

Overview of Nuclear Graphite R&D in Support of Advanced Reactors

Dr. Will Windes

Idaho National Laboratory, USA

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Meet the Presenter

Dr. Windes has over 35 years' experience in extreme materials research with the majority being in nuclear materials. His material interests range widely from solid oxide fuel cell development to space nuclear propulsion systems to spent nuclear fuel issues. However, his focus for the past 20 years has been in the areas of nuclear graphite and carbon-based composite materials for the new High Temperature Reactor design. As the Advanced Reactor Technologies graphite program technical lead, he has overseen the large Advanced Graphite Creep (AGC) irradiation experiment at INL, developed one of the largest unirradiated nuclear graphite material property databases, is the current chair in developing ASME graphite code, and has numerous interactions with the NRC, international organizations, and commercial HTR vendors on graphite related issues. Dr. Windes holds a doctorate in Material Science from the University of Idaho and a Master and Bachelor in Nuclear Engineering from the University of Illinois and UC Santa Barbara, respectively.

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Acknowledgements

Discussion Points

- **Why are we talking about graphite?**
- **What makes graphite tick?**
	- **Graphite crystal and microstructure (not to detailed)**
	- **Anisotropy and pores**
	- **Engineered composite**
- **Graphite behavior**
	- **Irradiation**
	- **Oxidation**
- **Some speculation on graphite behavior**
	- **Interesting irradiation mechanisms**
	- **Why graphite can not, will not burn**
	- **Molten salt interactions (a prelude)**

The Nuclear Renaissance

(Since about 2001) the term nuclear renaissance has been used to refer to a possible nuclear power industry revival:

- Rising fossil fuel prices
- Limiting greenhouse gas emission

Generation IV reactor designs

- Inherently & Passively safe
	- Natural shutdown and cooling from design
- New designs = new uses
	- Process heat
	- Small modular designs
	- Variety of coolants and fuels

Hea

On schedule for deployment in next few years

LOTUS – Laboratory forOperating and Testing in the U.S.

(Very) High Temperature Reactor (HTR)

Pebble Bed Prismatic

High Technology Readiness (Previous graphite Rx)

Dragon Reactor 1965 to 1976

THTR 1983 to 1989 **Magnox Reactors** 1956 to 2015

Advanced Gas Reactors 1967 to present

> **Peach Bottom** 1966 to 1974

Fort St Vrain 1979 to 1989

AVR 1967 to 1988

HTR-PM 2022 to present

2003 to present

HTTR HTR-10 2002 to present

Renaissance is happening now

• Coolant

• Size

heat pipe

– TRISO, UO2, UN, other

• Fuel type

What started in 2001 as government funded R&D projects has evolved into numerous commercial enterprises world-wide

'a Safe Nuclear

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Graphite is important

(V)HTR: one of the 1st GEN IV Designs

- Highest technology readiness
- worldwige under development under development and the unit of the • *Graphite* is important material for GEN IV design

Contract Contract Contract

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Graphite's unique structure key to behavior

Begins with benzene-like ring of carbon atoms – covalent bonds

Graphene basal planes – a 2D "chicken-wire"

Stack up covalently bonded graphene planes

Basal planes stacked to form an … ABAB… stacking structure

Graphite Crystal Structure

 3.354\AA

a

 $1.421A$

 (A)

- Hexagonal close pack (HCP)
	- A-B-A-B stacking of basal planes
	- Covalent within basal planes,
	- Van der Waals (electronic) between basal plane

Completely anisotropic

- Huge differences between c-axis and a-axis directions
	- c-axis CTE \sim 26 x 10⁻⁶ / $\rm K$ a-axis CTE \sim 1 x 10⁻⁶ / $\rm K$
	- c-axis thermal conductivity ~ 6.8 W/m K : a-axis conductivity ~ 2000 W/m K

What do we do? Make a composite

- Long fabrication process
	- Filler particles: petroleum coke/pitch coke
		- Grind filler particles to desired size grain size
	- Bind the particles: pitch-based liquid binder
		- Mix filler (grains) randomly in liquid binder
	- Form into final billet
		- Fabrication by extrusion, vibrationally molded, or isostatically molded
	- Baked green billet is heterogenous mix of filler particles bound by carbonaceous binder phase
		- Multiple pitch impregnations to increase density
	- Graphitize at >2200 °C
		- Most grades are heat treated to >3200 °C

Microstructure of synthetic graphite

- Three phases in graphite microstructure
	- **Filler phase**
	- **Binder phase**
	- And Pore phase
- Nuclear graphite grades ~20% total porosity
	- Pores and pore structure define graphite behavior
	- Pore size range : nm to mm

• **Must. Have. Pores**

- More accurately : must have accommodating porosity
- Pores occur at all length-scales

Crystal length-scale

Grain-size length-scale

To the sample and component level (100X)

A "near" isotropic material

Material Properties

- Near-isotropic material response
- High thermal stability > 3000°C
	- *Well above any accident temperatures*
- High heat capacity (thermal sink)
- High thermal conductivity (better than metal)
- Density: 15% 20% porosity
- Purified graphite: Low activation (Medium waste)
- Chemically inert (Molten salt)
- Neutron moderator (thermal designs)
- Easy machinability / cheap material
- High compressive / Low tensile strength
	- *Ceramic composites for tensile*
- Ceramic like material response
	- *Low fracture toughness (~ 1-2 MPa √m)*
	- *Quasi-brittle cracking*

Component Behavior

Decent irradiation response

- Smooth dimensional change
	- o *Life-limiting mechanism*
	- o *Multiple decades ofsafe operation*
	- o *And even longer at lower temperatures*
- Generally gets stronger with irradiation
- Isotropy stays relatively constant
- Thermal stability and capacity are unaffected

Oxidation and molten salt intrusion

- Graphite does oxidize at all temperatures *But it does not burn!*
- Oxidation and molten salt behavior depends on pore structure

There is **no** "nuclear graphite" fabrication standard All grades are proprietary

Degradation: Oxidation Behavior

- Internal Oxidation:
	- Air/oxygen into the pore structure
	- Oxidation rate is temperature dependent
		- *Oxidation occurs at all temperatures*
- Oxidation Effects
	- Internal oxidation has larger effect on residual strength
	- High temperature oxidation only attacks outside of component

Oxidation mechanisms

- **Only on the Edges**
	- **Oxygen chemical reaction occurs only on the edge of basal planes**
	- **Covalent bonds within planes are too energetic**
- **RSA sites**
	- **Reactive Surface Area**
	- **Most chemically reactive**
		- *Fission product and other impurities end up here*
- **Can not sustain reaction**
	- **No fires!**

Irradiation Behavior

Significant changes occur during normal operation:

- Density Densification
	- *Graphite gets denser with irradiation until Turnaround dose*
	- *After Turnaround density decreases (volumetric expansion)*
	- *Formation of microcracks (molten salt consideration)*
- Dimensional change
	- *Turnaround dose is key parameter*
	- *Highly temperature dependent*
- Strength and modulus
	- *Graphite gets stronger with irradiation …*
	- *Until Turnaround dose is achieved. It then decreases*
- Coefficient of thermal expansion
	- *Initial increase but then reduces before Turnaround*
	- *CTE is why properties are so temperature dependent*
- Thermal diffusivity
	- *Decreases immediately to ~30% of unirradiated values*
	- At *high temperatures it is same* as *unirradiated* conductivity $\boxed{27}$

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120

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Irr. Dimensional change → **life limiting mechanism**

Internal stresses for crack formation develop from irradiation dimensional change

- Densification to volumetric expansion
	- *As dose increases, dimensional change creates internal stresses*
	- *Temperature accelerates dimensional change*
	- *Internal stress buildup at interface of dimensional change*
	- *Changes after turnaround most limiting*

G. Haag," Properties of ATR-2E Graphite and Property Changes due to Fast Neutron Irradiation", Juel-4183, 2005

Mechanisms underlying irradiation damage

- Ballistic event physically displaces atoms for lattice position
- Sub-plane formation, vacancy clusters

Model describing irradiation damage

- Ballistic event physically displaces atoms for lattice position
- Sub-plane formation, vacancy clusters

- Sub-plane formation between basal planes
- Vacancy cluster/loop collapse

- **Crystallites shrink parallel to basal planes**
- **They expand perpendicular to planes**

Mrozowski cracks (Accommodating porosity) *Basal planes (Graphene planes)* Mrozowski cracks (Accommodating porosity)

Model describing irradiation dimensional change

Observations: Crack closure

Observations: c-axis expansion

• Buckle, ruck & tuck

- **Buckling of basal planes.**
- Not sub-plane formation.

• New techniques offer new data for mechanisms

- Defect formations
- Dimensional change
- **Property changes**

• But we must understand limitations of new techniques

Graphite does not, can not burn

Graphite does not, can not burn

It oxidizes but can not sustain the reaction

- *Only outer edges of crystals can react*
	- *Analogy: burning a thick paper book*
- *Reactive Surface Area (RSA) sites on the edges*

Graphite "Fires" and "Explosions"

Maize Dust

Graphite "Fires" and "Explosions"

Graphite Dust

(Potential) Molten Salt Issues

Large molten salt tests are being initiated

- Salt impregnation into graphite pores
	- Physical damage/cracks
	- "Hot spots" from fueled molten salt
- Wear/abrasion/erosion
	- Molten salt has higher density than graphite
	- Liquid flow over soft graphite has potential
- Chemical coupling with metallic systems
	- Graphite MS is inert
		- *Fluorination questions remain.*
	- There are questions when a metallic component is added to the MS system

Dr. Nidia Gallego on 24 May 2023

Salt residue

After immersion in FLiNaK

Before immersion in FLiNaK

In summary

- **Graphite is an important material for future GEN IV reactor designs**
	- *Carbon-based materials*
	- *Ceramic composites*
	- *Ceramics*
- **Much work still remains in characterizing nuclear graphite**
	- *Chronic oxidation behavior*
	- *Irradiation behavior*
	- *Molten salt interactions*
	- *New coolant and fuel systems for advanced reactor concepts*

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