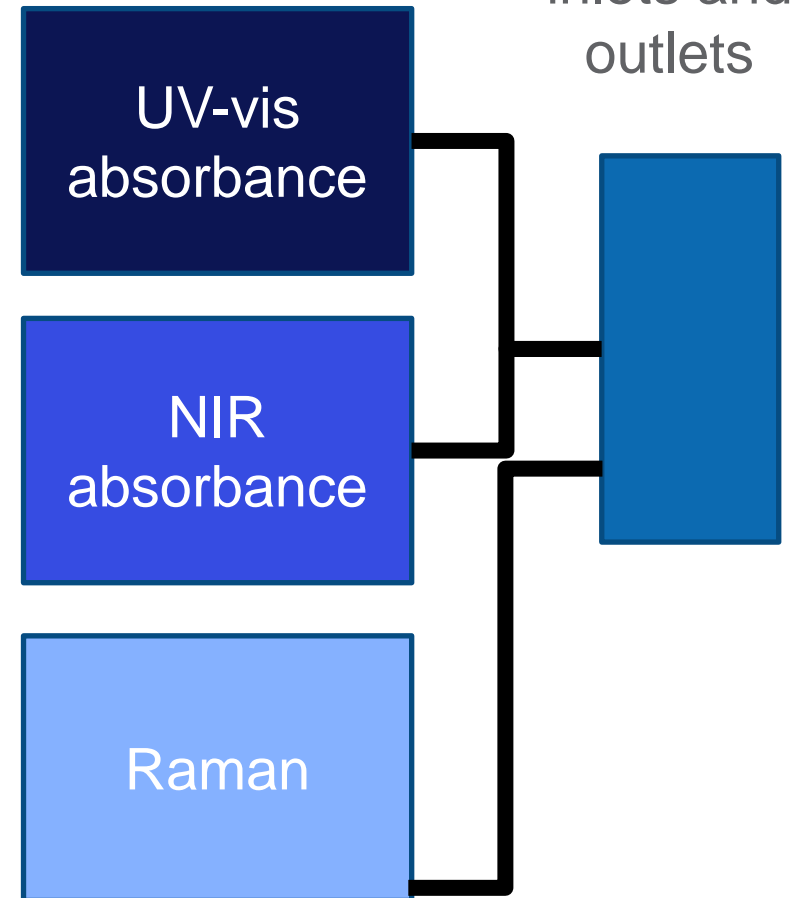


Fission products

Pu: 3+, 4+, 6+  
U: 4+

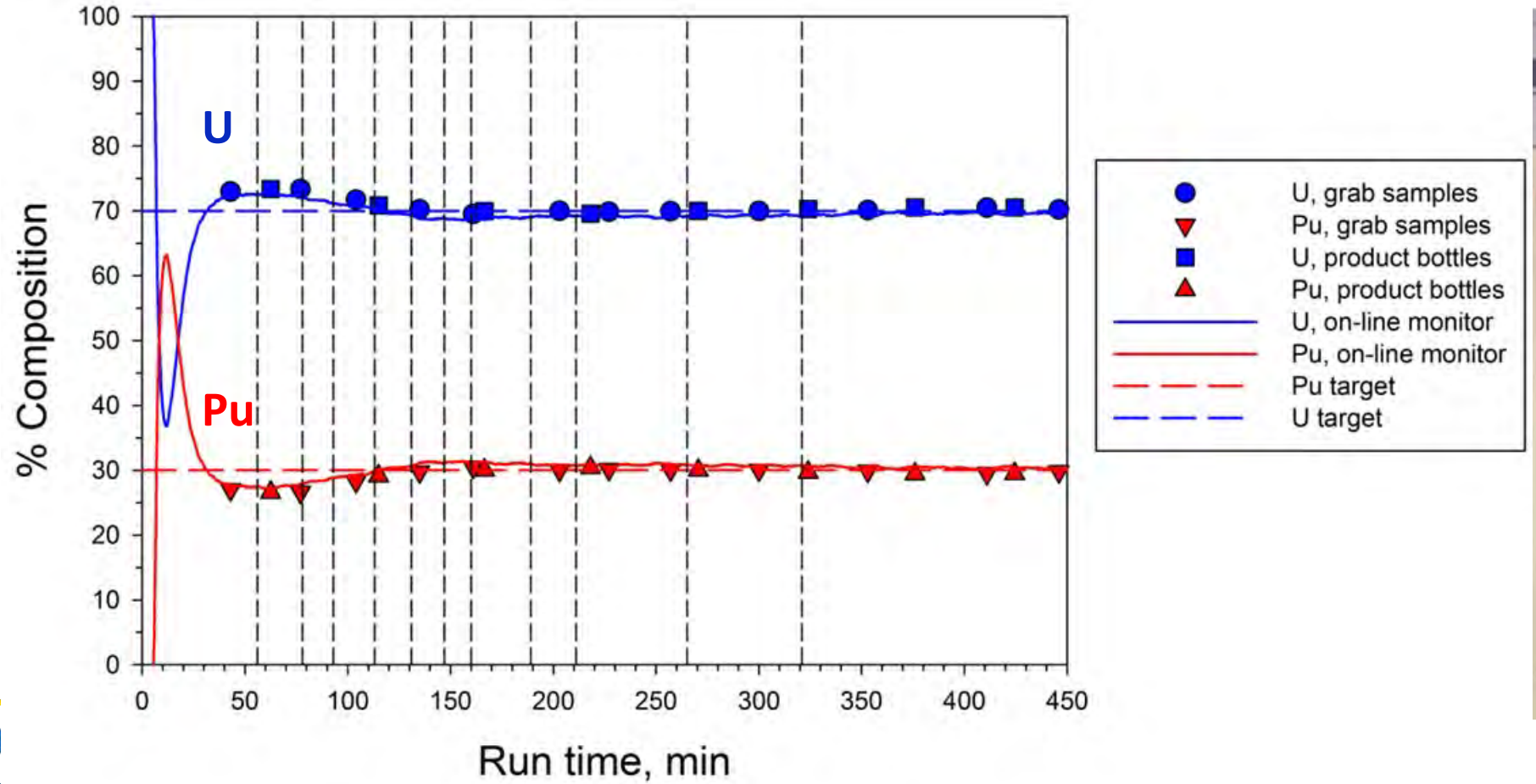
Np: 4+, 5+, 6+

Uranyl  
HNO<sub>3</sub>



Flow cell  
installed  
on all  
inlets and  
outlets

# CoDCon Application: Used Nuclear Fuel Recycle



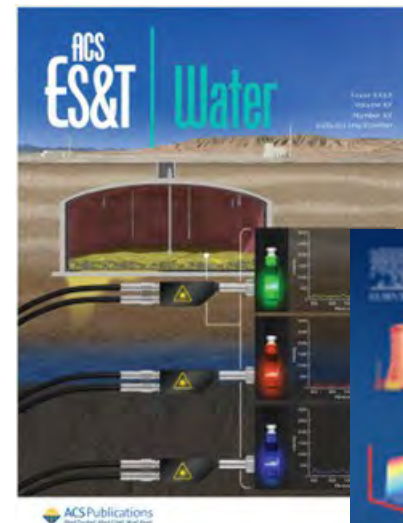
# Hanford waste applications

- Legacy waste clean-up efforts present some of the most challenging process conditions
  - Often not a great record of what is in the waste
  - Conditions can be very harsh
- For sites such as Hanford, this is particularly true
  - Millions of gallons of radioactive waste that was produced during plutonium production and recovery
  - Chemically complex
  - Highly turbid



# Hanford waste applications: past demonstrations

- Hanford tank salt cake retrieval (2002-2005)
  - Flow measurements, up to 40 gal/min
  - 8 analytes by Raman:  $\text{NO}_3^-$ ,  $\text{NO}_2^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{PO}_4^{3-}$ ,  $\text{AlO}_2^-$ ,  $\text{CO}_3^{2-}$ ,  $\text{CrO}_4^{2-}$ ,  $\text{OH}^-$
- Pretreatment Engineering Platform (PEP-WTP) Project (2008-2009)
  - Discrete sample analysis (9 analytes by Raman,  $\text{C}_2\text{O}_4^{2-}$  added)
  - Full QA/QC HASQARD compliant procedure
- Near-Tank Treatment System: Pilot Scale CSL Testing (2011)
  - Real-time monitoring during continuous sludge leaching (CSL) campaign February - April (~2.5 months)
  - Measured Al and  $\text{OH}^-$  in real-time during 2.5 mo. processing



Felmy, Heather; Lackey, Hope; Schafer Medina, Adan; Minette, Michael; Bryan, Samuel; Lines, Amanda, "Leveraging multiple Raman excitation wavelength systems for process monitoring of nuclear waste streams", accepted, February 11, 2022, *ACS ES&T Water*. DOI: [10.1021/acsestwater.1c00408](https://doi.org/10.1021/acsestwater.1c00408)

Tse, P., N.P. Bessen, S.A. Bryan, A. M. Lines, J.C. Shafer. "Review of On-line and Near Real Time Spectroscopic Monitoring of Processes Relevant to Nuclear Material Management" *Analytica Chimica Acta*, 2020, 1107:1-13

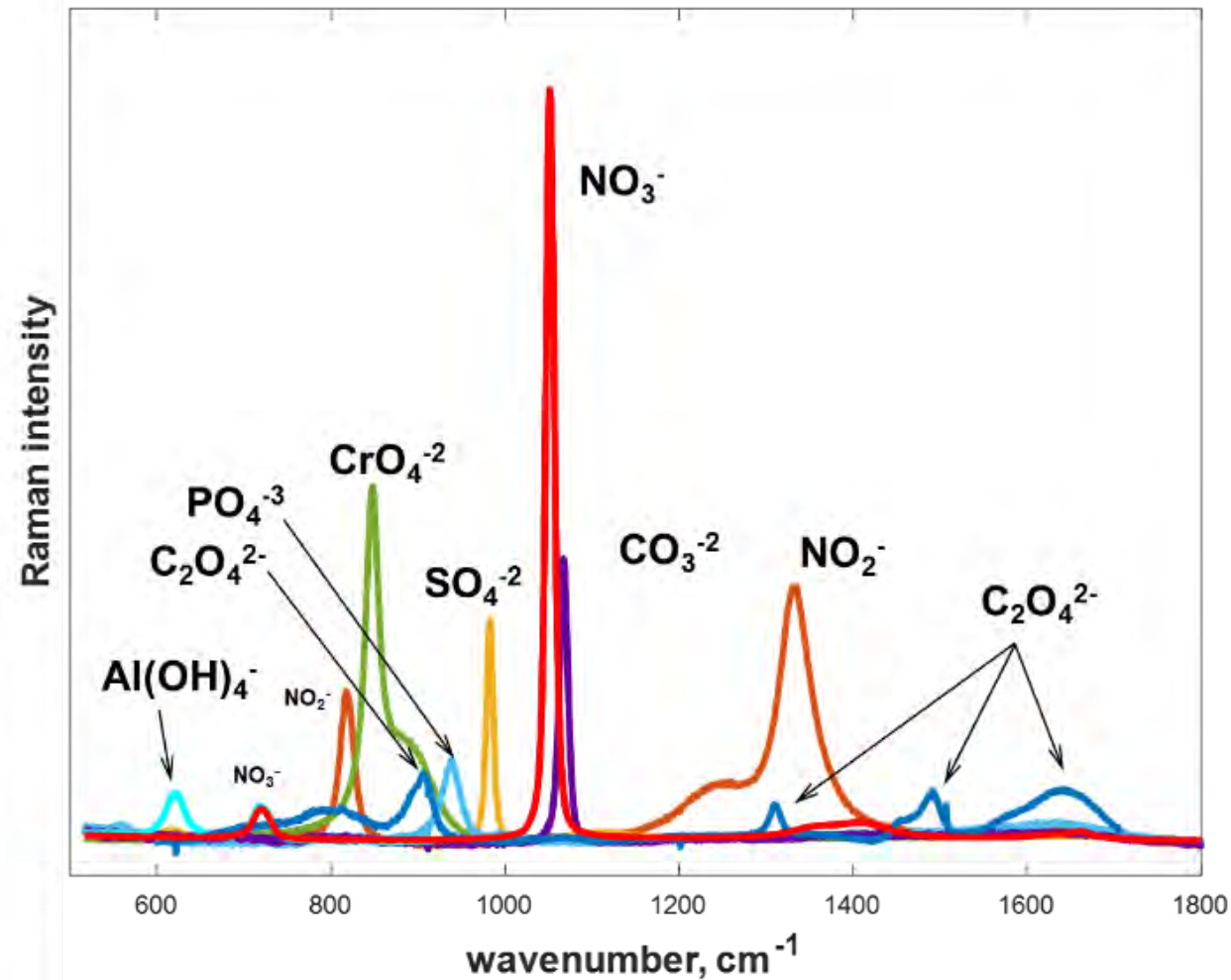


Lines, A.M., P. Tse, H.M. Felmy, J.D. Wilson, J. Shafer, K. Denslow, A.N. Still, C.K. King, S.A. Bryan. "On-line, real-time analysis of highly complex processing streams: Quantification of analytes in Hanford tank sample", *Ind. Eng. Chem. Res.*, 2019, 58, 47, 21194-21200.



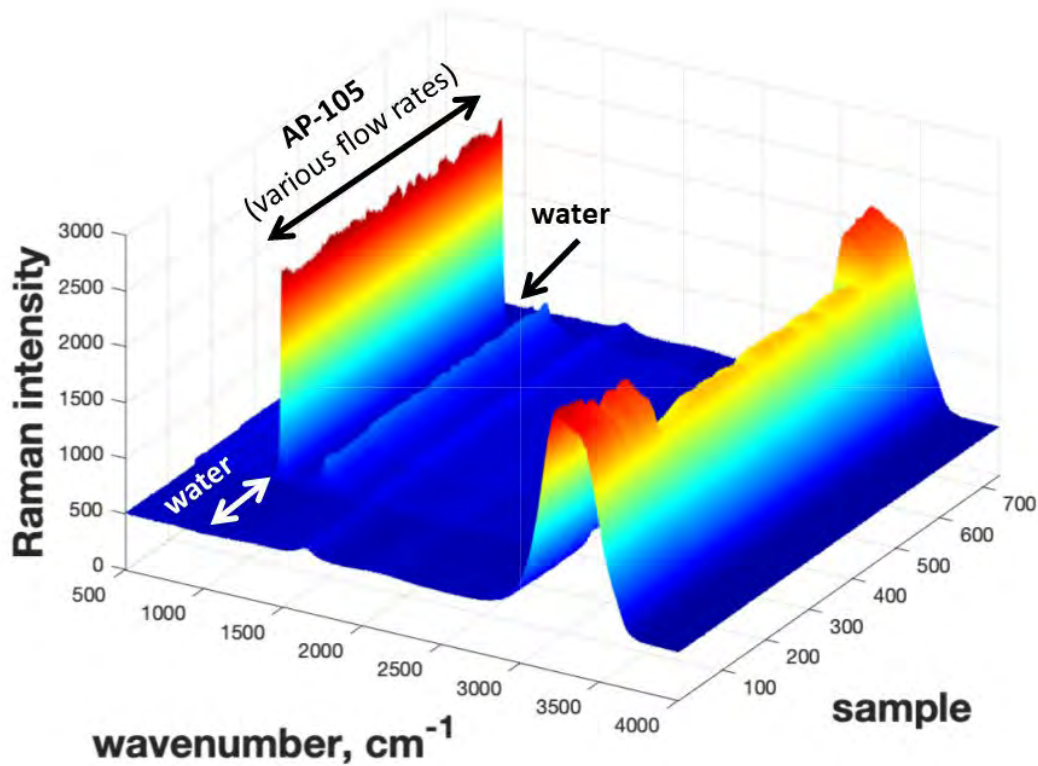
# Hanford waste applications: Raman spectroscopy

- Significant sampling burden on clean up sites, can on-line process monitoring help reduce this?
- Raman spectroscopy is an ideal tool for characterizing many of the polyatomic species within tank wastes
- This includes targets that represent process upset concerns, e.g., phosphate as well as a variety of species that are valuable to know when optimizing glass formulation

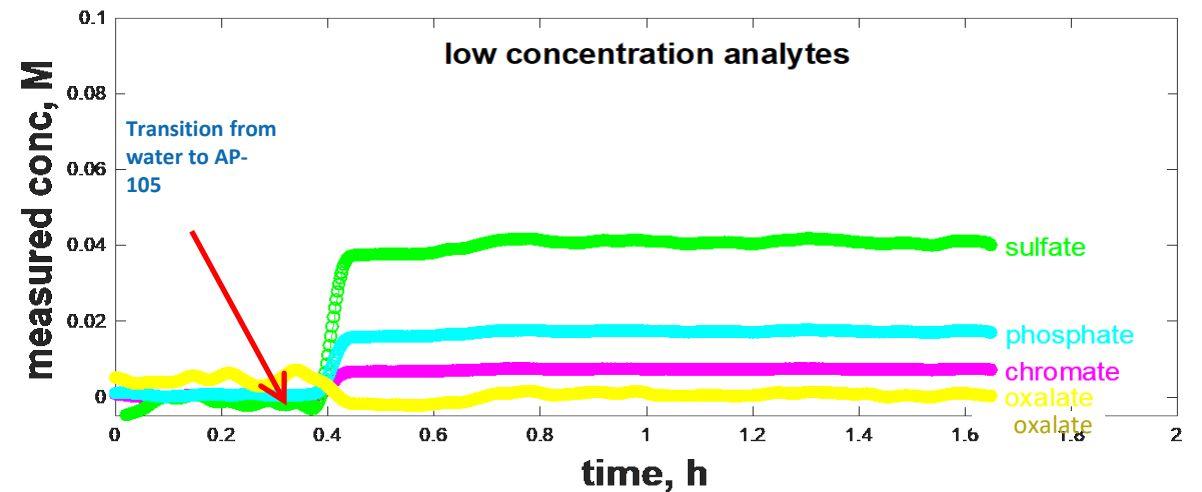
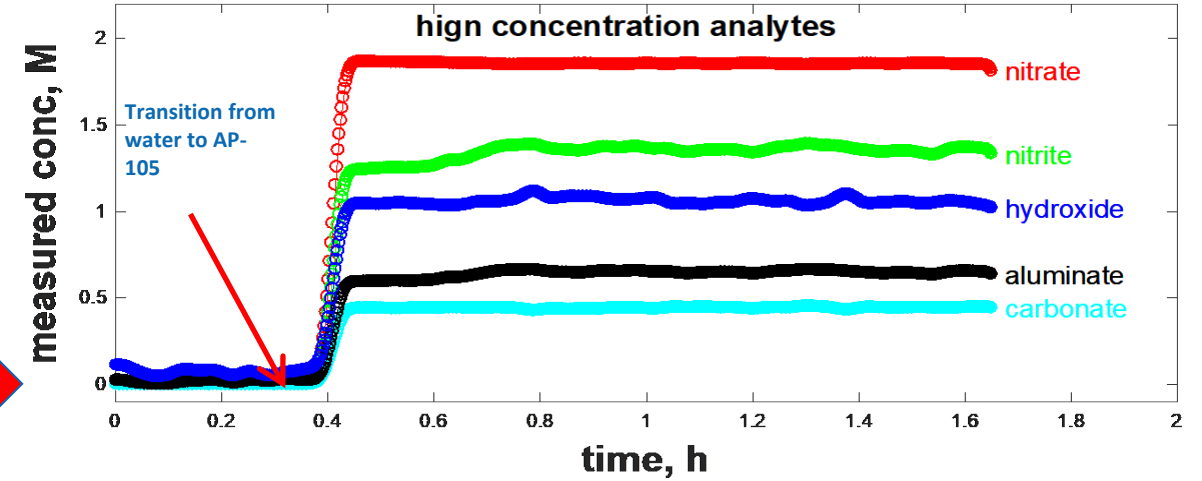




# Hanford waste applications: actual waste demo



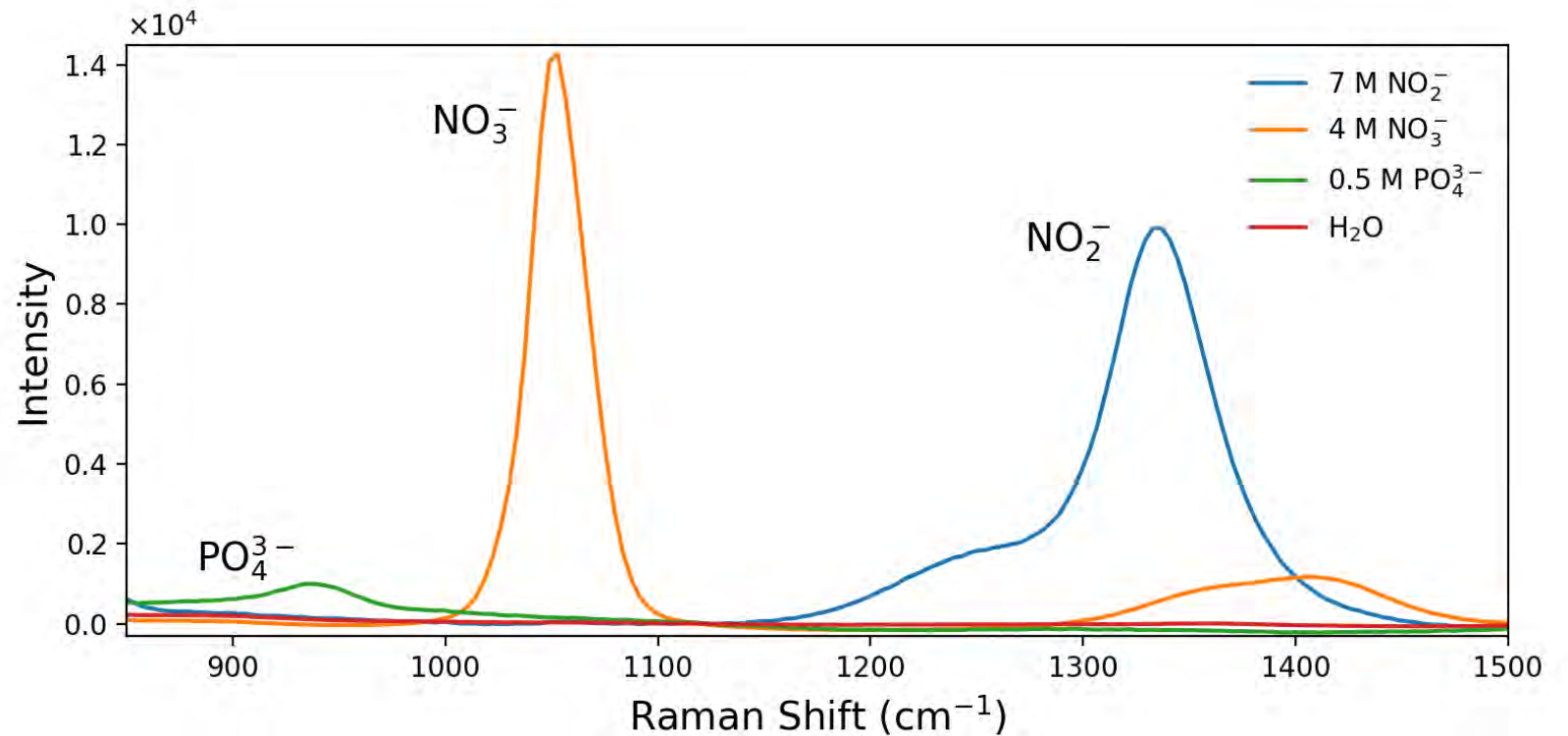
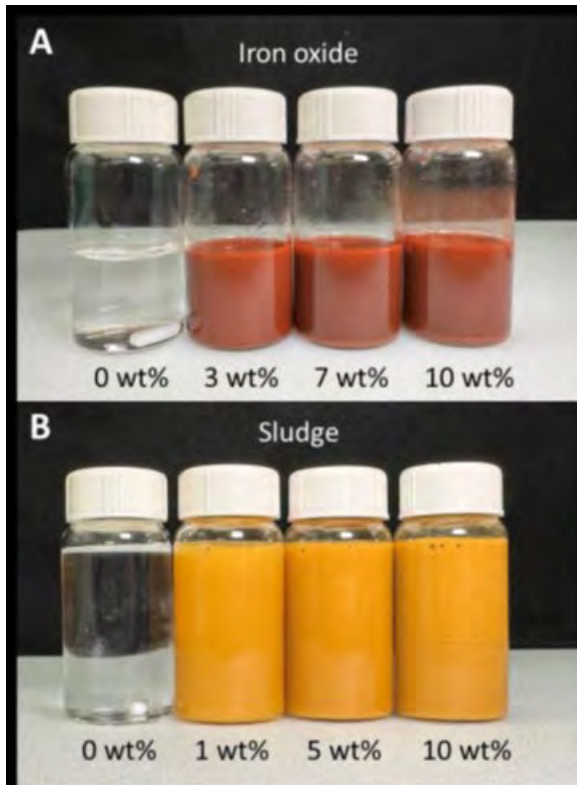
model



*Ind. Eng. Chem. Res.*, 2019, 58, 47, 21194-21200

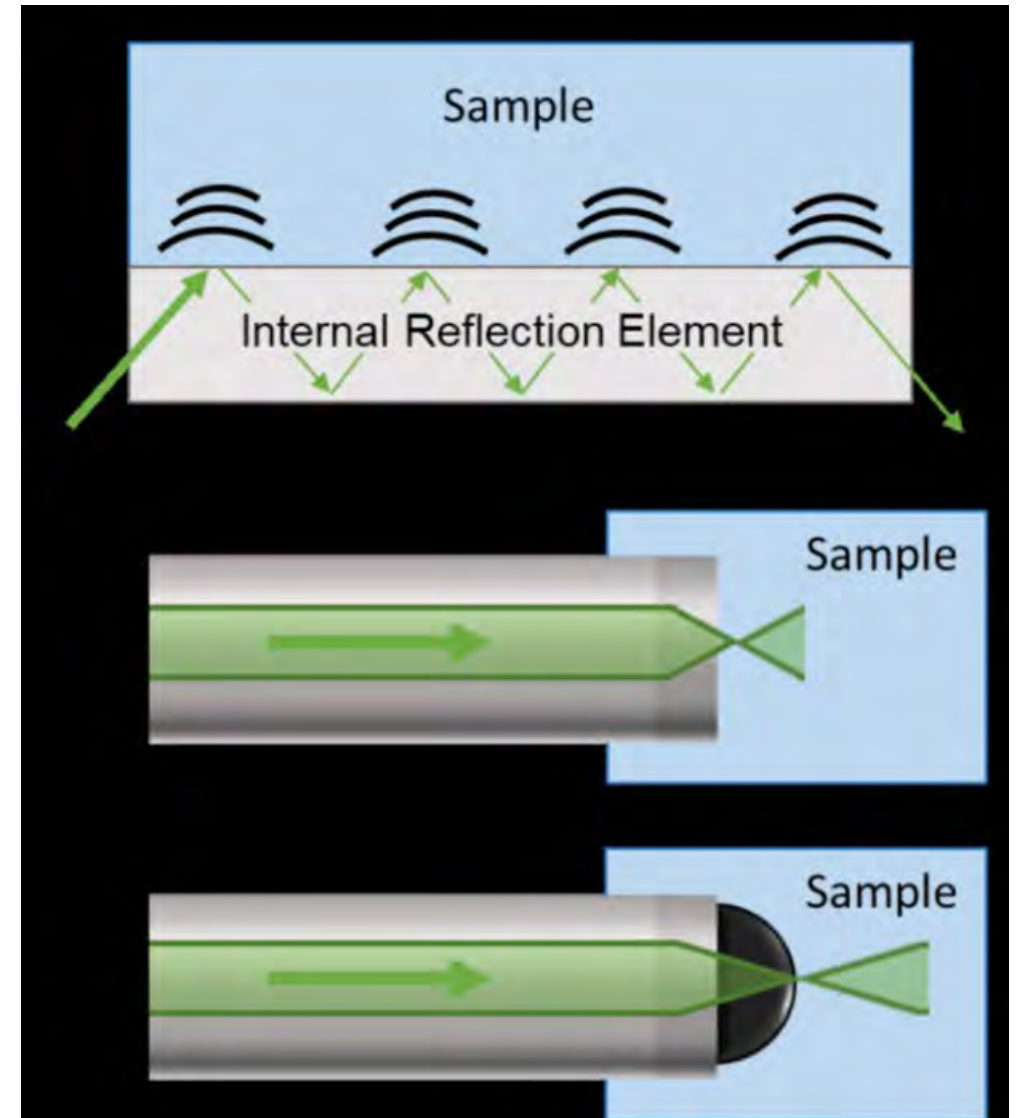
# Hanford waste applications: high turbidity / solids

- Explore applications to three chemical targets of interest for process control purposes
- Vary turbidity of system using both iron oxide solids and a tank sludge simulant



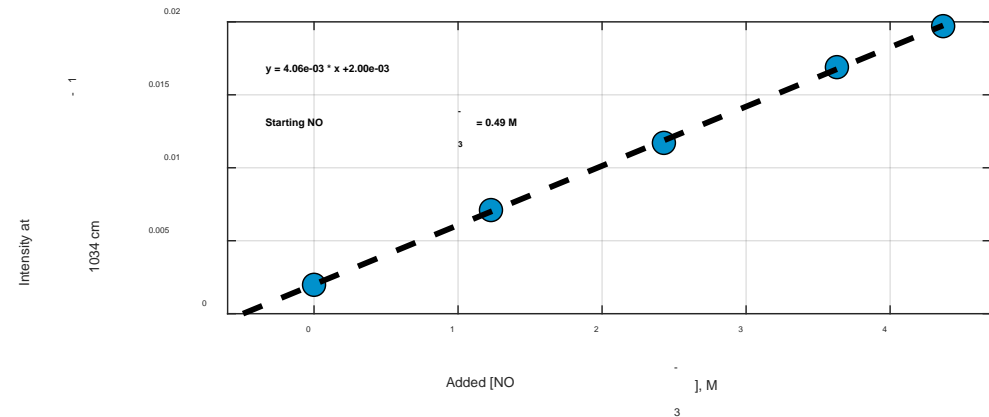
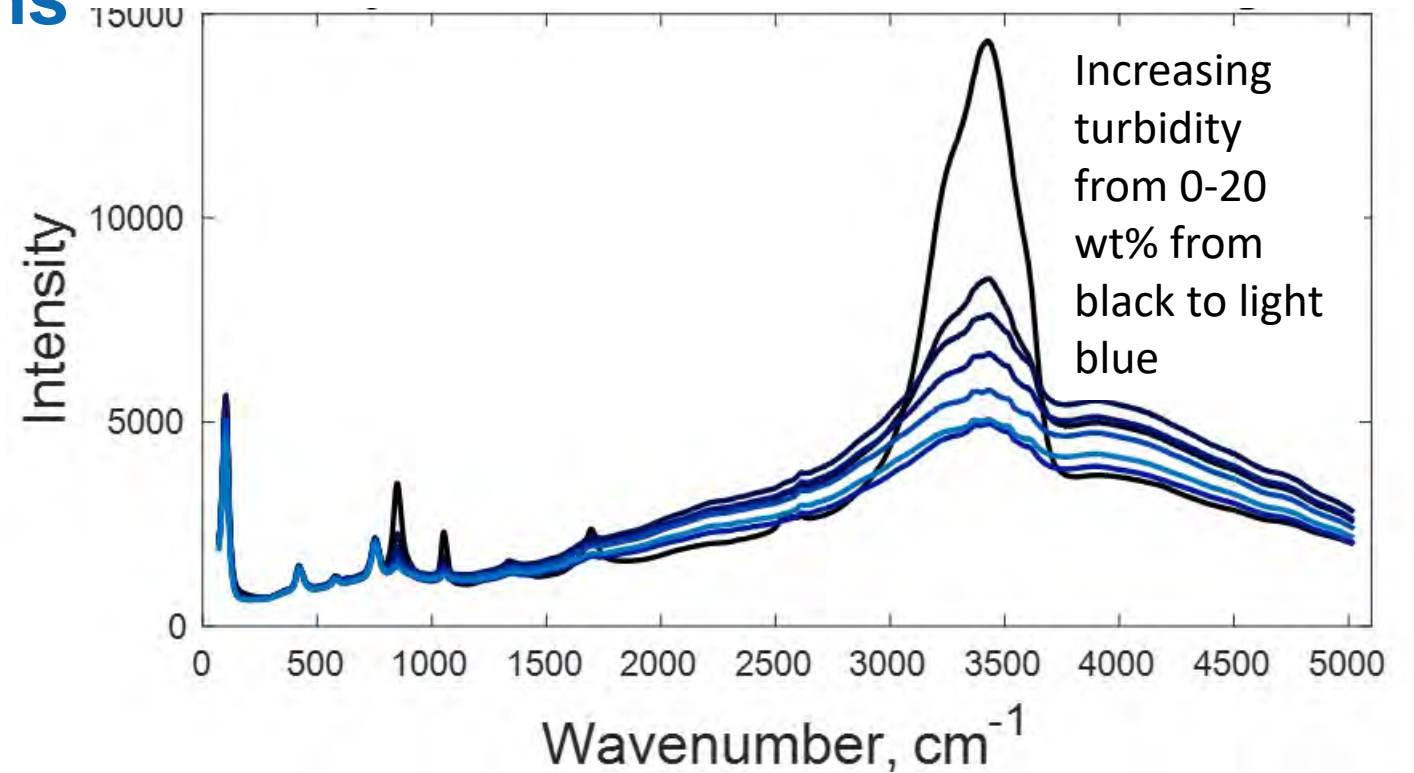
# Hanford waste applications

- Re-think probe design to better interrogate systems of high and variable turbidity
- Exploring very close focus probes that are available commercial off the shelf
- Ultimately tie data from optimized probes into chemical data science analysis to build highly robust tools for quantifying tank waste components



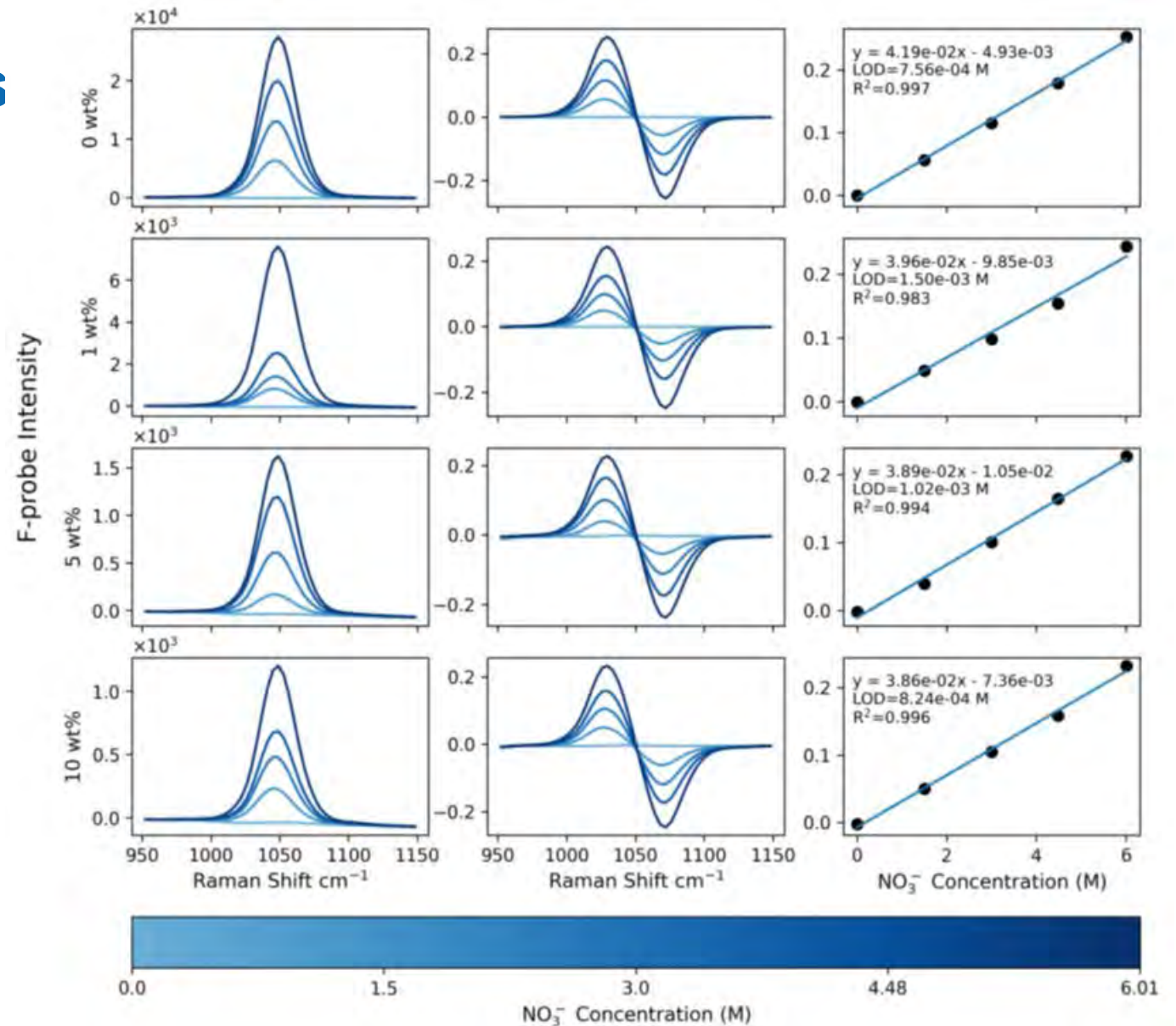
# Hanford waste applications

- Explore application to high level waste simulant provided by SRNL
- Expanded range of study from 0-20wt% solids loading
- With data pre-processing, qualitative analysis of results suggests highly accurate models could be built

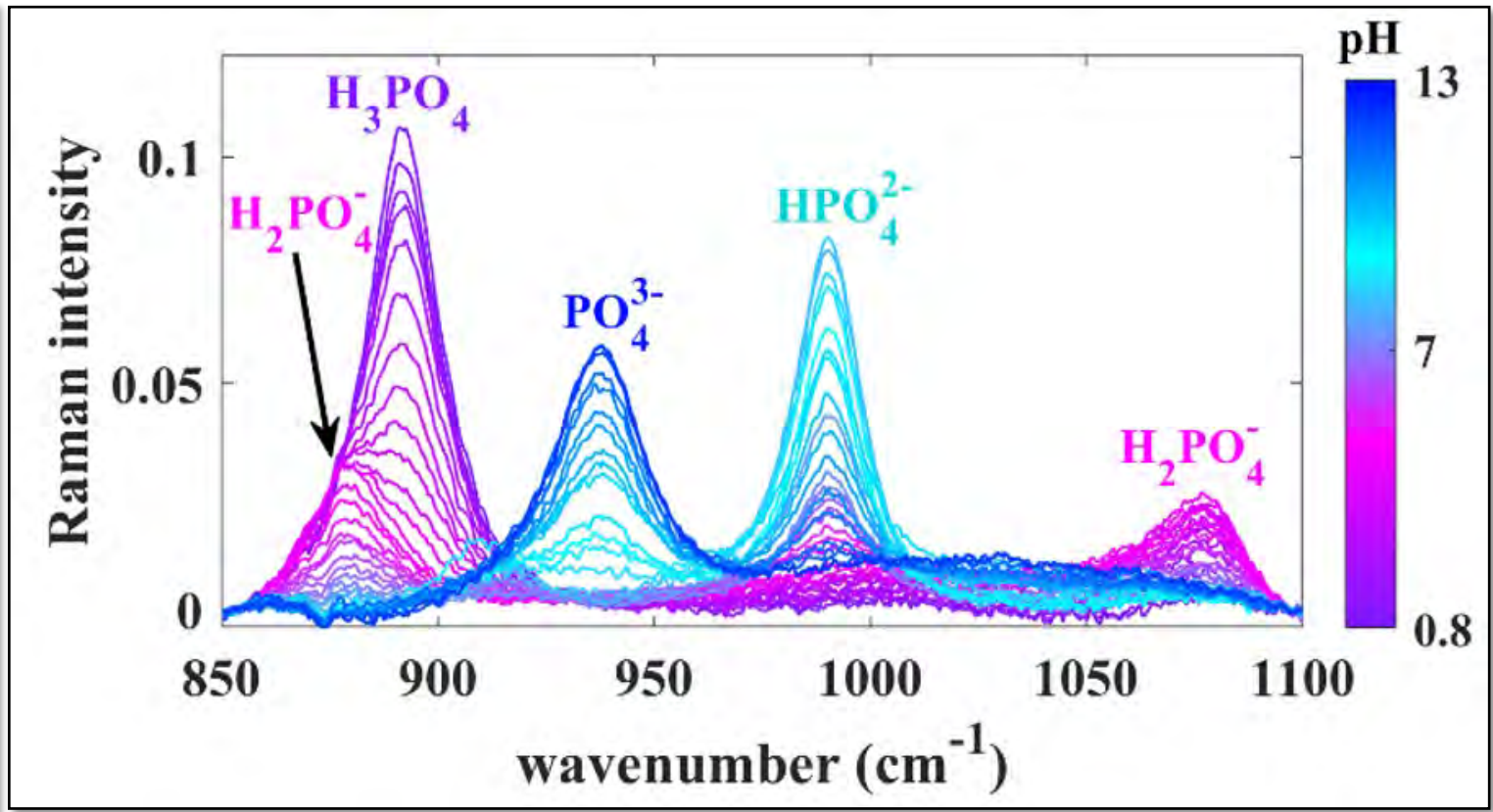
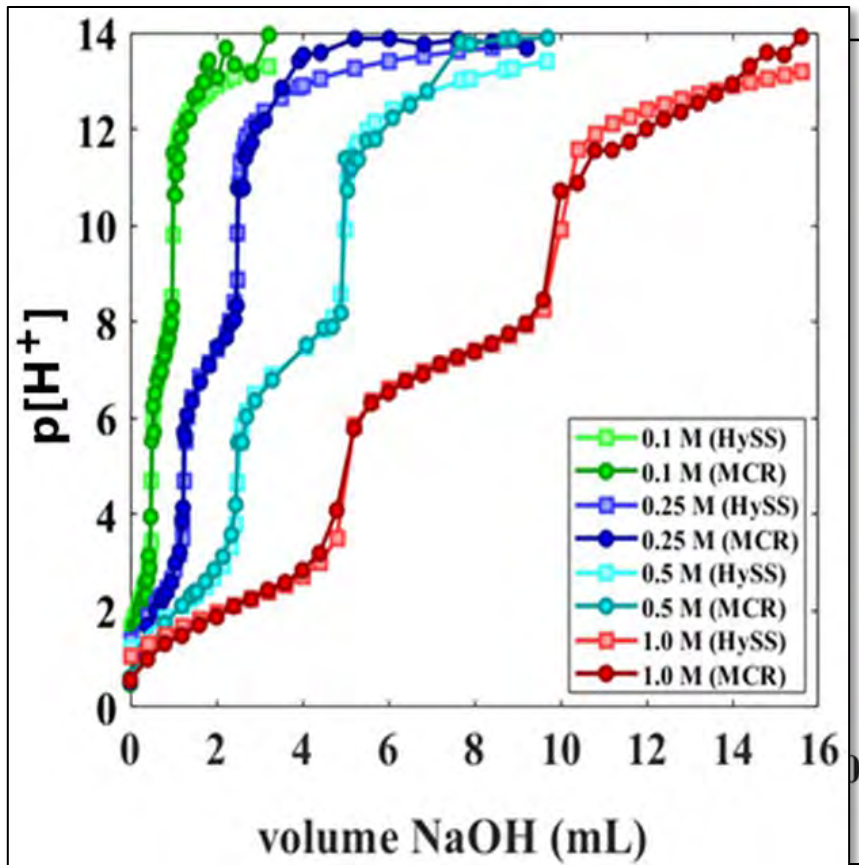
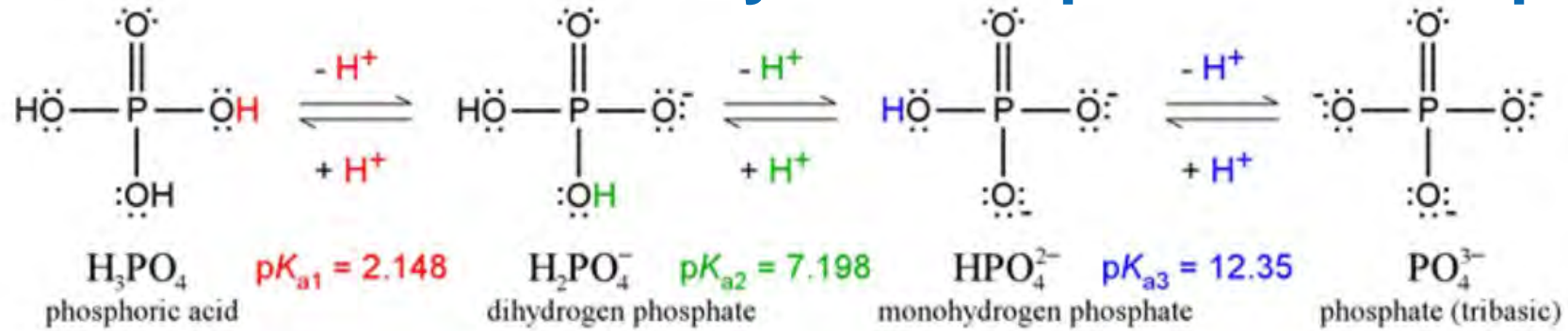


# Hanford waste applications

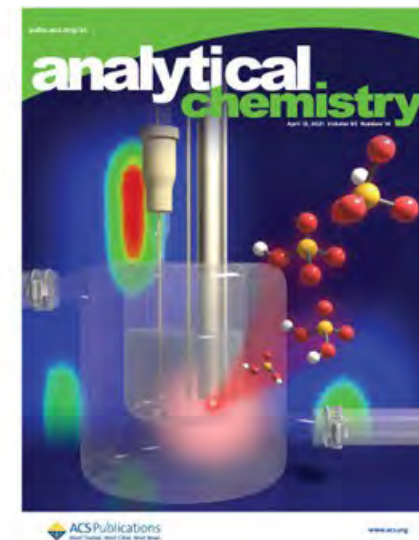
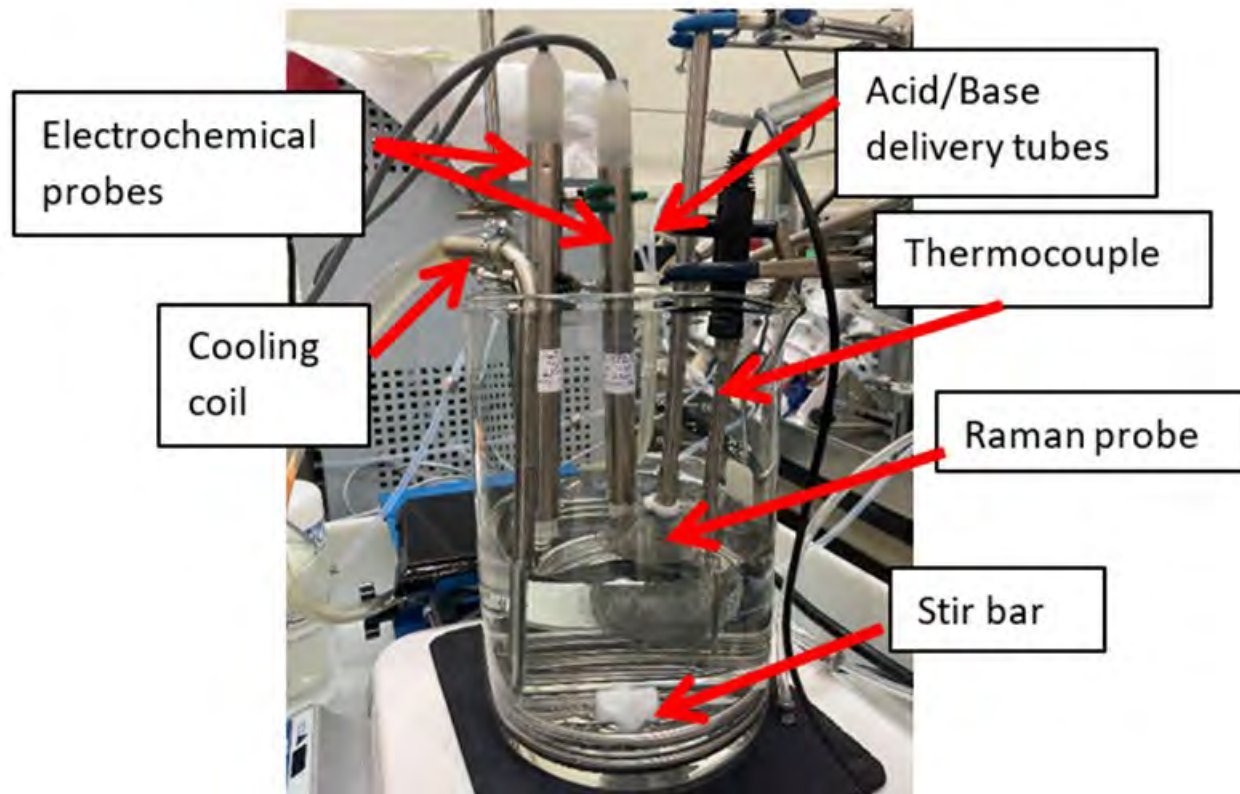
- Utilizing data pre-processing, many impacts of turbidity can be mitigated
  - If using the correct probe/detector combo
- Here, even single variate results begin to look better after simple steps like normalization



# pH measurement: Raman Activity of Phosphoric Acid Species



# Automated Process Control: pH Dependent Processing Schemes



Casella, A. J.; Levitskaia, T. G.; Peterson, J. M.; Bryan, S. A., Water O-H Stretching Raman Signature for Strong Acid Monitoring via Multivariate Analysis. *Anal Chem* 2013, 85 (8), 4120-4128.

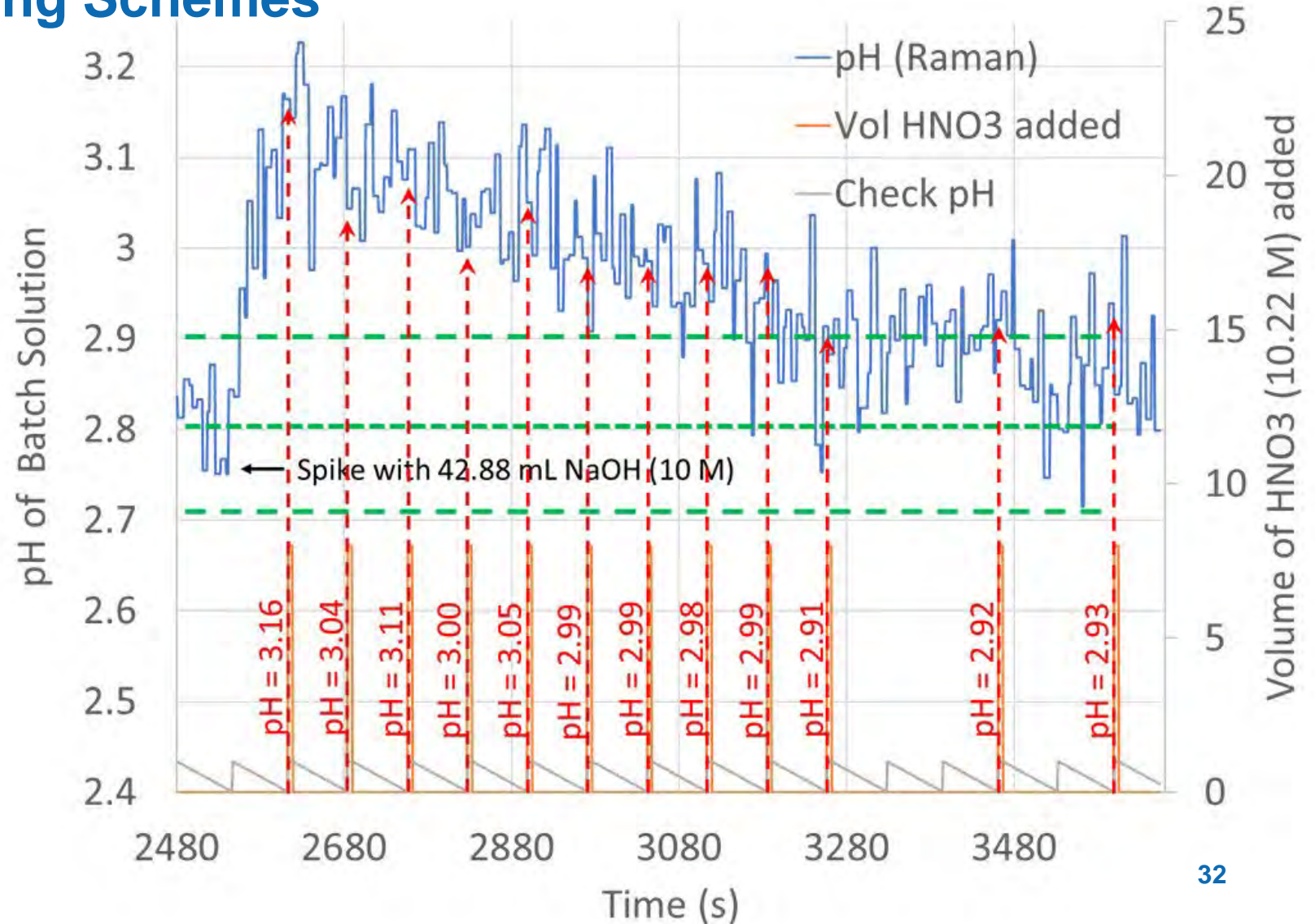
Lackey, H. E.; Nelson, G. L.; Lines, A. M.; Bryan, S. A., Reimagining pH Measurement: Utilizing Raman Spectroscopy for Enhanced Accuracy in Phosphoric Acid Systems. *Anal Chem* 2020, 92 (8), 5882-5889.

Clifford, A. J.; Lackey, H. E.; Nelson, G. L.; Bryan, S. A.; Lines, A. M., Raman Spectroscopy Coupled with Chemometric Analysis for Speciation and Quantitative Analysis of Aqueous Phosphoric Acid Systems. *Anal Chem* 2021, 93 (14), 5890-5896.

Heller, F. D.; Ahlers, L. R. H.; Nordquist, Z. E.; Gunawardena, N. H.; French, A. D.; Lines, A. M.; Nelson, G. L.; Casella, A. J.; Bryan, S. A., Development of Online pH Monitoring for Lactic, Malonic, Citric, and Oxalic Acids Based on Raman Spectroscopy Using Hierarchical Chemometric Modeling. *Anal Chem* 2022, 94 (50), 17467-17476.

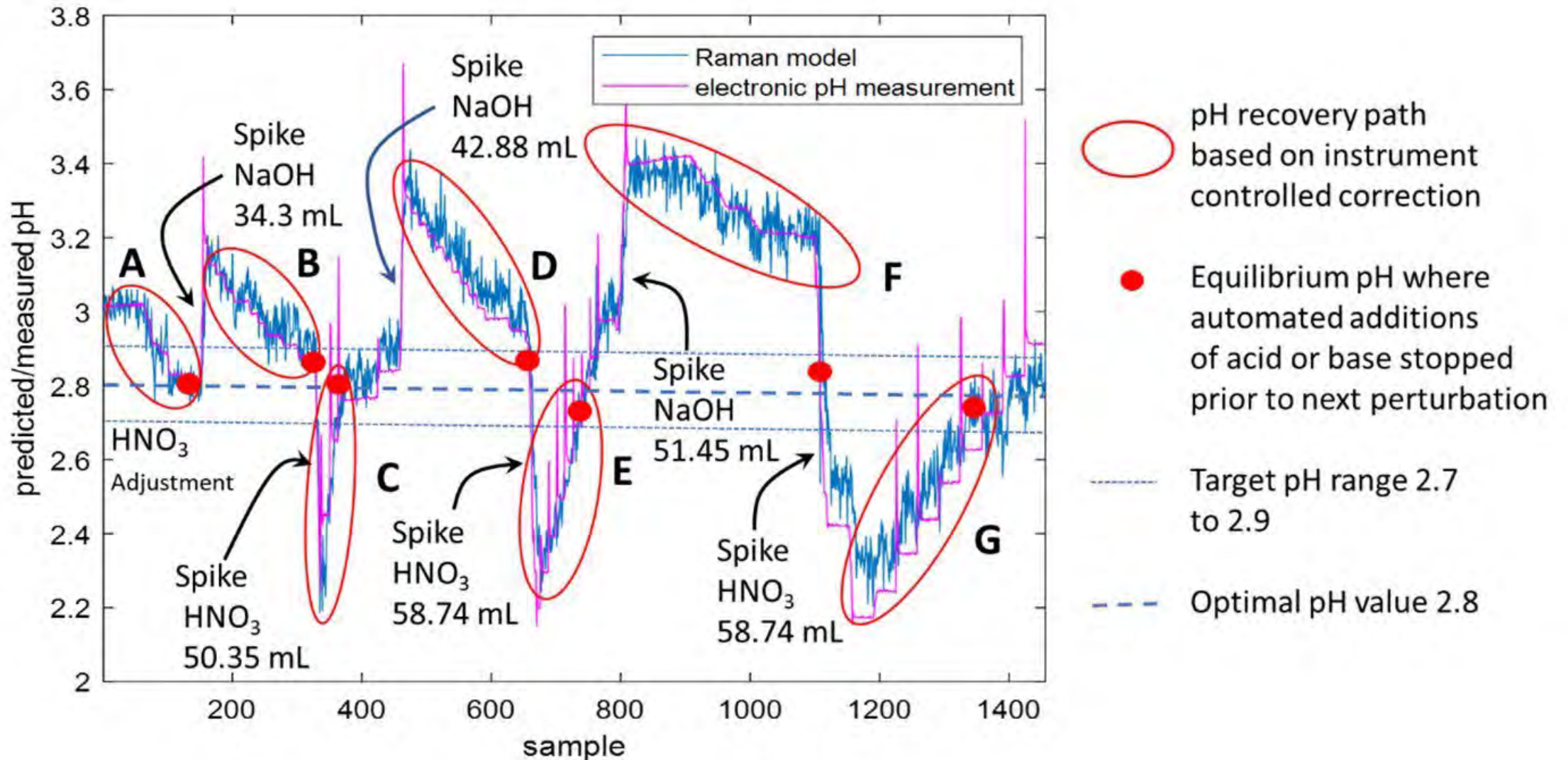
# Automated Process Control: pH Dependent Processing Schemes

- PLC injects volume preset by operator (here constant 8 mL) of acid or base to compensate for  $\Delta$ pH incrementally
- PLC calculates injection time required based on pump strength to set the constant volume additions



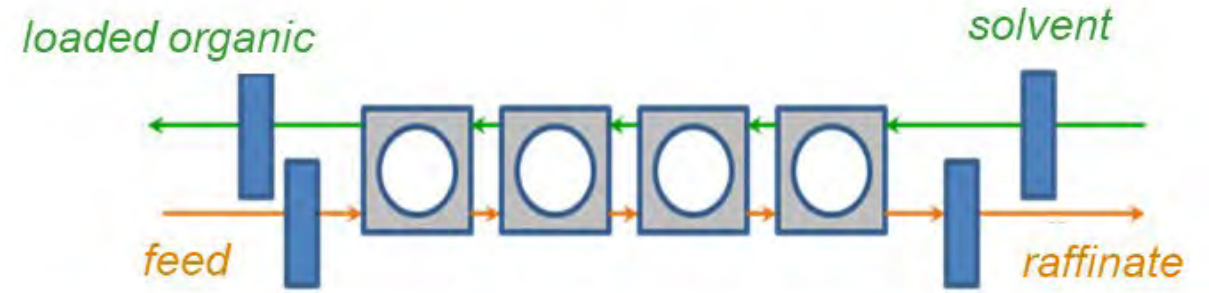


# Automated Process Control: pH Dependent Processing Schemes

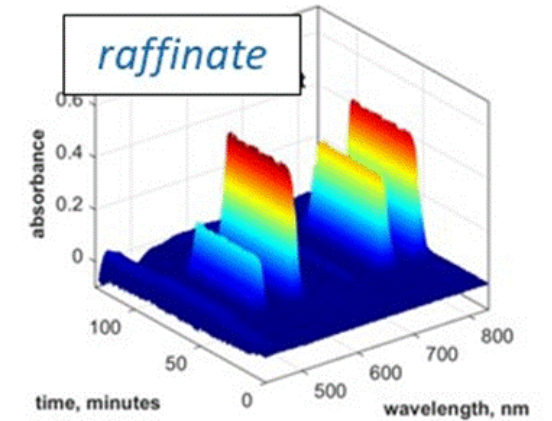
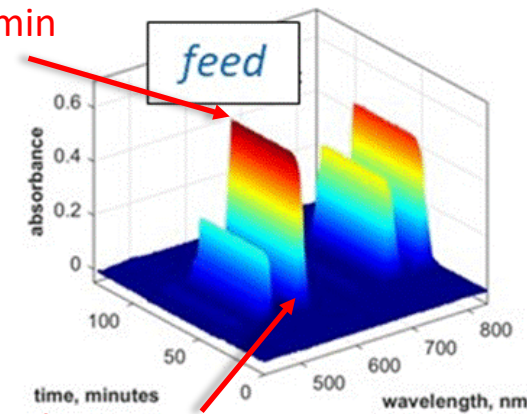


# Automated Process Control: Mass Balance During Extraction

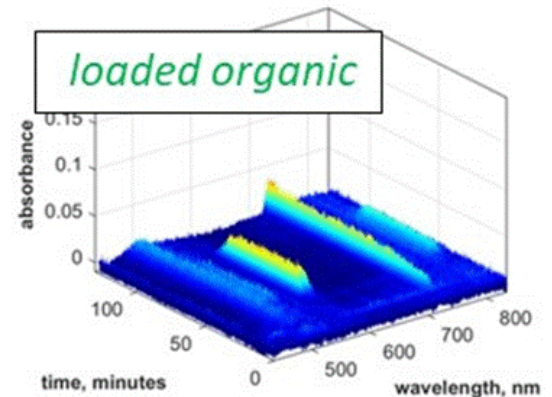
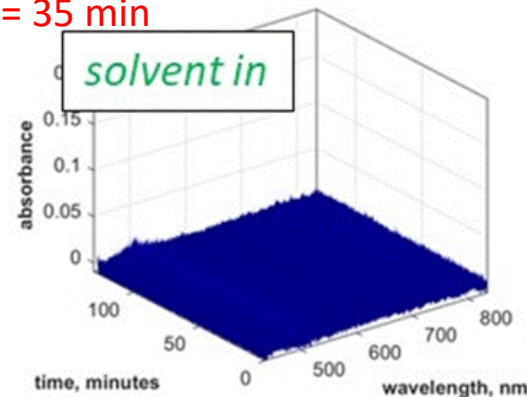
- Transition to characterizing counter current separation process on a bank of centrifugal contactors
- Outfit all inlets and outlets with optical sensors
- In addition to Raman for pH, utilize UV-vis for metal quantification
- gaining comprehensive insight into process chemistry



Nd feed stopped  
t = 90 min

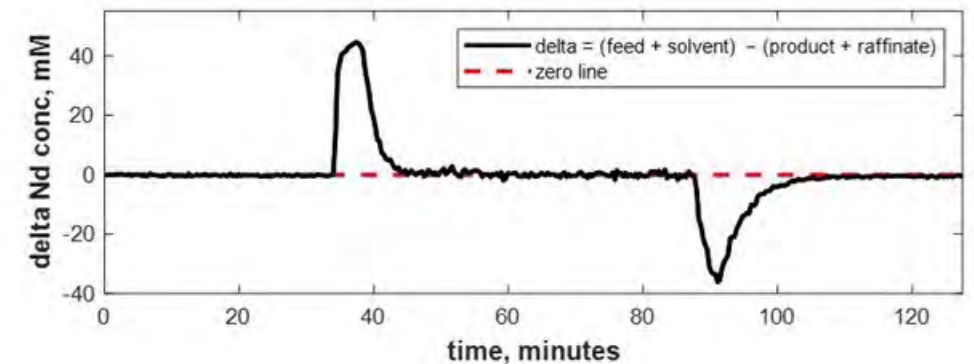
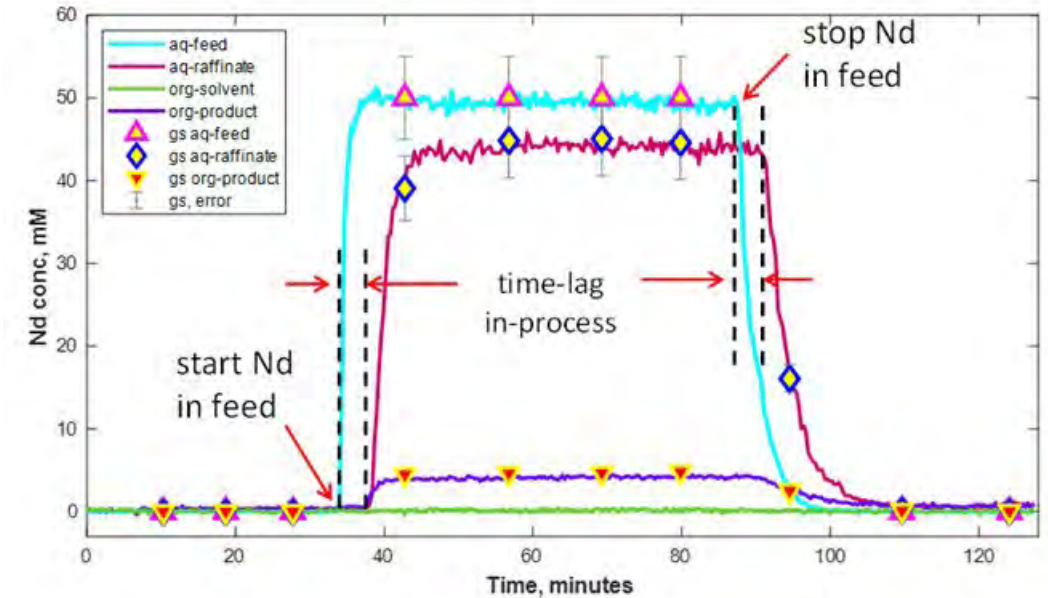


Nd feed started  
t = 35 min



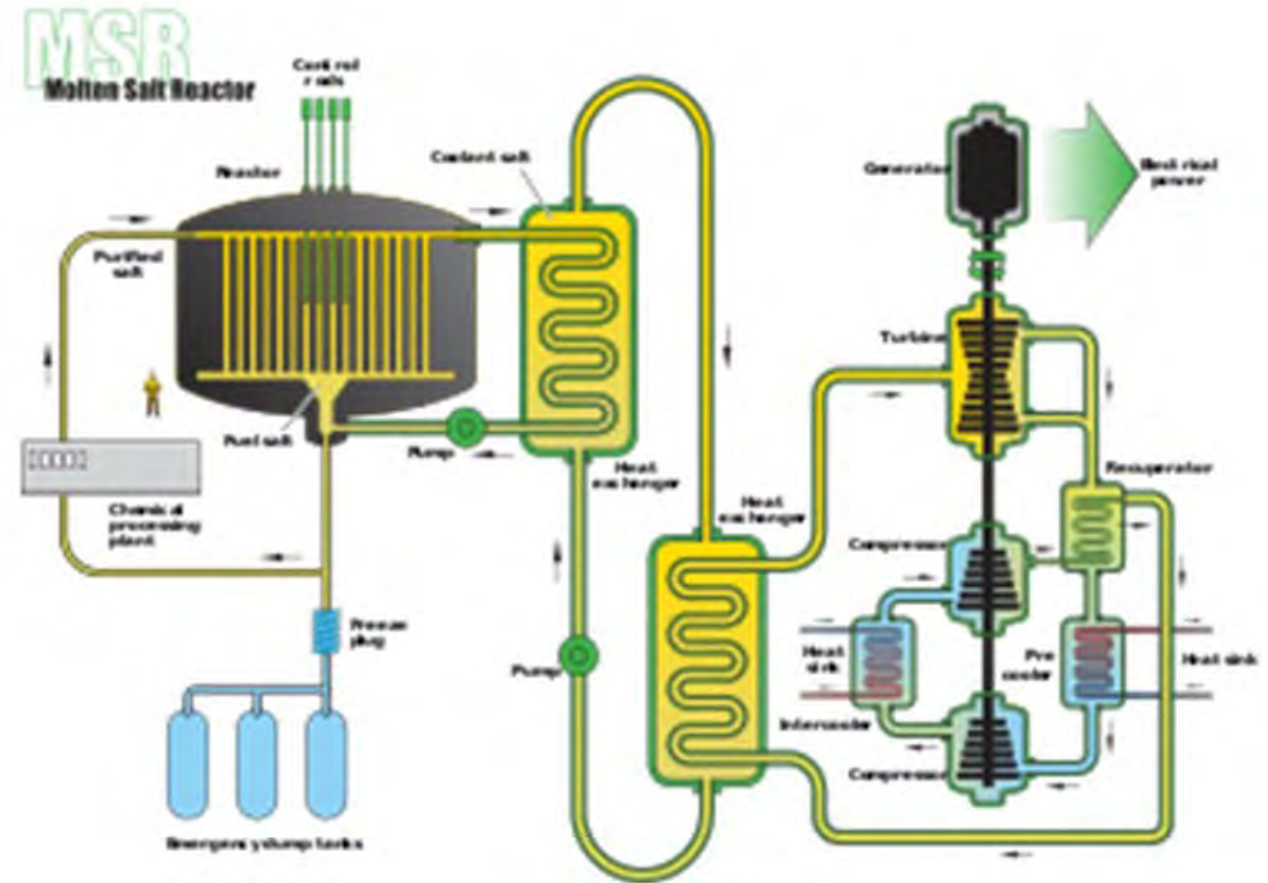
# Automated Process Control: pH Dependent Processing Schemes

- Run flow tests to first demonstrate chemometric models for metal concentration are functional
- Models utilized LWR PLS algorithms to overcome nonlinear response in variable pH environment
- Note, this data can be paired with flow/density measurements to allow for mass balance



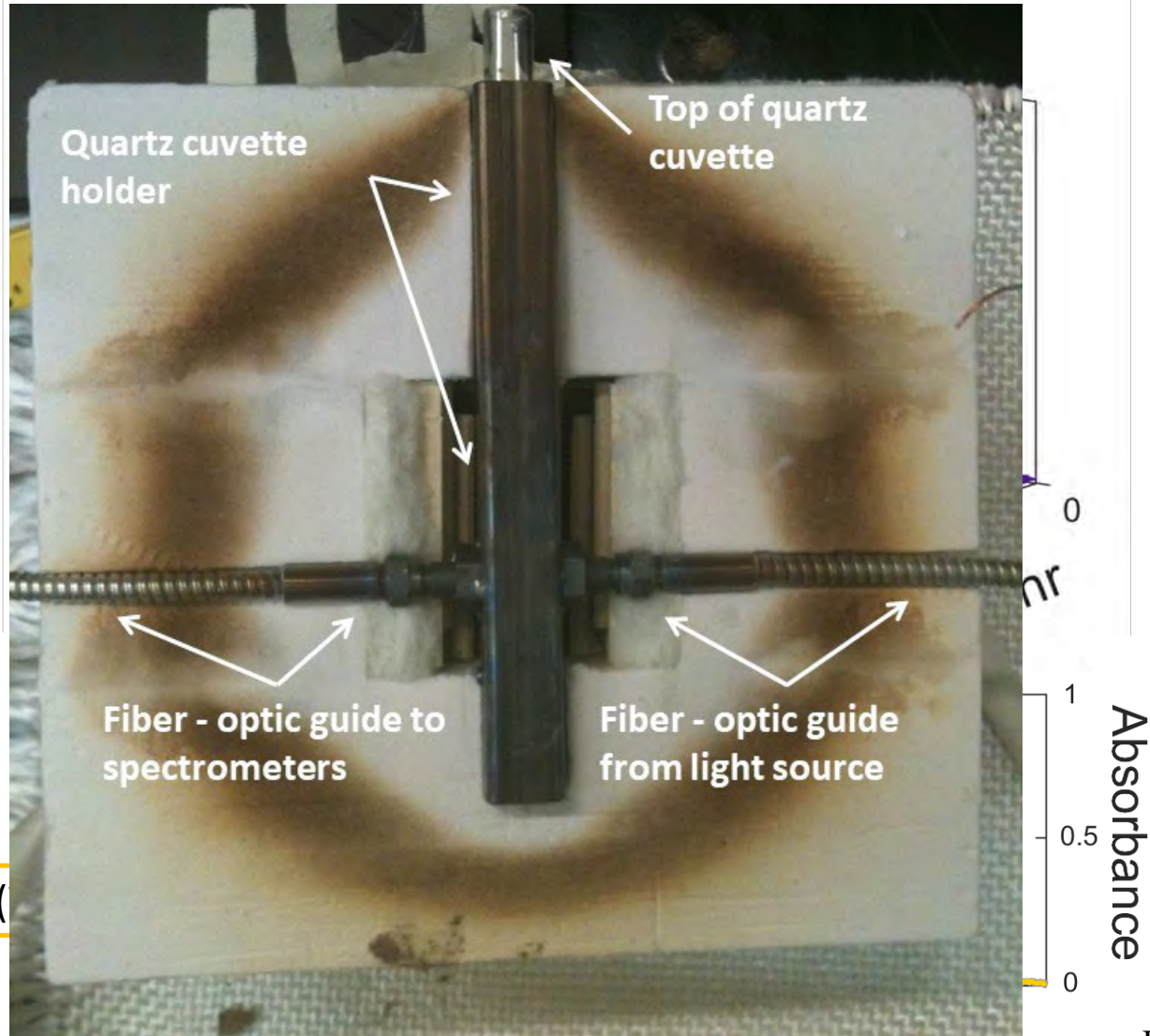
# Application to MSR Salts

- Build and demonstrate applications throughout the treatment process
- Sensors operating in extremely harsh environments
  - Temperatures  $> 500^{\circ}\text{C}$
  - Highly corrosive solutions/gases
  - Radiation environments
- Applications for material accountancy, process control, and fundamental characterization



# Building library for chemometric modeling

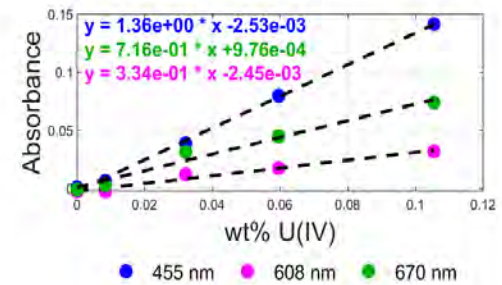
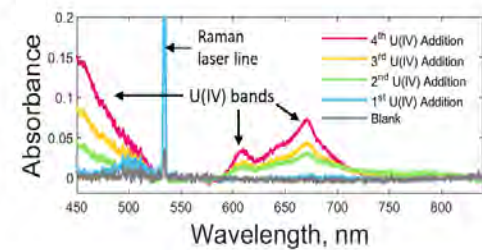
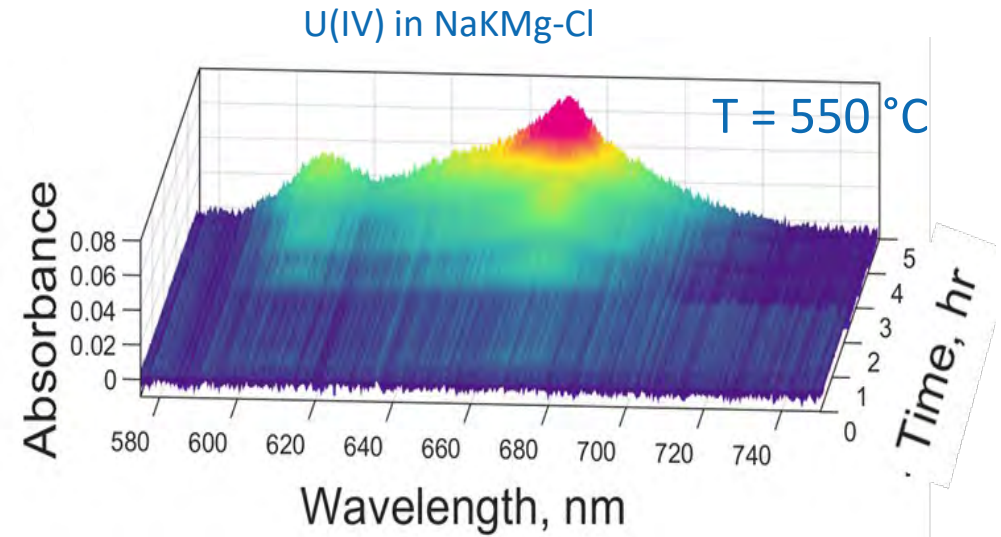
U(III) in NaMg-Cl



U(

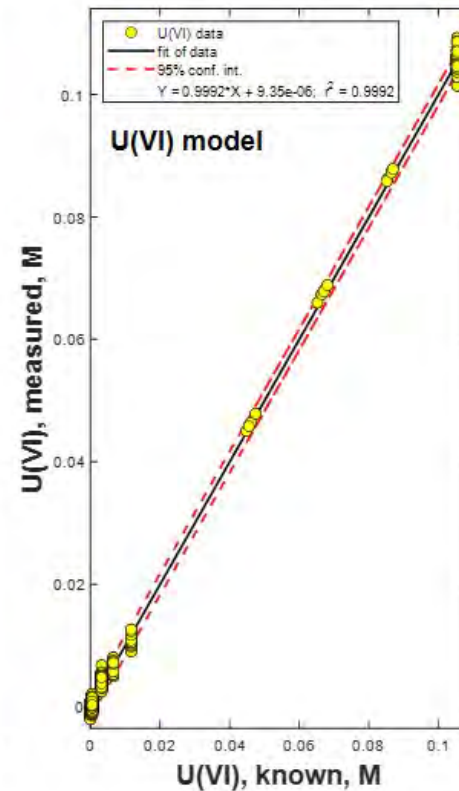
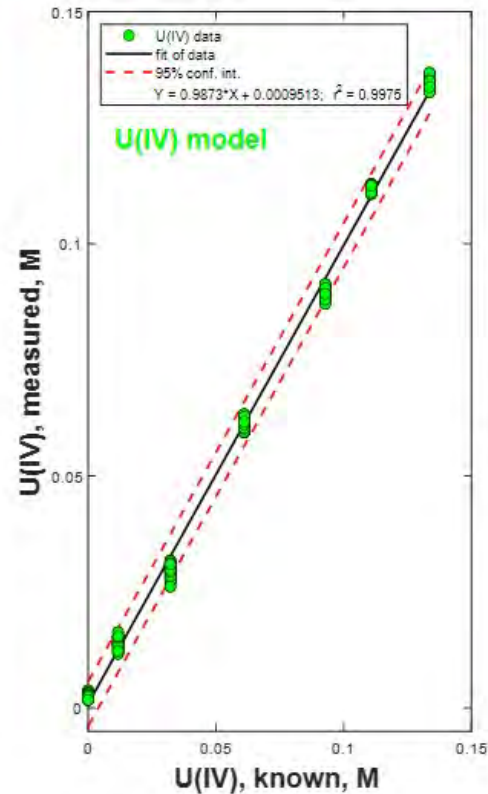
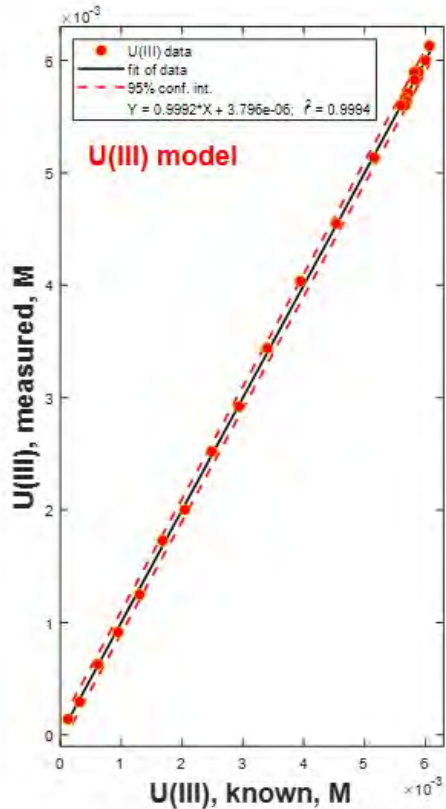
Wavelength, nm

Absorbance



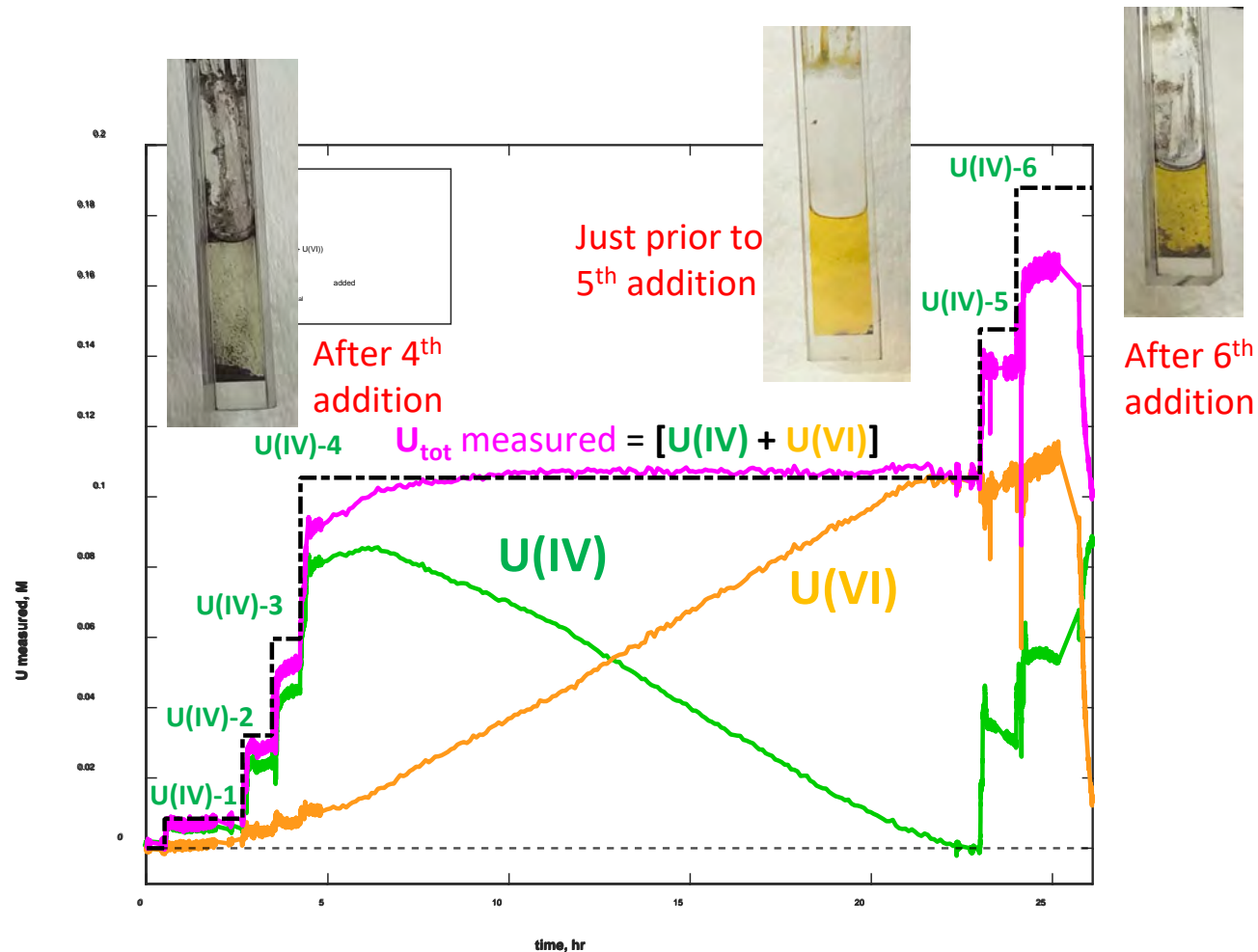
# Chemometric model building

Initial chemometric models showing accurate analysis of U in (III), (IV), and (VI) oxidation states within molten salt environment



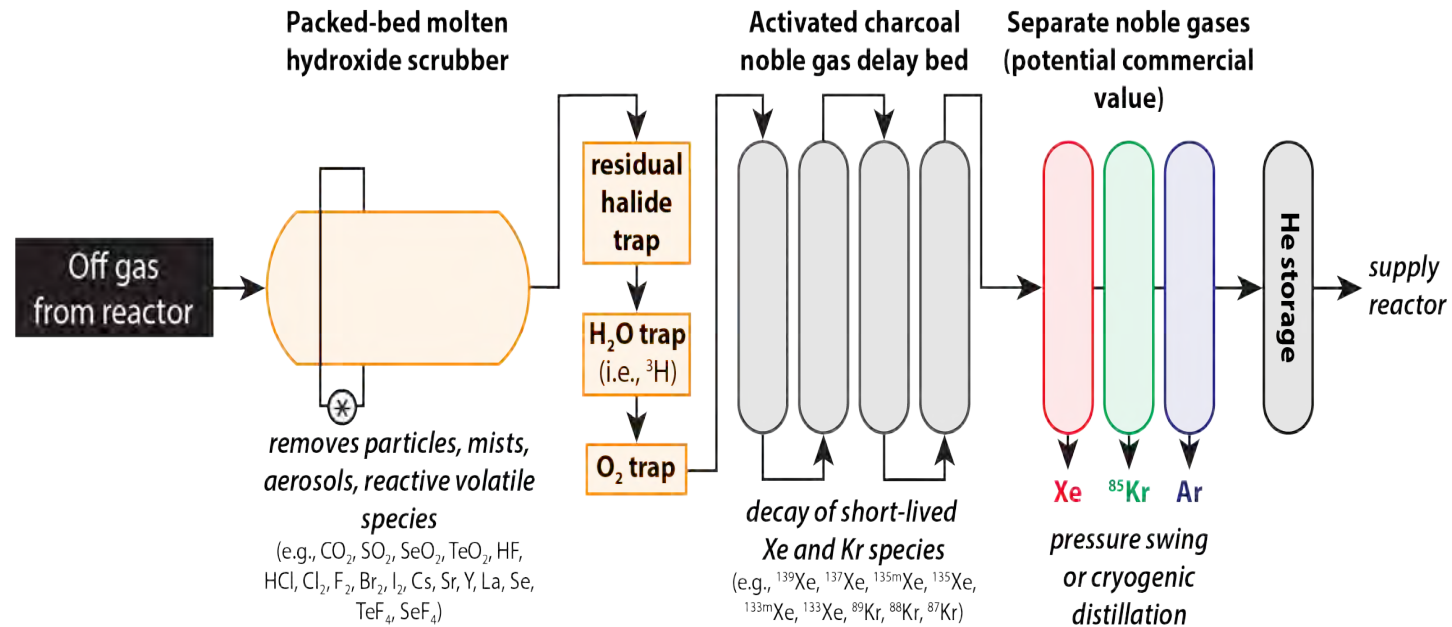
RMSECV (uncertainty)	
U(III)	0.00005 M
U(IV)	0.00416 M
U(VI)	0.00137 M

# U(IV)/U(VI) conversion in NaCl-KCl-MgCl<sub>2</sub> eutectic



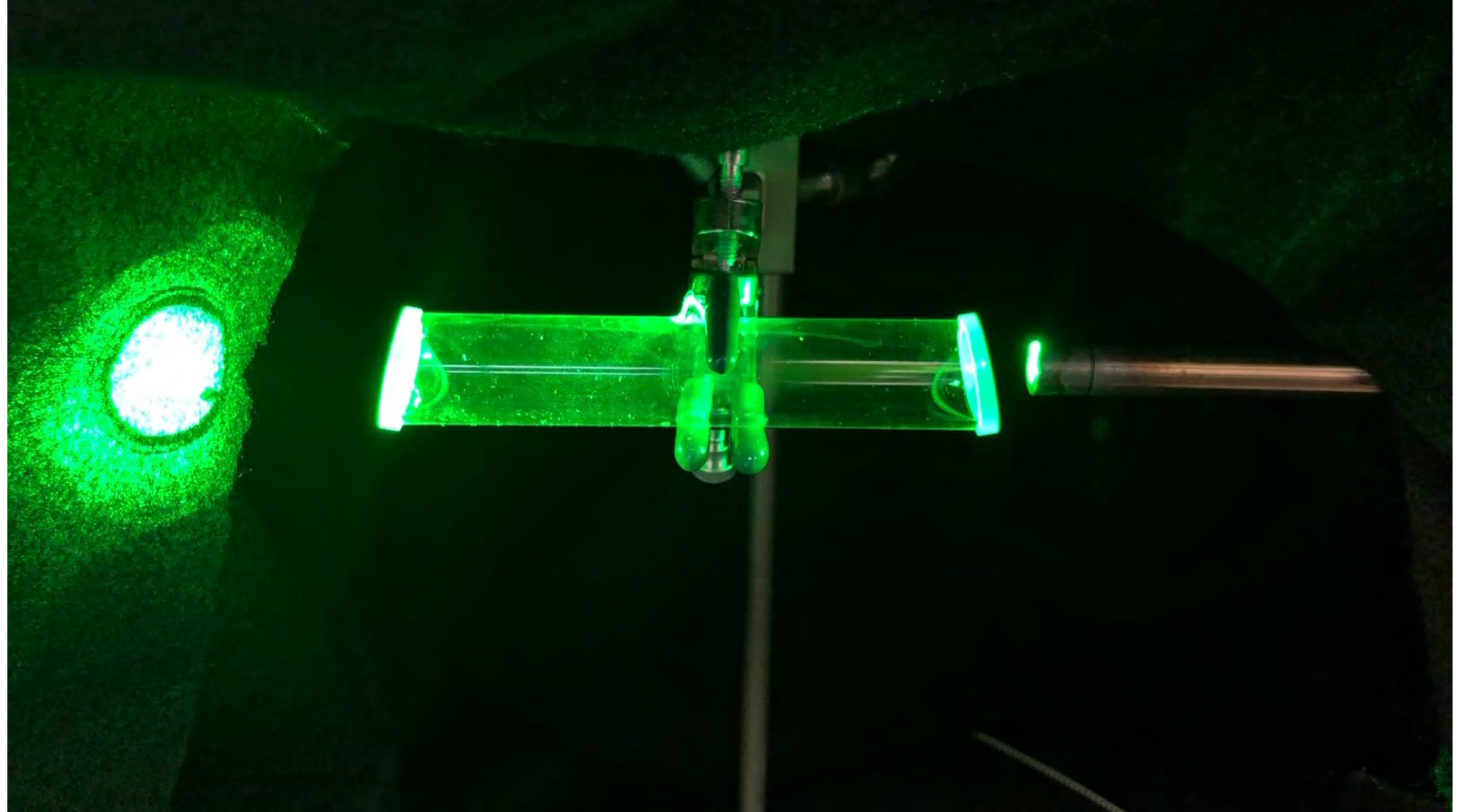
# Application to MSR off-gas streams

- PNNL developing OLM capabilities for chemical characterization of off-gas components
  - Demonstration on I<sub>2</sub>, ICl, and hydrogen isotopes
- Focus on Raman and FTIR
  - Ideal for deployment
- PNNL collaboration with ORNL





# Raman/fluorescence of I<sub>2</sub> gas



# Off-gas measurements: I<sub>2</sub>

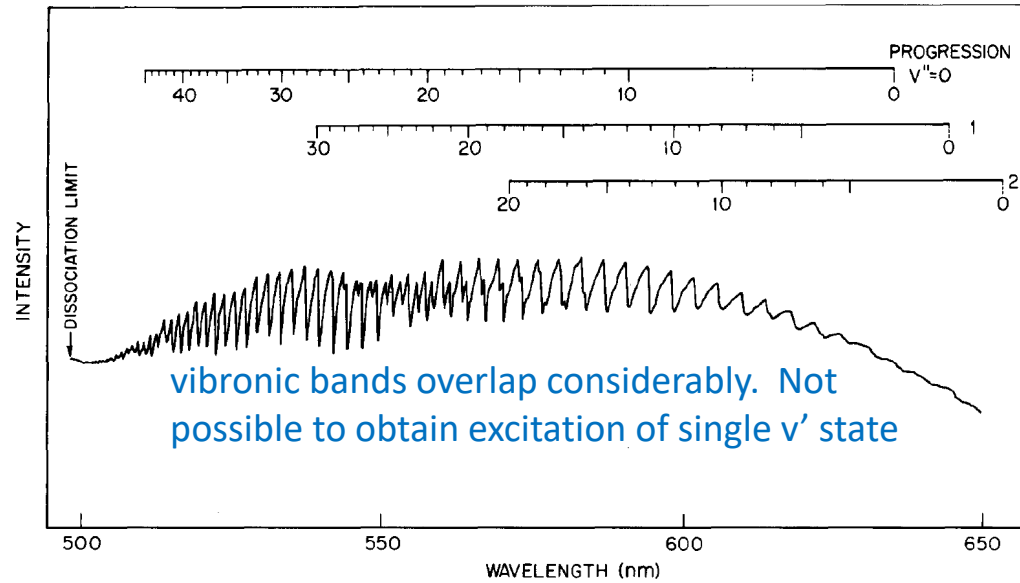
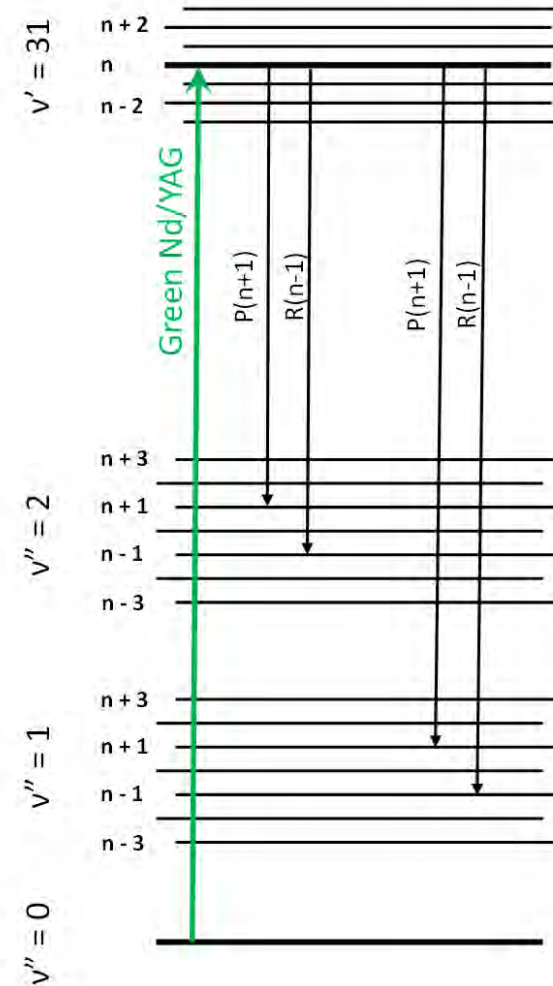


FIG. 2. Vibrational absorption spectrum of I<sub>2</sub> from 650 nm to the dissociation limit. Progressions to the lowest vibrational levels,  $v''=0, 1,$  and  $2,$  from various electronically excited vibrational levels  $v'$  are indicated.

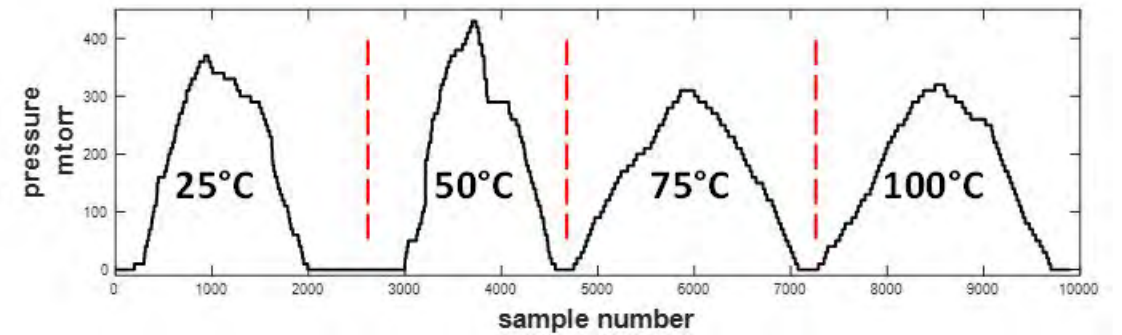
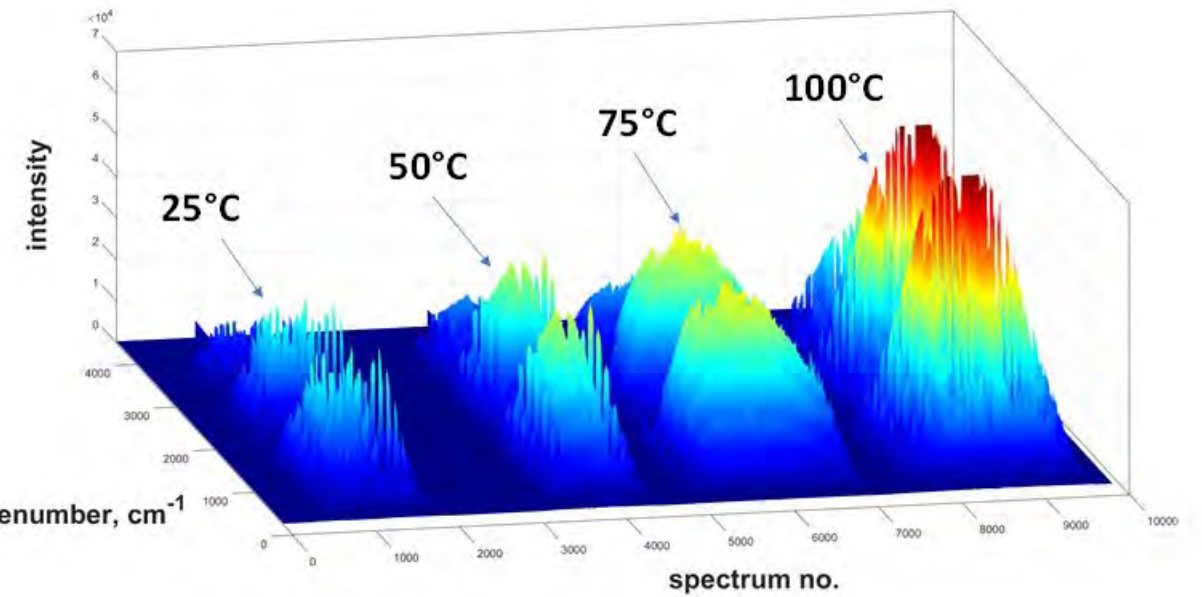
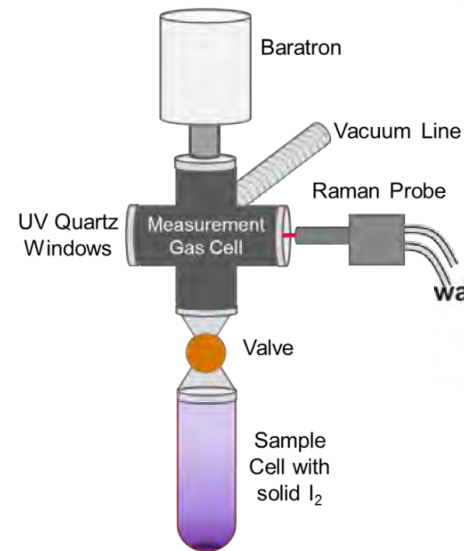
Capelle and Broida, *J. Chem. Phys.* 58, 4212–4222 (1973)

**B State**

**X State**



# Off-gas measurements: I<sub>2</sub>

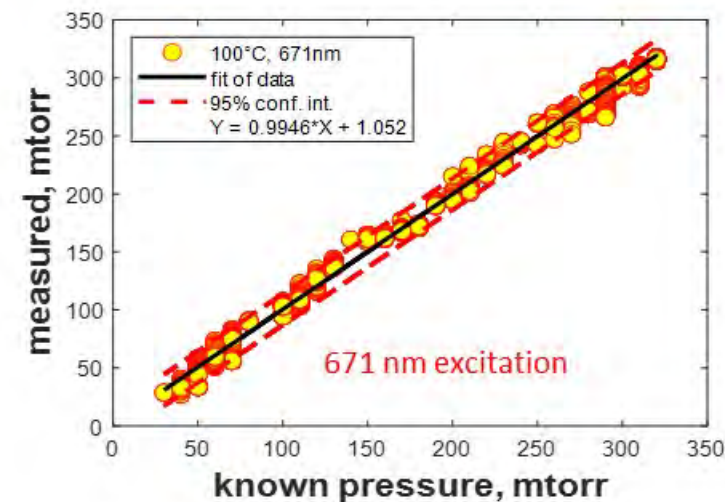
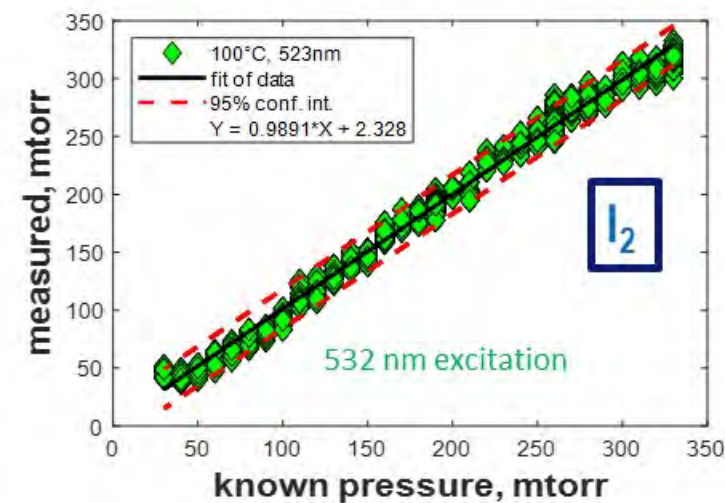
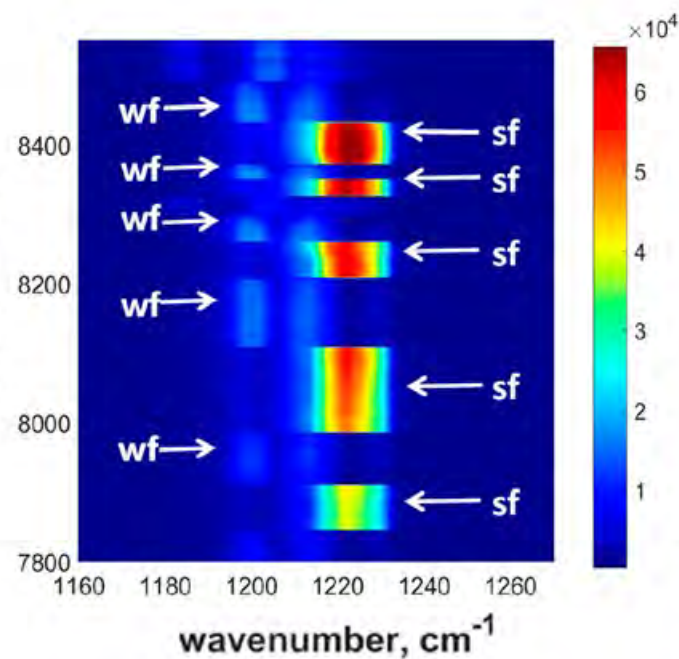


# Off-gas measurements: I<sub>2</sub>

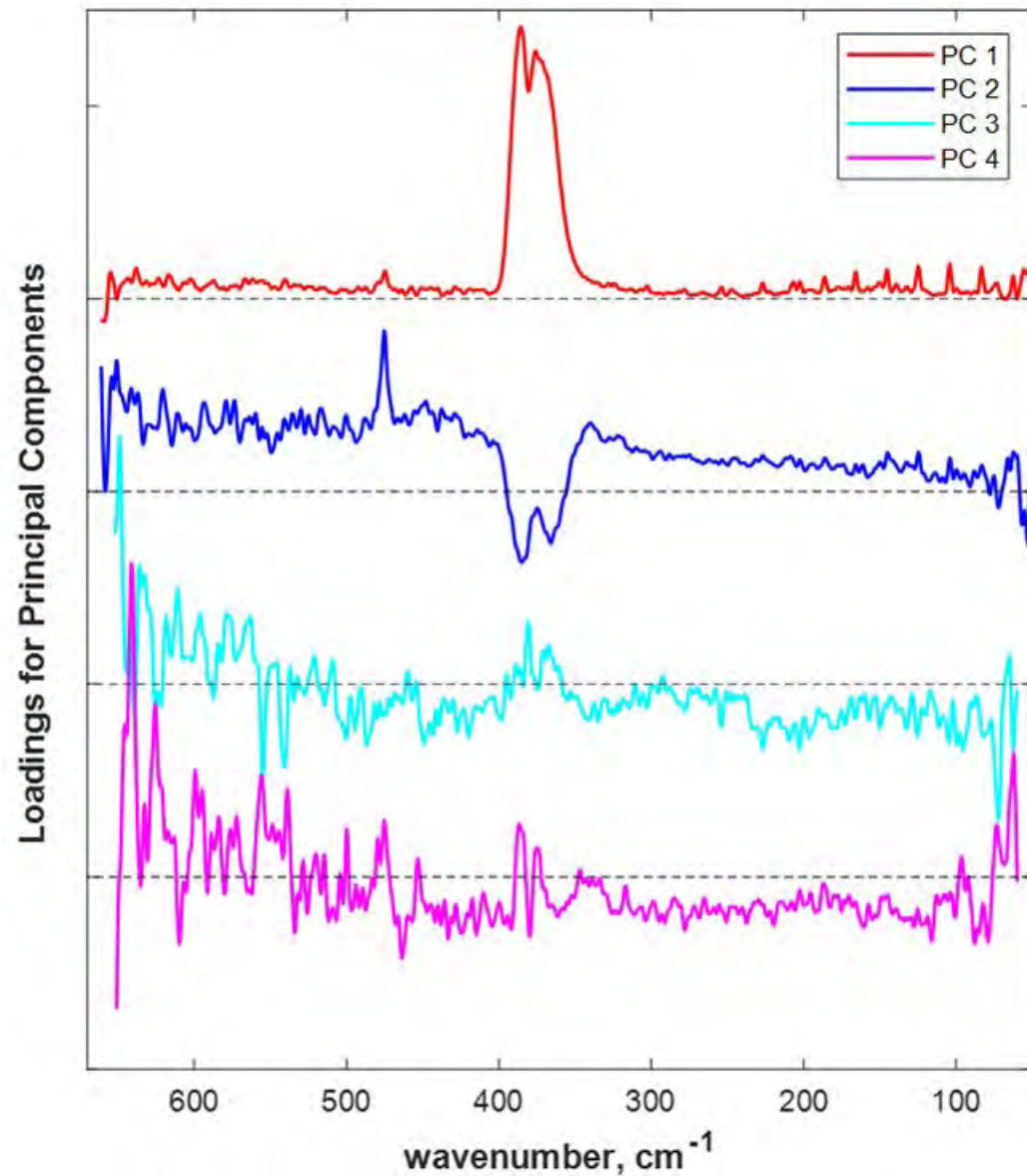
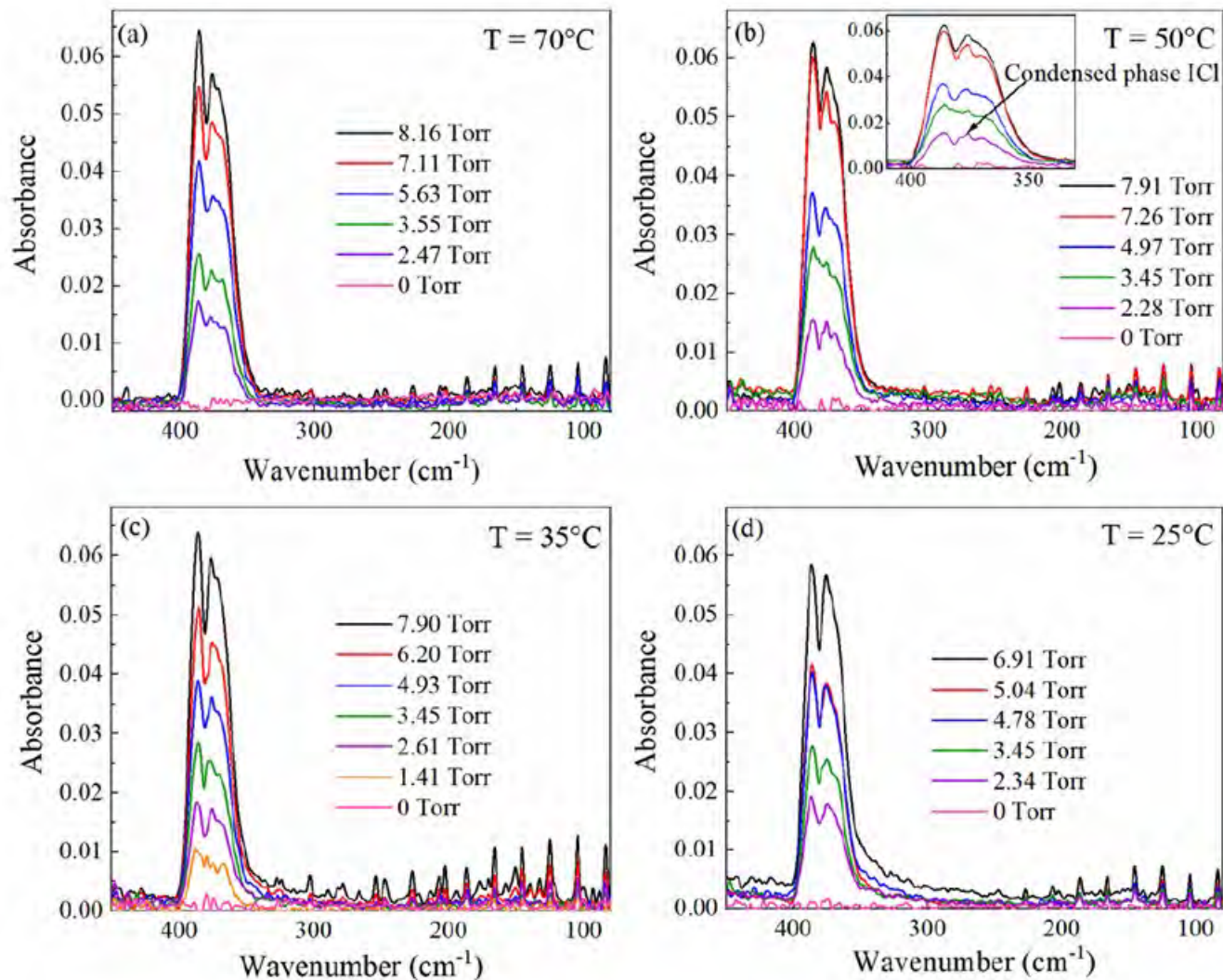


ACS Publications  
Most Trusted. Most Cited. Most Read.

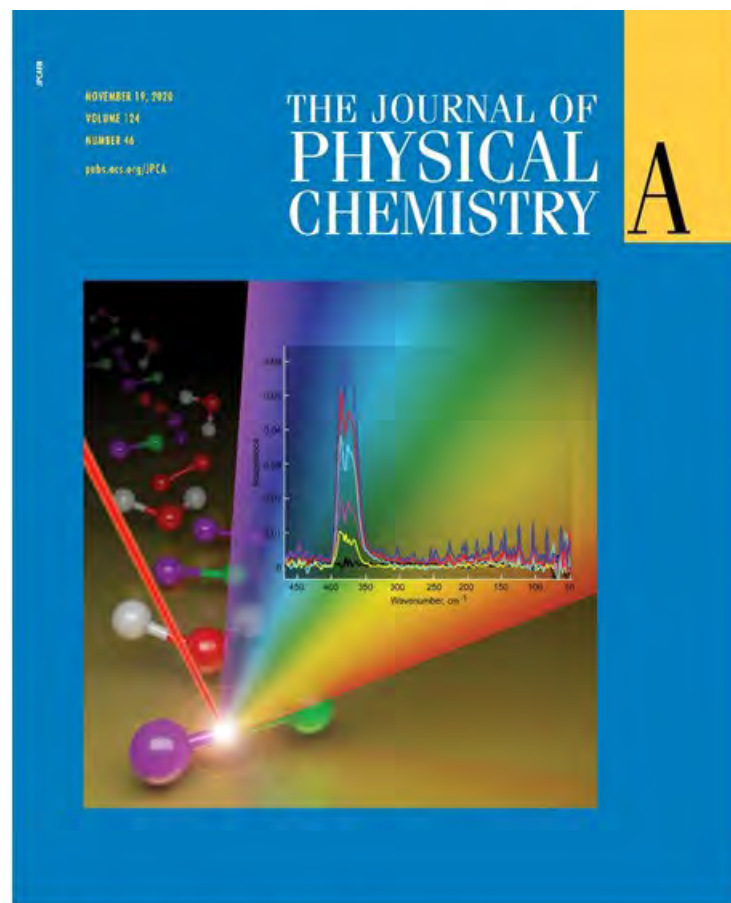
www.acs.org



# FTIR/Off-gas measurements: ICI



# Off-gas measurements: ICI



ACS Publications  
Most Trusted. Most Cited. Most Read.

HUGHEY ET AL.

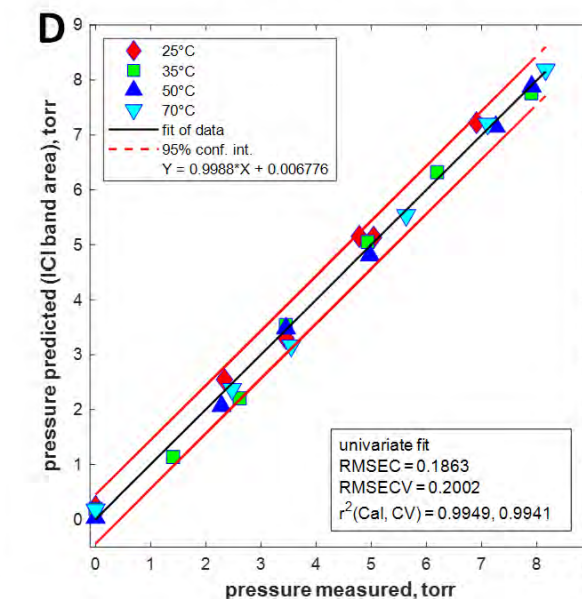
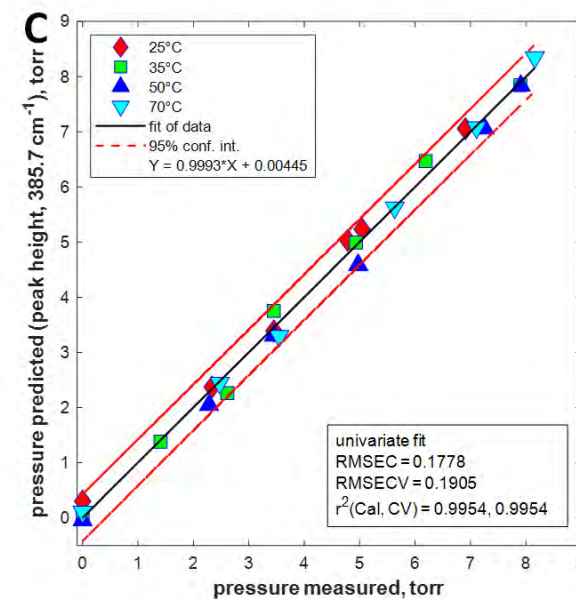
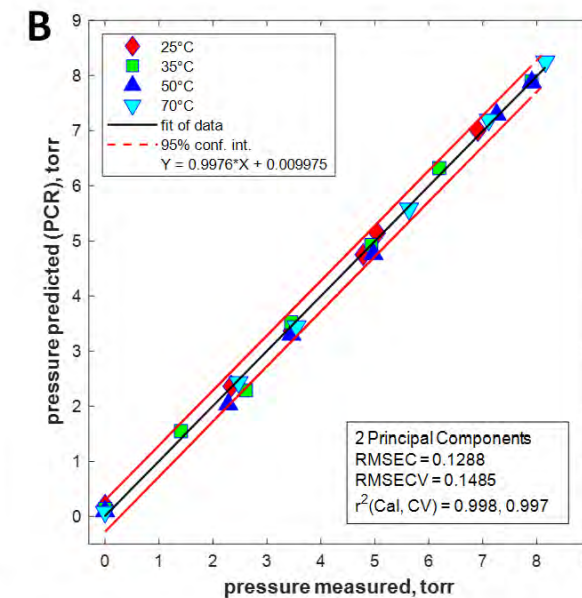
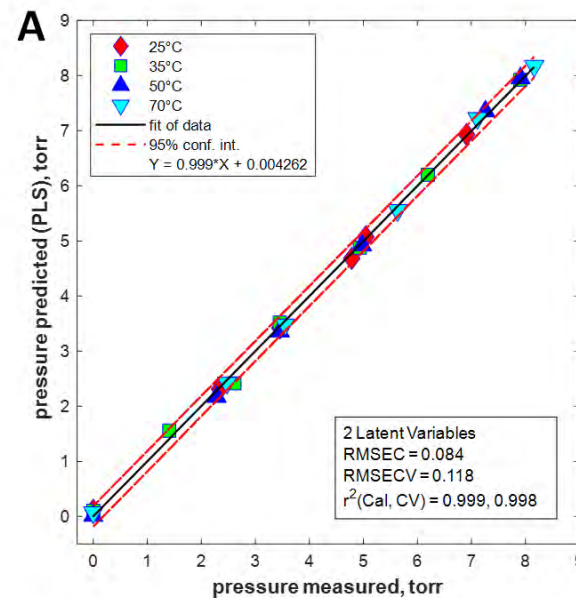
Absolute Band Intensity of the iodine Monoxide Site Functinal trace for infrared, Seeing and Quantitative Analysis

Hughey, Bradley, Tonkyn, Felmy, Blake, Bryan, Johnson, Lines,

*J Phys Chem A* 2020, 124 (46), 9578-9588.

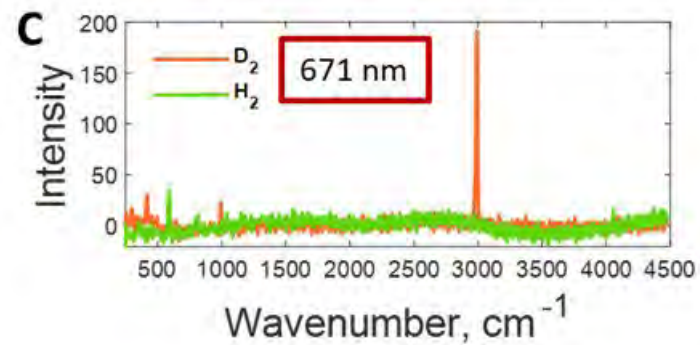
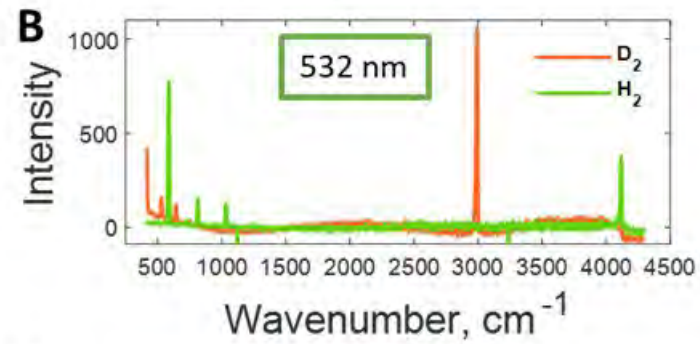
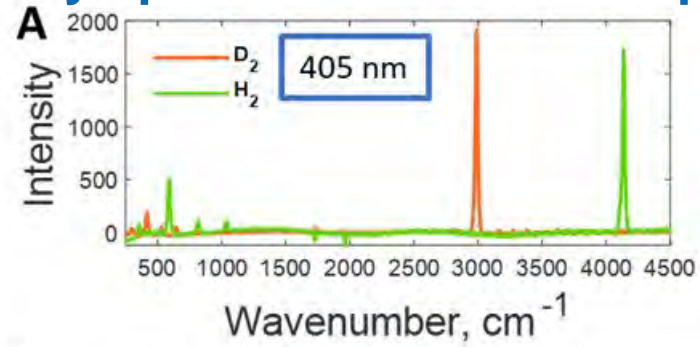
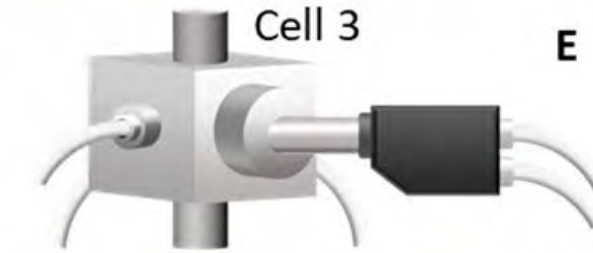
Multivariate  
methods

Univariate methods  
(Beers law)



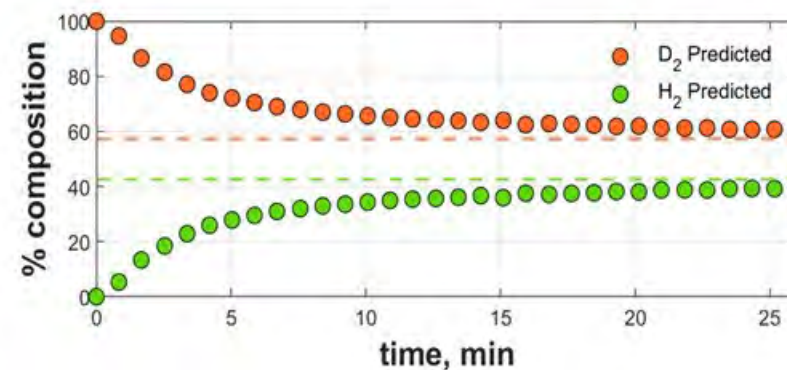
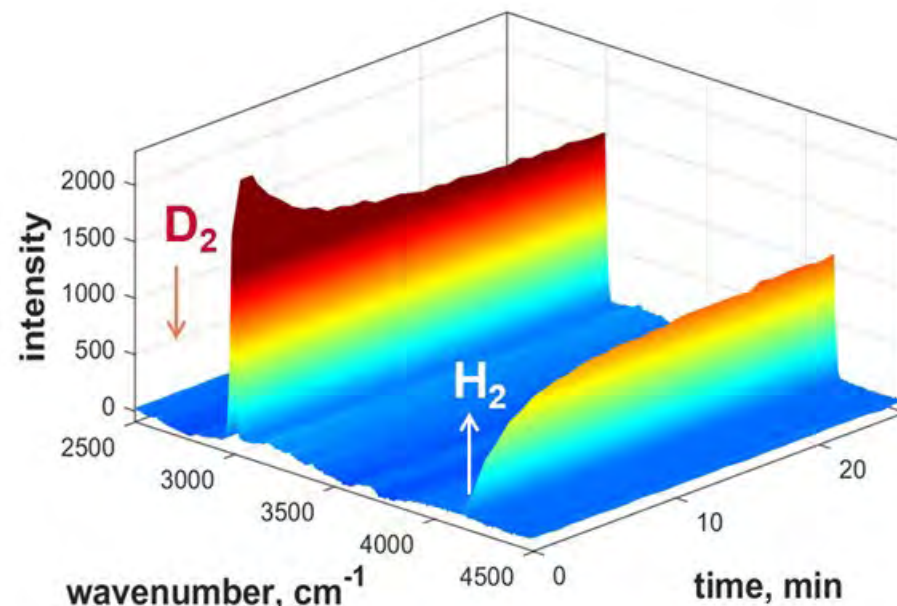
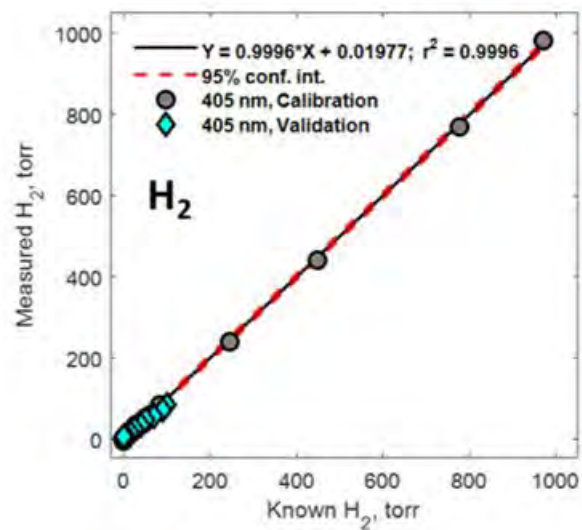
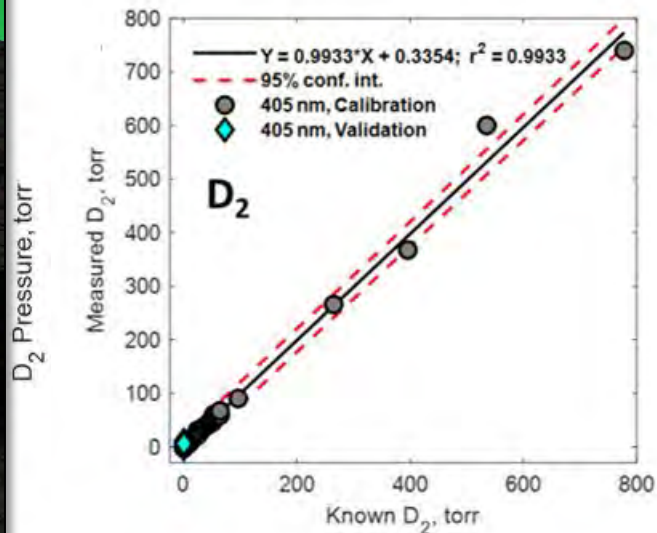
# Building a better gas cell for hydrogen isotopes

## Raman can identify and quantify speciation and H isotopics



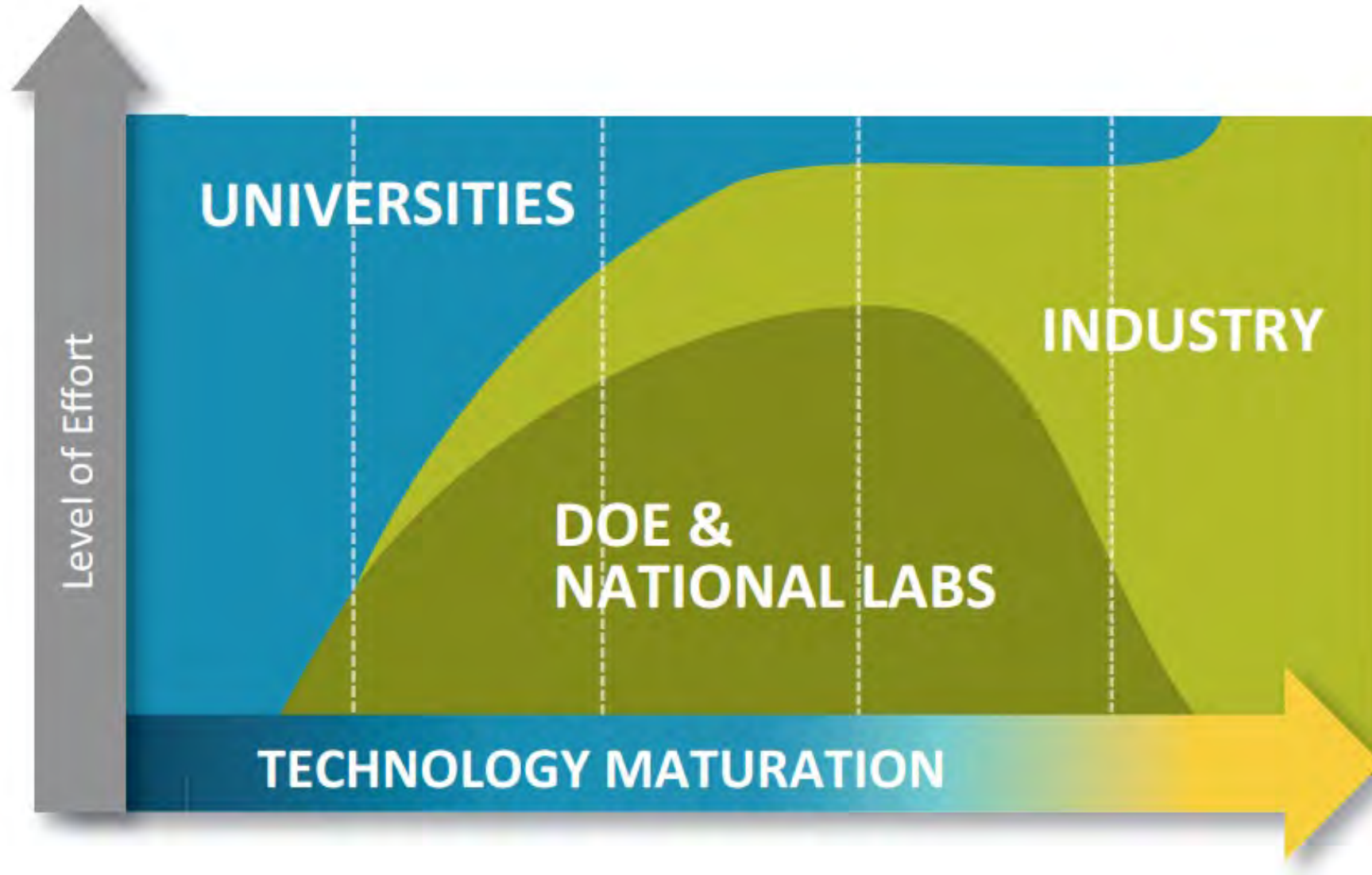
405 nm    532 nm    671 nm

# Demonstration of Hydrogen isotopes





# Building the Bridge Between Fundamental and Applied: Enabling Transfers of Technology



# Key Take-Aways

- Real-time process monitoring has diverse and wide-ranging applications
- Optical spectroscopy tools can provide powerful pathways to characterizing chemical composition of a given process (batch or flow)
  - But in any moderately complex process, advanced analytical tools are required for fast and accurate translation of data into information
- This approach can be leveraged in even some of the most challenging applications: nuclear materials processing
- Utilizing these tools can enable better, faster, safer, and more cost-effective processing in highly harsh environments

# Acknowledgements

## PNNL Team:

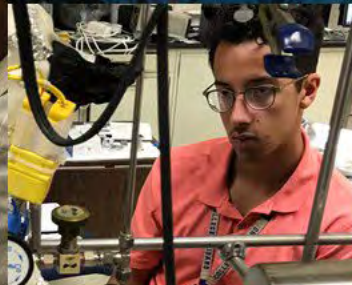
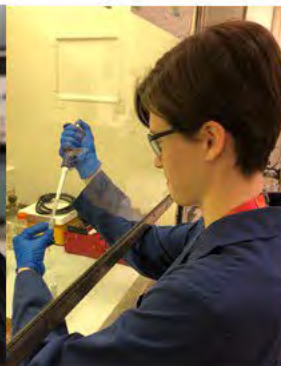
Amanda Lines  
Sam Bryan  
Heather Felmy  
Shirmir Branch  
Brian Riley

Adan Schafer Medina  
Poki Tse  
Nathan Bessen  
Gabe Hall

Gregg Lumetta  
Brienne Seiner  
Thomas Serrano  
Jarred Allred

Prof. Gilbert Nelson (C. Idaho)  
Job Bello (Spectra Solutions Inc.)  
Hope Lackey (WSU)

Alyssa Espley  
Savannah Potter  
Nikolas Boily



# Upcoming Webinars

Date	Title	Presenter
28 August 2024	International Molten Salt Research in Support of MSR Development	Aslak Stubsgaard, Denmark Isabelle Morlaes, France Ed Pheil USA Markus Piro, Canada Jeremy Pearson, USA
September 2024	Overview and Update of Sodium Fast Reactor Activities within the Gen IV International Forum	Yoshitaka Chikazawa, JAEA, Japan
October 2024	Overview and Update of Gas-cooled Fast Reactor Activities within the Gen IV International Forum	Petr Vacha, UJV, Czech Republic