

Molten Salt Reactors Taxonomy and Fuel Cycle Performance

Dr. Jiri Krepel Paul Scherrer Institut 25 January 2023













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Meet the Presenter

Dr. Jiri Krepel is a senior scientist in Advanced Nuclear Systems group of Laboratory for Scientific Computing at Paul Scherrer Institut (PSI) in Switzerland and chairman of the Steering Committee of GIF MSR project. He earned his PhD in 2006 at the Czech Technical University (CTU), Prague / Helmholtz-Zentrum Dresden-Rossendorf, Germany for his thesis entitled "Dynamics of Molten Salt Reactors." At PSI, he is the coordinator of the PSI MSR research and responsible for fuel cycle analysis and related safety parameters of Gen IV reactors. He has experience in the neutronics of liquid-metal and gas-cooled fast reactors and in neutronics and transient analysis of thermal and fast MSRs.



Email: jiri.krepel@psi.ch



Outline

- I. MSR definition and taxonomy
- II. Applicable materials cross-sections and reactor physics characterization
- III. Five Neutronic performance parameters
- IV. Breeding capability and core size estimate
- V. Self-sustaining breeding in (breed and burn) open cycle
- VI. Burnup definition for liquid fuel
- VII.Radionuclides distribution and release during accidental conditions



Definition of MSRs:

MSR is any reactor where a molten salt has a prominent role in the reactor core (i.e., fuel, coolant, and/or moderator).



IAEA Technical Report Series, Status of Molten Salt Reactor Technology, document in preparation, International Atomic Energy Agency, 2021.

MSR taxonomy



Taxonomy

Cat	egory:	Γ	Nolten Sal					
CIa I. Graphite k	sses: based MSRs	s II. Homogeneous MSRs III. Heterogeneous MSRs					IV. Other	MSRs
I. 1. Fluoride salt cooled reactors	I. 2. Graphite moderated MSRs	II. 3. Homogeneous fluoride fast MSRs	II. 4. Homogeneous chloride fast MSRs	III. 5. Non-graphite moderated MSRs	III. 6. Heterogeneous chloride fast MSRs	8	SRs eactors	s SRs ed MSRs
Salt cooled reactor with pebble bed fuel Salt cooled reactor with fixed fuel	Single-fluid Th-U breeder Two-fluid Th-U breeder Uranium converters and other concepts	Fluoride fast Th-U breeder Pu containing fluoride fast rector	Chloride fast breeder reactor Chloride fast breed & burn reactor	Solid moderator heterogeneous MSR Liquid moderator heterogeneous MSR	Heterogeneous salt cooled fast MSR Heterogeneous lead cooled fast MSR	Directly cooled MS	Hybrid moderator M Chloride salt cooled fast I	Frozen salt MSR Hybrid spectrum MS Heterogeneous gas cool

Adopted from: IAEA Technical Report Series, Status of Molten Salt Reactor Technology, document in preparation, International Atomic Energy Agency, 2021.

F.I.1. Fluoride salt cooled reactors

Types definition: Primary heat exchange: Heat convection by fuel: Fuel form:

Struct. material in core: Neutronic performance: Self-sustaining breeding: Major fuel cycle: Leakage utilization: Characteristic: *By fuel form (pebble bed vs. prismatic or compacts) In core*

No, dedicated coolant **LiF-BeF**₂ (Li is enriched to ⁷Li) TRISO-particles in graphite matrix No, graphite moderator and coolant salt are compatible

Converter

Cannot be achieved Enr. U converter

Reflector

-⁷LiF-BeF₂ has certain moderation power, hence it has negative density effect on reactivity.

- -Very low specific fuel density in some designs:
 - → Unprocessed **spent fuel is volumetric**.
 - → Increased non-fuel parasitic neutron captures.
 - → Core transparency for neutrons (neutron leakage).



Salt cooled reactor with pebble bed fuel

Salt cooled reactor with fixed fuel

F.I.2. Graphite moderated MSRs

Types definition: Primary heat exchange: Heat convection by fuel: Fuel form:

Struct. material in core: Neutronic performance: Self-sustaining breeding: Major fuel cycle: Leakage utilization: Characteristic: *By fuel cycle type (Th-U breeder or enr. U converter) Ex core*

Yes

Ac. diluted in fluorides salts, for breeders it is exclusively **⁷LiF-BeF**₂ (⁷LiF?) No, graphite moderator and coolant salt are compatible

Breeder or converter

Can be achieved, is demanding Closed Th-U or enr. U converter Reflector, multi-zone core, blanket

-Specific fuel density is higher than in Fluoride salt cooled reactors.

- -Limited graphite life-span as the only reason for its exchange.
- -Hastelloy vessel protected by graphite reflector.
- -Need of fast FPs removal and/or ²³³Pa separation to achieve self-sustaining breeding.



Single-fluid Th-U breeder Two-fluid Th-U breeder Uranium converters

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F.II.3. Homogeneous fluoride fast MSRs

Types definition: Primary heat exchange: Heat convection by fuel: Fuel form:

Struct. material in core: Neutronic performance: Self-sustaining breeding: Major fuel cycle: Leakage utilization: Characteristic: *By fuel cycle type (Th-U breeder, enr. U converter, burner) Ex core*

Yes

Ac. diluted in fluorides salts, for breeders it is typically ⁷LiF (FLiNa, FNaK?) No, homogeneous salt-filled core Breeder, converter, dedicated burner Can be achieved Closed Th-U (U-Pu), enr. U conv., burner Blanket, Reflector (Hastelloy)

-Hastelloy vessel is exposed to neutron flux and should be regularly replaced.

- -Moderation power of ⁷LiF:
 - \rightarrow Softest fast spectra.
 - \rightarrow Low transparency for neutrons.
 - \rightarrow Possibility of compact cores.

Fluoride fast Th-U breeder

Pu containing

fluoride fast recto

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F.II.4. Homogeneous chloride fast MSRs

Types definition: Primary heat exchange: Heat convection by fuel: Fuel form:

Struct. material in core: Neutronic performance: Self-sustaining breeding: Major fuel cycle: Leakage utilization: Characteristic: *By fuel cycle type (U-Pu breeder or breed & burn cycle) Ex core*

Yes

Ac. diluted in chloride salts, for breeders it is typically **Na³⁷Cl**

No, homogeneous salt-filled core

Breeder, Breed and Burn

Can be achieved

Closed U-Pu or Breed-and-Burn U-Pu Blanket, Reflector (lead?)

- -Reactor vessel is exposed to neutron flux and should be regularly replaced.
- -Absence of scattering / moderation power:
 - \rightarrow Transparent for neutrons.
 - \rightarrow Hardest spectra from all fast reactors.
 - → Large reactor cores, unsuitable for Th-U cycle.

Chloride fast breeder reactor

Chloride fast breed & burn reactor



F.III.5. Non-graphite moderated MSRs

Types definition: Primary heat exchange: Heat convection by fuel: Fuel form:

Struct. material in core: Neutronic performance: Self-sustaining breeding: Major fuel cycle: Leakage utilization: Characteristic: By moderator state (solid or liquid moderator) Ex core* Yes*

Ac. diluted in fluorides salts, for breeders it is exclusively ⁷LiF-BeF₂ (⁷LiF?) Yes, for separation of fuel salt and moderator

Converter, burner

Impossible or very demanding** Closed Th-U**, enr. U converter, burner Reflector (moderator)

-Moderator requires structural material for separation:

- → Limited life-span of separation material.
- → Determination of neutronic performance.
- * Unless if liquid moderator acts as coolant.

** Relying on low capture structural material (SiC, C composites?).



Solid moderator heterogeneous MSR

Liquid moderator heterogeneous MSR

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F.III.6. Heterogeneous chloride fast MSRs

Types definition: **Primary heat exchange**: Heat convection by fuel: Fuel form:

Struct. material in core: **Neutronic performance**: Self-sustaining breeding: Major fuel cycle: Leakage utilization: **Characteristic**:

By dedicated coolant type (salt or lead cooled) In core Usually no, dedicated coolant Ac. diluted in chloride salts, for breeders it is typically **Na³⁷Cl** Yes, for separation of fuel salt and dedicated coolant Converter, Breeder, B&B is demanding Can be achieved Closed U-Pu or enr. U converter Blanket, Reflector (lead?)

-*Coolant requires structural material for separation:*

- \rightarrow Limited life-span of separation material.
- \rightarrow Reduced neutronic performance.
- \rightarrow It provides additional scattering XS.
- \rightarrow Possibly smaller cores that homogeneous chloride fast MSRs.



Heterogeneous lead cooled fast MSR



Applicable materials cross-sections



Overview of applicable materials

Water (light & heavy): ¹H, ²H, ¹⁶O Liquid metals (sodium, lead, lead-bismuth): ²³Na, ^{nat}Pb, ²⁰⁹Bi Gases (helium, CO₂): ⁴He, ¹²C, ¹⁶O Salts (fluorides, chlorides): ⁶Li, ⁷Li, ⁹Be, ¹⁹F, ^{nat}Mg, ³⁵Cl, ³⁷Cl, ^{nat}K, ^{nat}Ca

BTW:

Capture XS: 1/v rule, i.e. capture chance depends on the time, which neutrons and nuclei spend together.

Scattering XS is rather flat and based on "geometrical" interaction.



Moderation power and capture XS

Logarithmic decrement of energy *\mathcal{\xi}* describes neutron energy loss by scattering. Product of *\mathcal{\xi}* and scattering XS is used here as a moderation power* criteria.

Thermal moderation power Thermal capture Fast moderation power Thermal scattering XS ξ Fast capture 1.000 H1 Η1 H1 Η1 H1 24 0.003 0.005 0.056 Cl35 CI35 CI35 CI35 Cl35 111 0.010 Pb-nat Pb-nat Pb-nat Pb-nat ш Pb-nat 0.010 Bi209 Bi209 Bi209 Bi209 Bi209 0.207 Be9 Be9 Be9 Be9 Be9 0.158 C12 C12 C12 C12 C12 0.120 016 016 016 016 016 0.725 H2 H2 H2 H2 H2 2.3 X 0.102 F19 F19 F19 F19 F19 0.078 Mg-nat 11 Mg-nat Mg-nat ///// Mg-nat Mg-nat 1 0.084 Na23 Na23 Na23 Na23 Na23 0.003 0.045 Ca-nat 111 Ca-nat Ca-nat Ca-nat Ca-nat 0.049 K-nat 1 K-nat K-nat K-nat K-nat 💋 0.053 Cl37 CI37 Cl37 Cl37 Cl37 1/// 0.260 Li7 Li7 Li7 Li7 Li7 0.425 He He He He /// /////// He 479 0.750 0.299 Li6 Li6 Li6 Li6 Li6 0.5 1.5 0 0.2 0.3 0.5 1.5 0.001 0.002 0.003 20 30 0 0.1 0 10 16 ξ x av. scattering XS at Av. capture XS at ξ x av. scattering XS at Av. capture XS at Av. scattering XS at 0.1eV (b) 0.1eV (b) 0.1MeV (b) 0.1MeV (b) 0.1eV (b)

*It is not a standard definition, because it uses microscopic

instead of macroscopic XS.

Thermal moderation power

Summary of materials characteristics

Based on the moderation power and capture XS, 4 coolant nuclides performance characteristics can be defined:

The	Thermal moderation power			Thermal capture				Fast moderation powe			er	
H1			2/	H1				H1				
Cl35			Z4	Cl35				Cl35	ш		11	
Pb-nat	ш			Pb-nat			22	Pb-nat				
Bi209	н			Bi209	п —			Bi209	н —			
Be9				Be9	1			Be9				
C12				C12	1			C12				
016				016				016				
H2			шцц	H2				H2				l
F19			۷.۱	F19	1			F19			2.3	
Mg-nat	11.			Mg-nat	~ .			Mg-nat	111	111		
Na23				Na23				Na23		•		
Ca-nat	ш			Ca-nat			Ш	Ca-nat	н —			
K-nat	4			K-nat	1111	/////	1114	K-nat	2			
Cl37	2			Cl37	1111	/////	/ 1.2	Cl37	2			
Li7				Li7	-			Li7				
He	1777.			He			179	He	111	1111		
Li6				Li6			775	Li6				
8	0 0 x av. sc	.5 : atterir	1 1. 1 XS a	.5 at	0 0. Av. c	.1 0 apture	.2 0. XS at	.3 8 :	0 x av. s	0.5	1 1. g XS at	5
	0.	1eV (b)		0	.1eV (b)	۰ د	0.1	LMeV (b)	

nt	Sup	opressing fas	st B	reeding in fa	st
		neutrons:	Breeding in	spectrum:	
Мо	oderator:	th	nermal spectrui	m:	
H1	Yes*	Yes	No	No	
CI35	No	No	No	No	
Pb-nat	No	No	No	Yes	
Bi209	No	No	No	Yes	
Be9	Yes	Yes	Yes	No	
C12	Yes	Yes	Yes	No	
O16	No	No	Yes	Yes	
H2	Yes	Yes	Yes	No	
F19	No	Yes**	Yes	Yes***	
Mg-nat	No	Yes**	Yes	Yes	
Na23	No	No	No	Yes	
Ca-nat	No	No	No	Yes	
K-nat	No	No	No	Yes	
CI37	No	No	No	Yes	
Li7	No	Yes**	Yes	Yes	
He4	No	No	Yes	Yes	
Li6	No	Yes**	No	No	
*S	ubstantial	**Broad Sca	attering	***However the	•
Ca	apture XS	resonance	S	spectrum is	17
	-	around 0.1	MeV	quite soft.	

Performance of structural materials

Boron (¹⁰B, ¹¹B) as a absorber,
¹⁴N as ¹⁶O alternative.
Si as part of SiC,
Aluminum, Zirconium, Iron and Nickel.



- Zirconium: similar capture XS as ¹H.
- Silicon: similar capture XS as lead. (big hope for many MSR concepts)
- Aluminum: sometimes used as metallic fuel matrix for research reactors.
- Iron (steel) can be used in fast reactors but should be avoided in thermal spectrum.
- Nickel (alloys) foreseen for MSRs because of chemical resistance have
 2x higher capture XS than iron.
- Presence in the core, as a fuel cladding:
 1) Should be avoided in thermal systems.
 2) Reduce performance of fast systems.

Characterization from reactor physics perspective



Six major MSR families

- I.1. Fluoride salt cooled reactors
- **I.2.** Graphite moderated MSRs
- **II.3.** Homogeneous fluoride fast MSRs
- **II.4.** Homogeneous chloride fast MSRs
- **III.5.** Non-graphite moderated MSRs
- **III.6.** Heterogeneous chloride fast MSRs





Reactor physics features / issues

- Double heterogeneity
- Graphite limited lifespan and positive temperature effect
- Positive coolant and blanket density effect
- Large migration area
- Fuel volumetric heat up and homogenization
- Power level and peaking in core
- Local overheating or excessive burnup
- Fission Products (FPs) circulation
- Gaseous and non-soluble FPs removal
- ²³³Pa longer half-life than ²³⁹Np
- Limited structural material lifespan

 Krepel J., Ragusa C., MSR Reactor physics: characterization, neutronic performance, multiphysics coupling, and reduced-order modeling, chapter 4, Vol. 1 of a book:
 Dolan, T., J., Molten Salt Reactors and Thorium Energy, 2nd Edition, in preparation.



(1.1) (1.2) (I.1, III.5, III.6) (I.1, I.2, II.4, partly III.6) (I.2, II.3, II.4, partly III.5, III.6) (I.1, partly III.5, III.6) (**I.2**, **II.3**, **II.4**, partly **III.5**, **III.6**) (**I.2**, **II.3**, **II.4**, optionally **III.5**, **III.6**) (1.2, 11.3, 11.4, 111.5, 111.6) (all when operated in Th-U cycle) (all families)

11.4





Neutronic performance parameters



Five fuel cycle performance parameters

I. Breeding capability

- Breeding capability
 How many neutrons can be captured by ²³²Th or ²³⁸U so that the reactor is still critical MSR: possible absence of structured.
- It is about neutron economy.

II. Achievable burnup

- Is limited by FPs neutron capture and by fuel irradiation stability.
- Depends on initial reserve of fissile material and its renewal (breeding capability)

III. Initial fissile mass

- It is determined by neutron economy and spectrum type of the reactor.
- Higher burnup may impose higher initial fissile mass reserve.

IV. Means of criticality maintenance

- Ac. irradiation and FPs creation results in reactivity oscillations / swing.
- Compensation option for reactivity swing differ between reactor types.

V. Transmutation capability

- "Neutron costs" and "speed" of synthetic actinides fission.
- Synthetic Ac. compatibility with the fuel and fabrication process.

of structural materials

Radiation stability

Possible liquid fuel

reshaping / draining

Online rafuelling and

Absence of fabrication

Solubility of actinides?

removial of some FPS

of the salt

Reactor classification by breeding capability

Neutron economy



- Burner typically relies on synthetic Actinides (Ac) and excludes fertile isotopes as ²³⁸U or ²³²Th.
- Convertor, e. g. PWR or DMSR, is usually operates in open fuel cycle and burns ²³⁵U.
- Breeder profit from neutronics advantages only in the closed cycle. For Iso-breeding (EU) or Break-even (US) reactor => A=B.
- Extreme breeder can be operated in Breed & Burn mode. It can have high fuel utilization even without reprocessing.

Major path of the ²³²Th and ²³⁸U irradiation chains

- ²³²Th and ²³⁸U irradiation chains are similar, because of the repetitiveness of actinides properties (+2p +4n).
- Nonetheless, there is the exception caused by ²⁴¹Pu fast decay (x²³⁵U).
- Furthermore, nuclides in ²³⁸U chain have more nucleons and generally slightly **shorter half-lives**.
- For the same reason, they produce **more neutrons** per fission.



Krepel, J., Self-sustaining breeding in advanced reactors: Characterization of natural resources, Encyclopedia of Nuclear Energy, (Greenspan, E., Ed.), Elsevier, 2020.



Neutron balance of the equilibrium actinide chains

- The major indicator for **breeding capability** is the **neutron balance**.
- It has several **components**: neutron leakage, neutron parasitical absorption on non-actinides materials and **neutron balance of actinides itself**.
- Neutron balance of the equilibrium actinides composition can be enumerated by:
 - 1. Eta-2 with correction factors:

Balance₁ =
$$\eta_{233U} - 2 + F_{232Th} - \frac{C_{233U}}{C_{233U} + F_{233U}} D_{234U} - \left(2\frac{C_{233Pa}}{C_{233U} + F_{233U}} + \frac{C_{233Pa}}{C_{233U} + F_{233U}} D_{234U}\right)$$

2. Nu_bar-2 with correction factors:
Balance₂ = $\overline{v} - 2 + F_{232Th} - \frac{\sum_{i} C_{i}}{\sum_{i} F_{i}} + 2R_{232Th}^{(n,2n)}$
3. Neutron costs of fission:

1

$$Balance_{3} = \overline{v} - \sum_{i} F_{i} \left(u_{i} - 232 + 1 \right) - 4 \sum_{i} \alpha_{i}$$

4. D-factor of major fertile nuclide:



$$Balance_4 = -D_{232T}$$

Neutron balance of the ²³²Th actinides chain

GEN IV International Forum	Reactor family	I.1	I.2	I.2	II.3	II.3	II.4	II.4
	Reactor type	FHR	MSBR	MSBR	MSFR	MSFR	MCFR	MCFR
Neutron balance of the -	Salt used	FLIBE	FLIBE	FLI	FLIBE	FLI	NaCl	AcCl
	Migration area	1020	360	309	194	167	1384	1769
²³² Th actinides chain	\overline{v} average	2.50	2.50	2.50	2.52	2.53	2.53	2.53
	\overline{v}_{233U}	2.50	2.50	2.50	2.51	2.52	2.53	2.54
	²³² Th fission probability (F _{232Th})	0.00	0.00	0.01	0.01	0.01	0.03	0.04
_	²³³ U fission probability (F _{233U})	0.90	0.89	0.89	0.87	0.87	0.90	0.91
_	²³³ U capture probability (C _{233U})	0.10	0.11	0.11	0.13	0.13	0.10	0.09
_	²³² Th D-factor (D _{232Th})	-0.06	-0.15	-0.12	-0.12	-0.20	-0.37	-0.43
_	²³⁴ U D-factor (D _{234U})	0.27	0.32	0.36	0.35	-0.02	-0.75	-0.96
C –	Synthetic actinides rel. capture (C _{Ac})	0.44	0.36	0.39	0.42	0.35	0.19	0.15
$nce_1 = \eta_{233U} - 2 + F_{232Th} - \frac{C_{233U}}{C} D_{234U}$	Neutron balance from η-2							
$C_{233U} + F_{233U}$	η - 2	0.25	0.24	0.22	0.18	0.19	0.28	0.31
	+ 1 correction term	0.25	0.24	0.23	0.19	0.21	0.31	0.35
$-12 - \frac{C_{233Pa}}{2} + \frac{C_{233Pa}}{2} D_{224W}$	+ 1&2 correction terms	0.22	0.21	0.19	0.14	0.21	0.38	0.44
$\begin{pmatrix} C_{233U} + F_{233U} & C_{233U} + F_{233U} \end{pmatrix}_{-}$	+ 1&2&3 correction terms	0.06	0.15	0.12	0.11	0.19	0.37	0.42
$\sum C$	Neutron balance from $\overline{v}_{_{233U}}$ - 2							
$\sum_{i} C_{i} = 2 \mathbf{p}^{(n 2n)}$	<u>v</u> _{233U} - 2	0.50	0.50	0.50	0.52	0.53	0.53	0.53
$Balance_2 = v - 2 + F_{232Th} - \frac{1}{\sum r} + 2R_{232Th}^{(n,2h)}$	+ 1 correction term	0.50	0.51	0.51	0.53	0.54	0.56	0.57
$\sum F_i$	+ 1&2 correction terms	0.06	0.14	0.12	0.11	0.19	0.36	0.42
i	+ 1&2&3 correction terms	0.06	0.15	0.12	0.12	0.20	0.38	0.44
$Balance_3 = \overline{v} - \sum F_i (u_i - 232 + 1) - 4 \sum \alpha_i$	Neutron balance from \overline{v} - fission cost	:						
	$\overline{\mathcal{V}}$ - fission cost	0.06	0.15	0.12	0.13	0.21	0.37	0.43
	+ 1 correction term	0.06	0.15	0.12	0.12	0.19	0.37	0.43
Expertise Collaboration Excellence $Balance_A = -D_{232Th}$	Neutron balance from D-factor							
	-D232Th	0.06	0.15	0.12	0.12	0.20	0.37	0.43

$$Balance_{1} = \eta_{233U} - 2 + F_{232Th} - \frac{C_{233U}}{C_{233U} + F_{233U}} D_{234U} - \left(2\frac{C_{233Pa}}{C_{233U} + F_{233U}} + \frac{C_{233Pa}}{C_{233U} + F_{233U}} D_{234U}\right)$$

Neutron balance of the ²³⁸U actinides chain

	Reactor family	I.1	l.2	l.2	II.3	II.3	II.4	II.4
	Reactor type	FHR	MSBR	MSBR	MSFR	MSFR	MCFR	MCFR
_	Salt used	FLIBE	FLIBE	FLI	FLIBE	FLI	NaCl	AcCl
_	Migration area	939	380	305	191	177	1205	1874
_	\overline{v} average	2.96	2.95	2.95	2.95	2.95	2.93	2.92
	\overline{v}_{239Pu}	2.86	2.86	2.86	2.91	2.92	2.94	2.95
_	²³⁸ U fission probability (F _{238U})	0.00	0.03	0.04	0.05	0.07	0.11	0.15
_	²³⁹ Pu fission probability (F _{239Pu})	0.63	0.63	0.63	0.62	0.66	0.79	0.87
_	²³⁹ Pu capture probability (C _{239Pu})	0.37	0.37	0.37	0.38	0.34	0.21	0.13
_	²³⁸ U D-factor (D _{238U})	0.26	0.21	0.18	-0.01	-0.24	-0.69	-0.90
_	²⁴⁰ Pu D-factor (D _{240Pu})	0.12	0.15	0.13	-0.32	-0.60	-1.10	-1.37
_	Synthetic actinides rel. capture (CAc)	1.22	1.19	1.17	1.00	0.79	0.36	0.19
)	Neutron balance from η-2							
2310 -	η - 2	-0.15	-0.15	-0.15	-0.17	-0.06	0.33	0.55
	+ 1 correction term	-0.14	-0.13	-0.11	-0.12	0.01	0.43	0.70
	+ 1&2 correction terms	-0.25	-0.24	-0.21	-0.03	0.20	0.67	0.90
234U	+ 1&2&3 correction terms	-0.26	-0.24	-0.22	-0.03	0.20	0.67	0.90
_	Neutron balance from \overline{v}_{239Pu} - 2							
	\overline{v}_{239Pu} - 2	0.96	0.95	0.95	0.95	0.95	0.93	0.92
n,2n)	+ 1 correction term	0.97	0.98	0.99	0.99	1.02	1.04	1.08
32 <i>Th</i>	+ 1&2 correction terms	-0.25	-0.22	-0.18	-0.01	0.23	0.68	0.89
_	+ 1&2&3 correction terms	-0.25	-0.21	-0.17	0.00	0.24	0.69	0.91
\overline{a}	Neutron balance from \overline{v} - fission cos	t						
	$\overline{\mathcal{V}}$ - fission cost	-0.21	-0.11	-0.08	0.26	0.44	0.75	0.93
	+ 1 correction term	-0.25	-0.21	-0.18	-0.01	0.23	0.68	0.90
	Neutron balance from D-factor							
32 <i>Th</i>	-D _{238U}	-0.26	-0.21	-0.18	0.01	0.24	0.69	0.90

$$Balance_{1} = \eta_{233U} - 2 + F_{232Th} - \frac{C_{233U}}{C_{233U} + F_{233U}} D_{234U} - \left(2\frac{C_{233Pa}}{C_{233U} + F_{233U}} + \frac{C_{233Pa}}{C_{233U} + F_{233U}} D_{234U}\right)$$

$$Balance_{2} = \overline{v} - 2 + F_{232Th} - \frac{\sum_{i}^{i} C_{i}}{\sum_{i}^{i} F_{i}} + 2R_{232Th}^{(n,2n)}$$

$$Balance_{3} = \overline{v} - \sum_{i} F_{i} (u_{i} - 232 + 1) - 4 \sum_{i}$$
GENIV International
Expertise | Collaboration | Excellence
Balance_{4} = -D_{232}

Breeding capability: comparison between ²³²Th and ²³⁸U actinides chain

- ²³⁸U actinides chain (U-Pu cycle) profits more from spectrum hardening.
- Better performance: Th-U in thermal and U-Pu in fast spectra.
- Graphite mod. MSR only in Th-U.
- Fluorides fast MSFR possible in both cycles (almost epithermal).
- Chloride fast MCFR possible in both cycles (bulky core for Th-U).
- B&B possible only for chlorides and U-Pu cycle.





Achievable burnup

- Burnup in liquid fuel will be defined later in this presentation.
- The parasitic neutron captures depends on FPs relative share.
- Fast spectrum reactors have higher fissile actinides share.
- Therefore, they can be operated with higher average FPs share.





Initial fissile mass

- Initial fissile mass can be defined as a product of core size and fissile actinides specific density.
- In general, initial fissile mass is lower in thermal reactors.
- Especially when moderated by heavy water.





Means of criticality maintenance

- Liquid fuel can allow for online FPs removal and actinides addition.
- It can also allow for unusual reactivity control methods, like salt expelling out of the core.
- For accidental conditions overflow and removal of the respective salt can be used.



Transmutation capability

- In open cycle the increase or decrease of radiotoxicity per produced unit of energy should be considered. (reprocessing losses in closed cycle)
- The "pace" of transmutation per atom is proportional to the respective cross-section.
- The neutron cost of transmutation in closed cycle can be expressed be Dfactor. (total neutron cost of given nuclide transmutation together with its daughters)





Breeding capability of moderated MSRs I.2 family for graphite III.5 family for other moderators



For comparison with other reactors refer to: Krepel, J., and Losa, E., Self-sustaining breeding in advanced reactors: Characterization of selected reactors, Encyclopedia of Nuclear Energy, (Greenspan, E., Ed.), Elsevier, 2021.

Th-U breeding capability with different moderators

- 5 fluoride salts were analyzed with
 6 selected moderators.
- Equilibrium k_{inf} is presented as a function of salt share and channel radius.
- FLi salt is neutronically the best.
- Good results for Be, BeO, and D₂O; however, they are not compatible with the salt without cladding (SiC..?).
- Hydrogen based moderators ZrH and H₂O not applicable for closed cycle.
- Graphite is not the best moderator, but the only one directly compatible with salt.
 GEN(V) International Forum



Hombourger, B.A., 2018. Ph.D. Thesis. EPFL Lausanne, Switzerland.

Neutronics impact of cladding (⁷LiF salt example)

 LiF salt combined with Be and D₂O moderators was selected to analyze the impact of cladding:

> Hastelloy, SS316, and SiC.

- Only **SiC** seems to have acceptable low parasitic neutron capture.
- Purely from neutronics perspective Heavy Water Boiling MSR would work
 HWB-MSR ③





Hombourger, B.A., 2018. Ph.D. Thesis. EPFL Lausanne, Switzerland.

Breeding capability of homogeneous fast MSRs

II.3 family for fluorides II.4 family for chlorides



Th-U cycle performance without moderator

- 8 salts were evaluated: FLi, FLiBe, FLiNa, FNaBe, FNaK, NaCl (nat), Na³⁷Cl, Ac³⁷Cl. 32% AcCl₃ 32% AcCl₃ 100% AcCl₄
- 4 options in Th-U (reasonable melting point and reactivity):
 FLi, FLiNa, FNaK, Na³⁷Cl.
- Na³⁷Cl provides the highest excess in Th-U of 13000 pcm.
- FLi is best fluoride salt with 6000 pcm.





IAEA Technical Report Series, Status of Molten Salt Reactor Technology, document in preparation, International Atomic Energy Agency, 2021.

U-Pu cycle performance without moderator

- 8 salts were evaluated: FLi, FLiBe, FLiNa, FNaBe, FNaK, NaCl (nat), Na³⁷Cl, Ac³⁷Cl. 32% AcCl₃ 32% AcCl₃ 100% AcCl₄
- 5 options in U-Pu (reasonable melting point and reactivity):
 FLi, FLiNa, FNaK, NaCl (nat), Na³⁷Cl.
- Na³⁷Cl provides the highest overall excess of 22000 pcm.
- FLi, FLiNa, FNaK have similar performance of ~6000pcm, PuF₃ solubility is the major limiting issue.



IAEA Technical Report Series, Status of Molten Salt Reactor Technology, document in preparation, International Atomic Energy Agency, 2021.

Self-sustaining breeder core size estimate



Self-sustaining breeder in closed cycle

- Using 1m Hastelloy reflector core size was estimated for single-fluid designs.
- It was compared with classical fast reactors.
- MSFR (Fli) in Th-U (4) is compact.
- MSFR (**Fli**) in U-Pu (5) is bigger.
- MCFR (Na³⁷Cl) in U-Pu (6) is comparable to MSFR in U-Pu (5).
- MCFR (**Na³⁷Cl**) in Th-U (7) is big.
- MSBR (ORNL design, 13% salt).





Hombourger, B.A., 2018. Ph.D. Thesis. EPFL Lausanne, Switzerland.

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Core radius estimate: Th-U cycle X U-Pu cycle

- MSFR with Li⁷F is the smallest MSR core and it has the same core size for both cycles. (very soft fast spectrum)
- By all other fast reactors U-Pu cycle provides smaller cores.
- SFR is the most compact bare iso-breeding core in both cycles.
- MCFR is the biggest bare iso-breeding core in both cycles.
- MSFR with BeF₂-Li⁷F is subcritical for U-Pu cycle.

GEN(IV International Expertise | Collaboration | Excellence



Core radius estimate in Th-U cycle

• Combining these two equations:

$$k_{eff} \cong k_{inf} \frac{1}{1 + M^2 B^2}$$
$$BG_{per} \cong \overline{\nu} \frac{k_{eff} - 1}{k_{eff}}$$

 Bare core size can be estimated for several BG values. Th-U cycle =>





Core radius estimate in U-Pu cycle

• Combining these two equations:

$$k_{eff} \cong k_{inf} \frac{1}{1 + M^2 B^2}$$
$$BG_{per} \cong \overline{\nu} \frac{k_{eff} - 1}{k_{eff}}$$

 Bare core size can be estimated for several BG values. U-Pu cycle =>





Self-sustaining breeding in open cycle (B&B)



Illustration of tap-like reactor https://www.subpng.com/



Breed & Burn cycle and burnup

- Initially fertile fuel will be loaded, then the fissile fuel will be bred and firstly later it will be burned.
- The B&B cycle in liquid fuel reactor substantially differs from solid fuel.
- Discharged fuel: Most burned in solid fuel case Average burned in liquid fuel case.
- There is fuel residence time distribution=>
- To increase the burnup and reduce the core size (single-fluid layout can be bulky), multi-fluid layout can be used.





Trivial criteria for breed-and-burn cycle operation

- In B&B cycle conditions:
 1) fresh fuel is only fertile material
 2) spent fuel is not recycled.
- B&B trivial criterion (tautology): I = II
 I: Fissile Fuel F_F share in the discharged fuel.
 II: New fissile fuel bred in the discharged fuel.

•
$$F_F = B(CR - 1) => \frac{1}{CR - 1} = \frac{B}{F_F}$$

where CR is conversion ratio and B is the fuel burnup.

- Reactor must be critical for CR, F_F , and B, e.g. for CR=1.2: F_F =10% \Leftrightarrow B=50% (1% \Leftrightarrow 5%)
- Fuel utilization in B&B cycle?
 It is equal to the burnup.



J. Krepel, B. Hombourger, E. Losa, Fuel cycle sustainability of Molten Salt Reactor concepts in comparison with other selected reactors, PHYTRA4, Marrakech, Morocco, September 17-19, 2018.

Self-sustaining breeder in open cycle (B&B)

- B&B is practically not possible in Th-U cycle.
- It is only possible in mixed U-Pu & Th-U cycle.
- B&B cores are bulky (chlorides = hard spectrum, but also high Migration area).
- The performance increases with growing actinides share in the core.





Hombourger, B. et al., 2019. Breed-and-Burn Fuel Cycle in Molten Salt Reactors. Submitted to special MSR edition of The European Physical Journal

Closed cycle

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Core size comparison for closed and open B&B cycle



Critical core sizes



Hombourger, B.A., 2018. Ph.D. Thesis. EPFL Lausanne, Switzerland.



• For solid fuel burnup is defined as:

 $B_{GWd/tHM}\left(t\right) = \frac{\int_{0}^{t} P(t) dt}{M_{Ac}\left(0\right)}$

 $B_{FIMA\%}\left(t\right) = \frac{\int_{0}^{t} F\left(t\right) dt}{N_{Ac}\left(0\right)}$

$$FPs_{share} = \frac{M_{FPs}(t)}{M_{Ac}(0)} = \frac{M_{FPs}(t)}{M_{Ac}(t) + M_{FPs}(t)} = B_{FIMA\%}(t)$$



- For liquid fuel two definitions are possible:
- Differential

$$\begin{split} B_{GWd/tHM}\left(t\right) &= \frac{P\left(t\right)}{\dot{M}_{Ac,in}\left(t\right)} \\ B_{FIMA\%}\left(t\right) &= \frac{F\left(t\right)}{\dot{N}_{Ac,in}\left(t\right)} \cong \frac{\dot{N}_{Ac,in}\left(t\right) - \dot{N}_{Ac,out}\left(t\right)}{\dot{N}_{Ac,in}\left(t\right)} \cong \frac{\dot{N}_{FPs,off-gas}\left(t\right) + \dot{N}_{FPs,out}\left(t\right)}{\dot{N}_{Ac,in}\left(t\right)} \end{split}$$

Integral

$$B_{GWd/tHM}(t) = \frac{\int_{0}^{t} P(t)dt}{M_{Ac,core}(0) + \int_{0}^{t} \dot{M}_{Ac,in}(t)dt}$$

$$B_{FIMA\%}(t) = \frac{\int_{0}^{t} F(t)dt}{N_{Ac,core}(0) + \int_{0}^{t} \dot{N}_{Ac,in}(t)dt} = \frac{\int_{0}^{t} F(t)dt}{N_{Ac,core}(t) + \int_{0}^{t} \dot{N}_{Ac,out}(t)dt + \int_{0}^{t} F(t)dt}$$
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• Due to the continuous FPs removal by off-gas system, the burnup and the fission products share in the core differs.

$$FPs_{share,core} = \frac{M_{FPs,core}(t)}{M_{Ac,core}(t) + M_{FPs,core}(t)} \neq B_{FIMA\%}(t)$$





Santora, J., 2022, Assessment of core minimization options for breed-and-burn molten chloride fast reactors, MSc thesis, EPFL Lausanne.

- The differential and integral definition provide different values.
- The integral definition includes the initial core loading.
- For stabilized and long enough operation, they can be equal.





Krepel, J., et al., 2022, Characterization of the Molten Chloride Fast Reactor fuel cycle options, proceedings of FR22, IAEA.

Radionuclides distribution and release during accidental conditions



Fuel reprocessing

- Many MSR concepts rely on gaseous FPOs removal and fuel salt reprocessing.
- As an example the EVOL and MARS projects benchmark is taken here.
- The active core is divided into blanket and fuel salt.
- Gaseous and volatile FPs
 Z = 1, 2, 7, 8, 10, 18, 36, 41, 42, 43, 44, 45, 46, 47, 51, 52, 54 and 86 are removed with 30s cycle time.
- For later use: Zr (Z=40) is not included in volatile FPs.
- Fuel salt is reprocessed with cycle time of 450 days:
 Z = 30, 31, 32, 33, 34, 35, 37, 38, 39, 40, 48, 49, 50, 53, 55, 56, 57, 58, 59, 60, 61, 62, 63, 64, 65, 66, 67, 68, 69, 70





Brovchenko, M., et al,. 2019, Ph.D. Neutronic benchmark of the molten salt fast reactor in the frame of the EVOL and MARS collaborative projects, EPJ Nuclear Sci. Technol.

Radiotoxicity distribution in core, blanket, reprocessing unit and off-gas system.

Ingestion radiotoxicity after 200 EFPD of operation per 1 m³ of core volume divided into FPs chains and zones.

Rank	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15
Atomic number	133	85	88	135	129	130	87	131	127	138	233	105	143	84	125
Nuclides	133In	85Ga	88Ge	135Sn	129In-m	130Pd	87Ge	131Cd	127Cd	138Sn	233Th	105Zr	143Xe	84Zn	125Ag
	133Sn	85Ge	88As	135Sb	129In	130Ag	87As	131In	127In-m	138Sb	233Pa	105Nb	143Cs	84Ga	125Cd
	133Sb	85As	88Se	135Te	129Sn-m	130Cd	87Se	131Sn	127In	138Te		105Mo	143Ba	84Ge	125In-m
	133Te-m	85Se-m	88Br	135I	129Sn	130In	87Br	131Sb	127Sn-m	1381		105Tc	143La	84As	125In
	133Te	85Se	88Kr	135Xe-m	129Sb	130Sn	87Kr	131Te-m	127Sn	138Xe		105Ru	143Ce	84Se	125Sn-m
	1331	85Br	88Rb	135Xe	129Te-m	130Sb-m	87Rb	131Te	127Sb	138Cs-m		105Rh-m	143Pr	84Br-m	125Sn
	133Xe-m	85Kr-m	88Sr	135Cs	129Te	130Sb	87Sr-m	131I	127Te-m	138Cs		105Rh	143Nd	84Br	125Sb
	133Xe	85Kr			1291	130Te		131Xe-m	127Te						125Te-m
Half-lives	0.18s	0.087s	0.129s	0.418s	1.23s	-	0.134s	0.106s	0.4s	-	22m	0.493s	0.30s	-	0.334s
	1.44s	0.250s	0.135s	1.71s	0.63s	-	0.8s	0.28s	3.73s	0.173s	27d	3.0s	1.78s	0.098s	0.68s
	2.5m	2.03s	1.50s	19.0s	6.9m	0.20s	5.6s	39s	1.14s	1.4s		36s	14.3s	1.2s	12.2s
	55.4m	19s	16.4s	6.57h	2.4m	0.29s	55.9s	23.0m	4.15m	6.5s		7.6m	14.1m	5.5s	2.36s
	12.4m	39s	2.84h	15.3m	4.40h	3.7m	1.27h	1.35d	2.12h	14.1m		4.44h	1.38d	3.3m	9.5m
	20.8h	2.87m	17.7m	9.10h	33.6d	6.5m	10.67l	25.0m	3.84d	2.9m		40s	13.57d	6.0m	9.63d
	2.19d	4.48h	stable	2.3e6y	1.16h	38.4m	2.81h	8.040d	109d	32.2m		35.4h	stable	31.8m	2.758y
	5.243d	10.73y			1.6e7y	stable		11.9d	9.4h						58d
Total ingestion	245.44	2 EE . 44	24E.44	4 45.44	4 45.44	1 05.11	4 65.40	445.40	2 7E . 40	2 2E . 40	4 25.40	4 25.40	4 4 5 . 40	0.05.00	0 7E . 00
radiotoxicity (Sv/m ³)	2.16+11	2.3E+11	2.1E+11	1.46+11	1.40+11	1.02+11	4.00+10	4.1E+10	2.7E+10	2.3E+10	1.3E+10	1.3E+10	1.12+10	9.90+09	8./E+U9
Off-gas system (%)	99.9	98.9	99.6	99.6	91.5	89.1	99.5	80.7	99.7	97.6	0.0	98.1	0.0	88.6	100.0
Fuel in core (%)	0.1	1.1	0.4	0.4	8.5	10.9	0.5	19.3	0.3	2.4	91.0	1.9	99.6	11.4	0.0
Reproc. unit (%)	0.0	0.0	0.0	0.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.3	0.0	0.0
Fuel in blanket (%)	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	8.6	0.0	0.1	0.0	0.0

Simulation of severe accident in MSFR with salt spill

- Assuming simple scenario of fuel salt spill to the bottom of the containment the radiotoxicity released as vapour and aerosols can be calculated.
- Linear heat up of the salt to 1500K in 2 hours was assumed.
- PSI in-house code GEMS was used to calculate the thermo-dynamics properties and loosely coupled to the MELCORE core (cGEMS).
- GEMS relies on the HERACLES database.





From the SAMOFAR Final meeting, E. Merle et al.

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Kalilainen, J., et al., 2020, Evaporation of materials from the molten salt reactor fuel under elevated temperatures, https://doi.org/10.1016/j.jnucmat.2020.152134

Data

0 0 1

File

MELCOR \checkmark Data \checkmark

cGEMS

Help an

Reference

TDB 8

Project

Data Bas

& Tools

GEMS

GEM Input

TSolMod: Mixing TKinMet: Kinetics

> Kernel Coo (GEMS3K

Process a

Reactive

Transpo

GEN IV International Forum

Heracles database extension

• The respective database for GEMS code was extended for the purpose of the simulation:

Species	Changes Made		0-	••••	CsF(g)
ThCl ₄	Imported as is from literature				LaF3(g)
Np	Imported as is from literature		-2-		Li2F2(g)
PuCl ₃	Adjusted previously existing data entry to conform with literature melting point				Li3F3(g)
UCl ₃	Missing liquid phase data manually matched based on literature values	(a)	_1-		LIF(g)
NpF ₃	Missing liquid phase constructed from melting-/boiling points and similarity to UF ₃	og	-4		NPF4(g) PuF4(a)
AmF ₃	Solid adjusted and liquid designed from assumed similarity to UF ₃	ζ,			SrF2(q)
ZrF ₄	Imported as is from literature	vit	-6		ThF3(g)
NdCl ₃	Imported as is from literature	cti			ThF4(g)
PrCl ₃	Imported as is from literature	Ă	-8-		UF4(g)
PrF ₃	Imported as is from literature			-	UF5(g)
Na ₂ ThCl ₆	Created in GEMS function ReacDC		10		ZrF4(g)
Pr	Imported as is from literature		-10-		in
A 1 1*				~ ~	

Additional Changes were made to: NPF₄, NdF₃, SrF₂, LaF₃, CeF₃, BaF₂, CsF



Compounds activity (proportional to vapor pressure) as a function of temperature

1100

Temperature, K

1000

1200

1300

F_sc_1.000

BaF2(g)

900

-12+800

Dietz, J., et al., 2022, MSR fuel cycle and thermo-dynamics simulations, proceedings of FR22, IAEA.

1400

1500

Total released mass during the accident (salt heat up from 800°C to 1500°C)



Fission products species Kalilainen, J., et al., 2020, Evaporation of materials from the molten salt reactor fuel under elevated temperatures, https://doi.org/10.1016/j.jnucmat.2020.152134

Characterization of released activity in form of aerosols and vapors





Total released activity in form of aerosols (left) and vapors (right) during the accident (salt heat up from 800°C to 1500°C)

Kalilainen, J., et al., 2020, Evaporation of materials from the molten salt reactor fuel under elevated temperatures, https://doi.org/10.1016/j.jnucmat.2020.152134

Major radiotoxicity component

 Based on the applied benchmark reprocessing scheme, ZrF₄ in form of aerosols seems to be the major activity carrier during the postulated accident.



during the accident (salt heat up from 800°C to 1500°C) (t=30'000s) of the accident (salt heat up from 800°C to 1500°C)

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Thank you for your attention





Adopted from: IAEA Technical Report Series, Status of Molten Salt Reactor Technology, document in preparation, International Atomic Energy Agency, 2021.

Upcoming Webinars

Date	Title	Presenter
22 February 2023	Safe Final Disposal of Spent Nuclear Fuel in Finland	Mr. Mika Pohjonen and Ms. Mari Lahti, Posiva, Finland
30 March 2023	Advanced Reactor Safeguards and Materials Accountancy Challenges	Dr. Ben Cipiti, Sandia National Laboratories, USA
05 April 2023	Overview of Nuclear Graphite R&D in Support of Advanced Reactors	Dr. Will Windes, Idaho National Laboratory, USA

