

Overview of Nuclear Graphite R&D in Support of Advanced Reactors

Dr. Will Windes

Idaho National Laboratory, USA

5 April 2023



Argon

National Labo





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Meet the Presenter

Dr. Windes has over 35 years' experience in extreme materials research with the majority being in nuclear materials. His material interests range widely from solid oxide fuel cell development to space nuclear propulsion systems to spent nuclear fuel issues. However, his focus for the past 20 years has been in the areas of nuclear graphite and carbon-based composite materials for the new High Temperature Reactor design. As the Advanced Reactor Technologies graphite program technical lead, he has overseen the large Advanced Graphite Creep (AGC) irradiation experiment at INL, developed one of the largest unirradiated nuclear graphite material property databases, is the current chair in developing ASME graphite code, and has numerous interactions with the NRC, international organizations, and commercial HTR vendors on graphite related issues. Dr. Windes holds a doctorate in Material Science from the University of Idaho and a Master and Bachelor in Nuclear Engineering from the University of Illinois and UC Santa Barbara, respectively.

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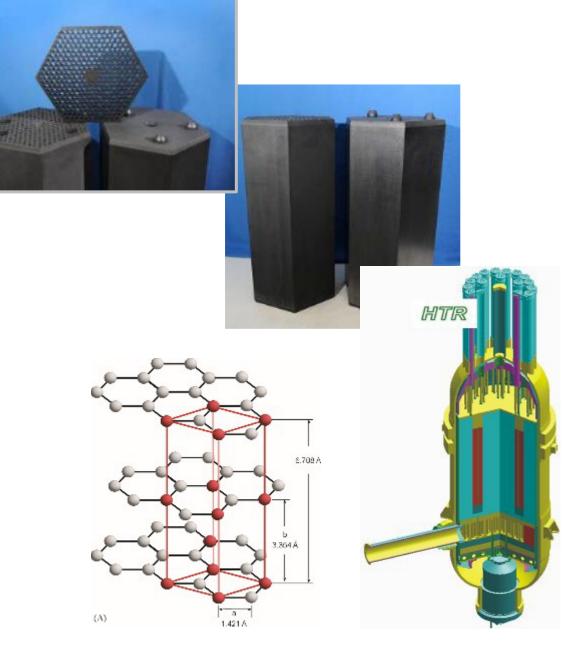


Acknowledgements

| Researcher | Expertise | Researcher | Expertise |
|------------------------------------------------|--------------------------------------------------|----------------------------------------------------------|-----------------------------------------------------|
| Andrea L. Mack andrea.mack@inl.gov | ASME Code | Mary Kaye Aimes <u>marykaye.ames@inl.gov</u> | Oxidation, Material testing |
| Anne Campbell <u>campbellaa@ornl.gov</u> | PIE, Irradiation damage, Irradiation behavior | Michael E. Davenport <u>michael.davenport@inl.gov</u> | Irradiation experiments |
| Arvin Cunningham arvin.cunningham@inl.gov | Oxidation, Split-disk testing | Nidia C. Gallego gallegonc@ornl.gov | Molten salt technical lead, irradiation damage |
| Austin C. Matthews austin.matthews@inl.gov | Material property testing, PIE, Oxidation | Paul, Ryan paulrm@ornl.gov | Oxidation, graphite manufacturing |
| David T. Rohrbaugh david.rohrbaugh@inl.gov | Unirradiated and Irradiated material properties | Philip L. Winston philip.winston@inl.gov | Irradiation experiments |
| Jose' D. Arregui-Mena arreguimenjd@ornl.gov | Microstructure, irradiation damage | Rebecca E. Smith <u>rebecca.smith@inl.gov</u> | Graphite oxidation (irr. and unirr) |
| Joseph L. Bass Joseph.Bass@inl.gov | Behavior Modeling | Steve Johns Steve.johns@inl .gov | Irradiation damage, Characterization, Split-disk |
| Lu Cai Lu.Cai@inl.gov | Pebble Oxidation | William Windes <u>william.windes@inl.gov</u> | Irradiation behavior, ASME |
| Martin Metcalfe martin.p.metcalfe@gmail.com | HTR operations, ASME, ASTM | Wilna Geringer geringerjw@ornl.gov | ASME, Composites, Graphite |

Discussion Points

- Why are we talking about graphite?
- What makes graphite tick?
 - Graphite crystal and microstructure (not to detailed)
 - Anisotropy and pores
 - Engineered composite
- Graphite behavior
 - Irradiation
 - Oxidation
- Some speculation on graphite behavior
 - Interesting irradiation mechanisms
 - Why graphite can not, will not burn
 - Molten salt interactions (a prelude)



The Nuclear Renaissance

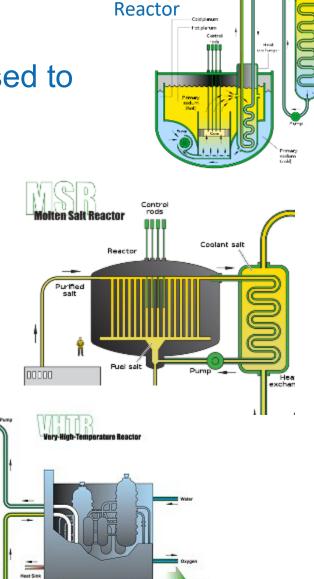
(Since about 2001) the term nuclear renaissance has been used to refer to a possible nuclear power industry revival: Lead-Cooled Fast Reactor

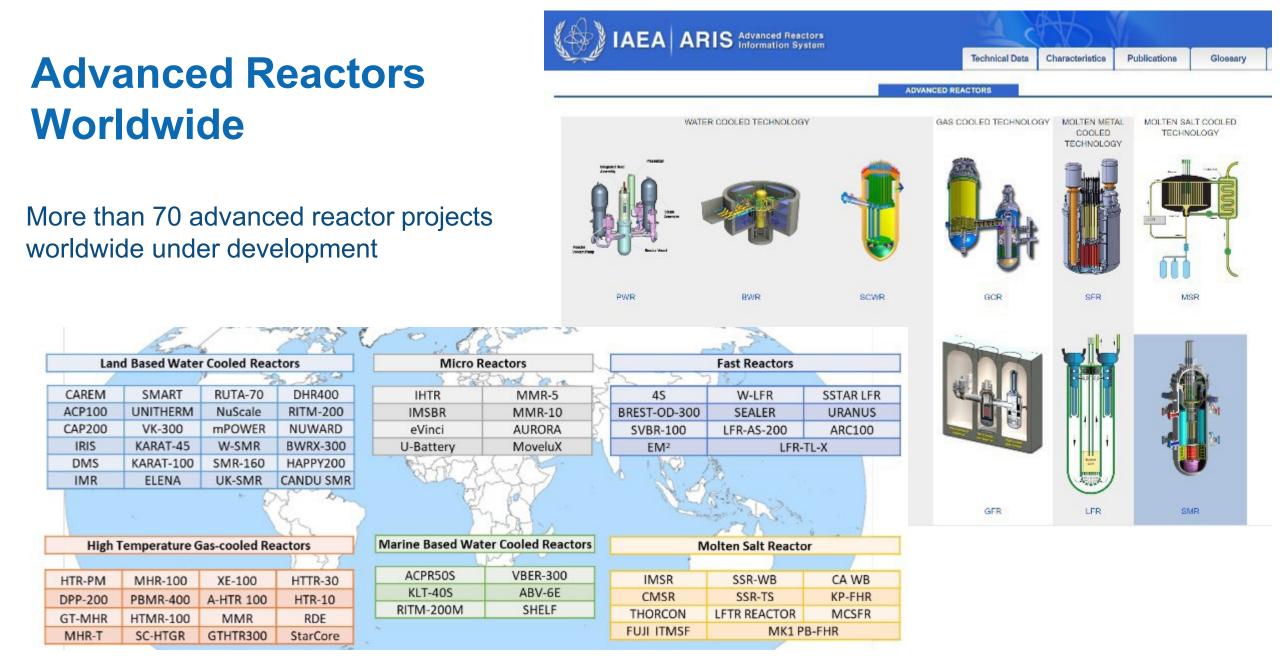
- Rising fossil fuel prices
- Limiting greenhouse gas emission

Generation IV reactor designs

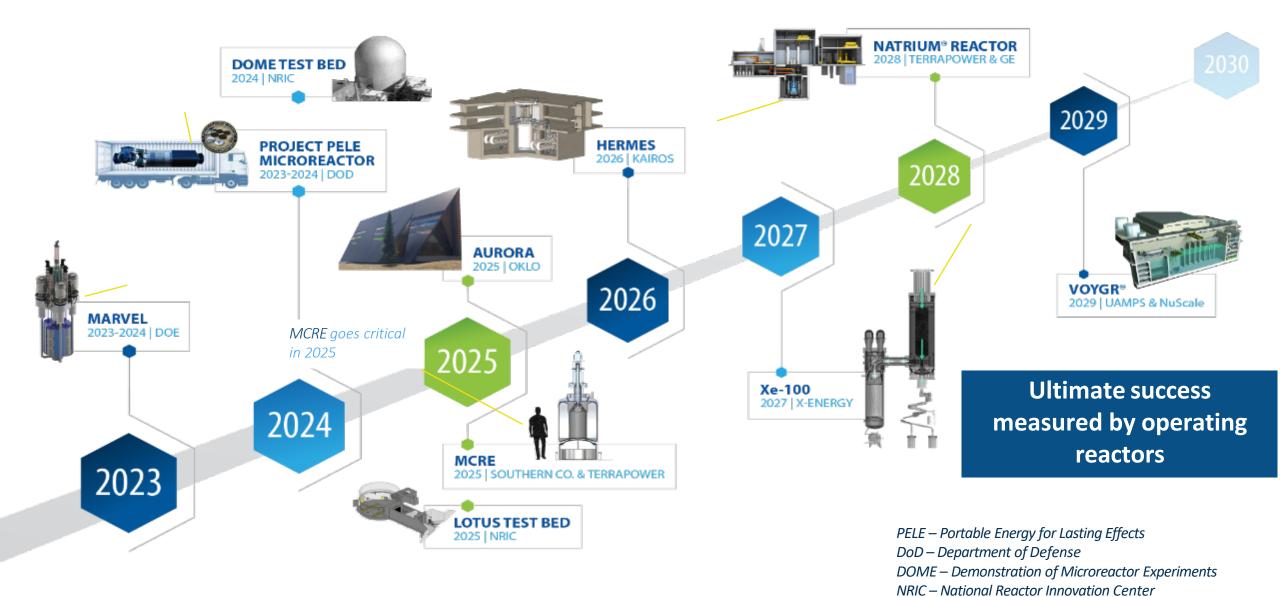
- Inherently & Passively safe
 - Natural shutdown and cooling from design
- New designs = new uses
 - Process heat
 - Small modular designs
 - Variety of coolants and fuels





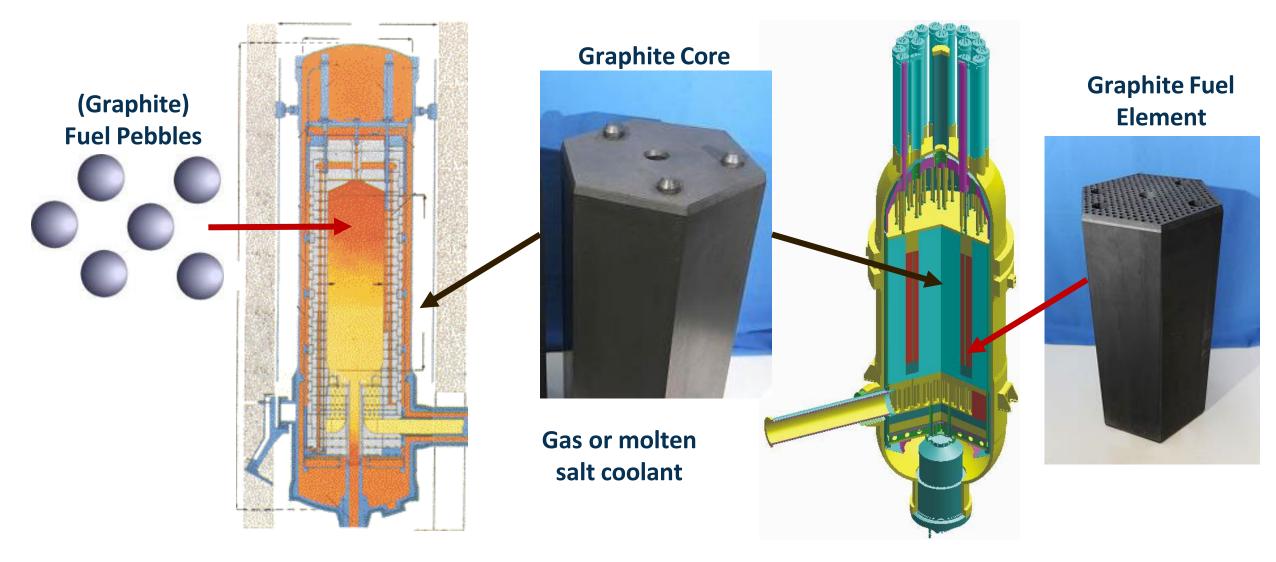


On schedule for deployment in next few years



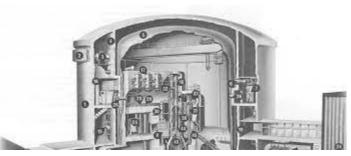
LOTUS – Laboratory for Operating and Testing in the U.S.

(Very) High Temperature Reactor (HTR)



Pebble Bed

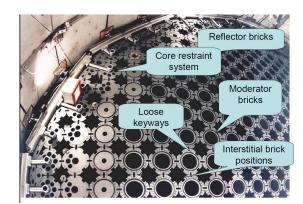
High Technology Readiness (Previous graphite Rx)



Dragon Reactor 1965 to 1976

THTR 1983 to 1989

Magnox Reactors 1956 to 2015

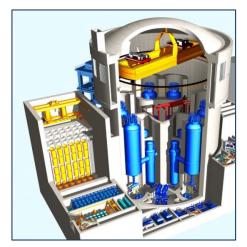


Advanced Gas Reactors 1967 to present

Peach Bottom 1966 to 1974

Fort St Vrain 1979 to 1989

AVR 1967 to 1988



HTR-PM 2022 to present

HTR-10 2003 to present HTTR 2002 to present

Renaissance is happening now

Coolant

Fuel type

Size

heat pipe

What started in 2001 as government funded R&D projects has evolved into numerous commercial enterprises world-wide

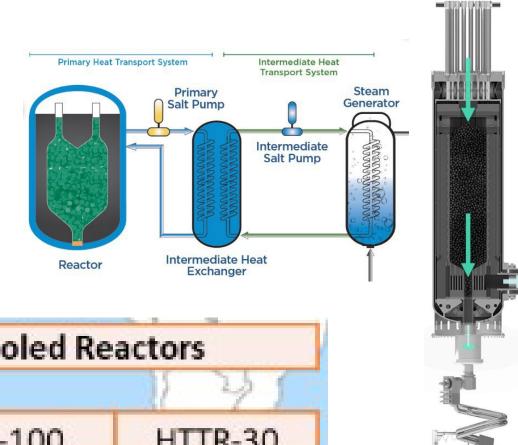


nergy

Graphite is important

(V)HTR: one of the 1st GEN IV Designs

- Highest technology readiness
- Graphite is important material for GEN IV design



| High Temperature Gas-cooled Reactors | | | | |
|--------------------------------------|---------------------------------|----------------------------------------------------|--|--|
| 10.9450 p.) 1 | | 1 235 | | |
| MHR-100 | XE-100 | HTTR-30 | | |
| PBMR-400 | A-HTR 100 | HTR-10 | | |
| HTMR-100 | MMR | RDE | | |
| SC-HTGR | GTHTR300 | StarCore | | |
| | MHR-100 PBMR-400 HTMR-100 | MHR-100 XE-100 PBMR-400 A-HTR 100 HTMR-100 MMR | | |

Graphite's unique structure key to behavior

Begins with benzene-like ring of carbon atoms – covalent bonds

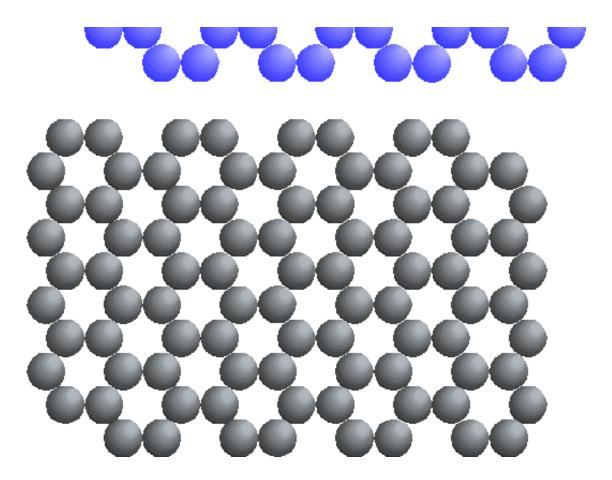
Graphene basal planes – a 2D "chicken-wire"





Stack up covalently bonded graphene planes

Basal planes stacked to form an ... ABAB... stacking structure



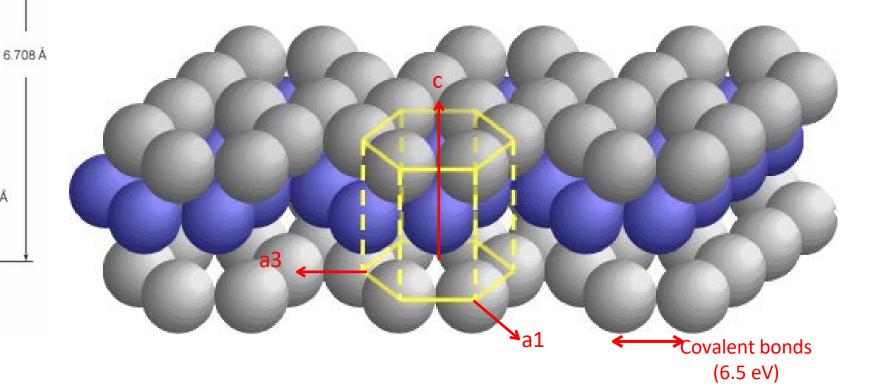
Graphite Crystal Structure

3.354 Å

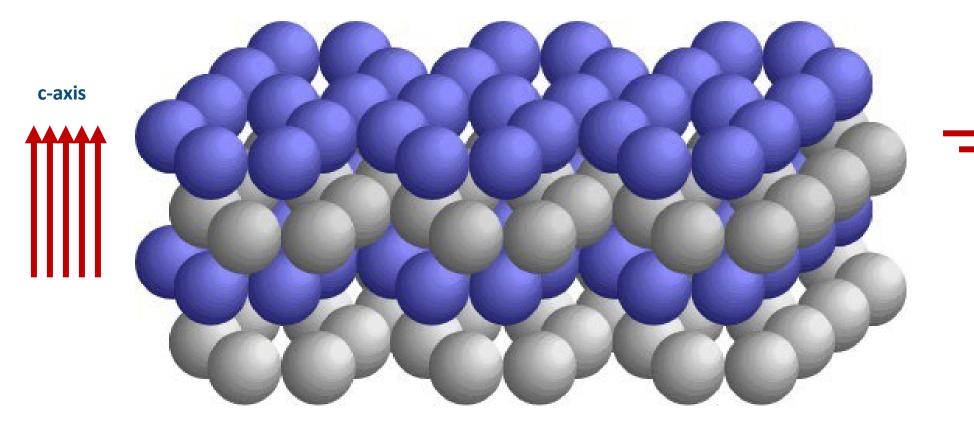
1.421 Å

(A)

- Hexagonal close pack (HCP)
 - A-B-A-B stacking of basal planes
 - Covalent within basal planes,
 - Van der Waals (electronic) between basal plane

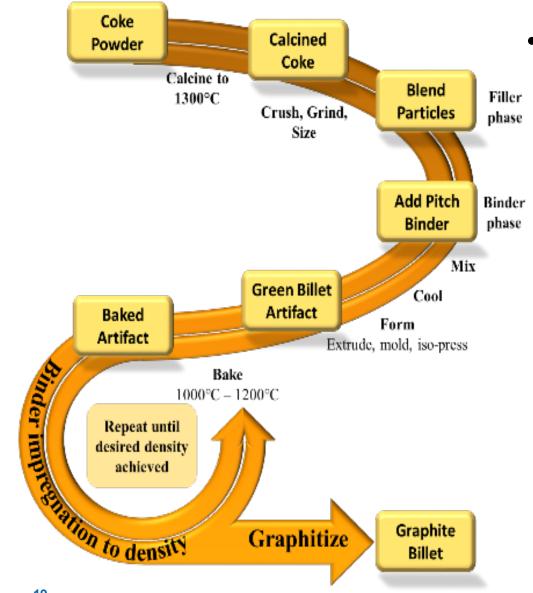


Completely anisotropic



- Huge differences between c-axis and a-axis directions
 - c-axis CTE ~ 26 x 10⁻⁶ /°K a-axis CTE ~ 1 x 10⁻⁶ /°K
 - c-axis thermal conductivity ~ 6.8 W/m K : a-axis conductivity ~ 2000 W/m K

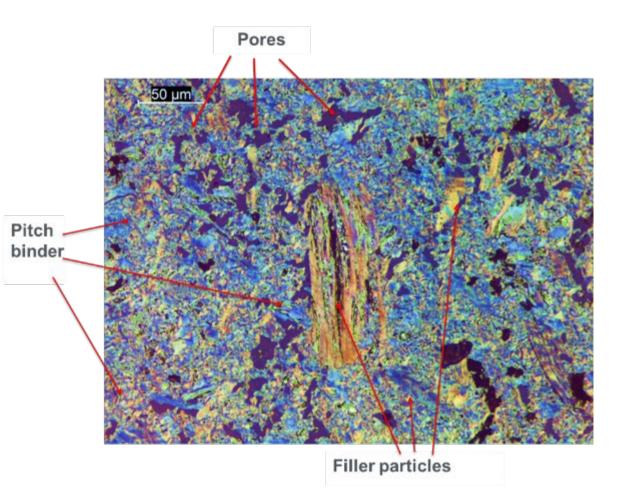
a-axis



What do we do? Make a composite

- Long fabrication process
 - <u>Filler particles</u>: petroleum coke/pitch coke
 - Grind filler particles to desired size grain size
 - Bind the particles: pitch-based liquid binder
 - Mix filler (grains) randomly in liquid binder
 - Form into final billet
 - Fabrication by extrusion, vibrationally molded, or isostatically molded
 - Baked green billet is heterogenous mix of filler particles bound by carbonaceous binder phase
 - Multiple pitch impregnations to increase density
 - Graphitize at >2200 °C
 - Most grades are heat treated to >3200 °C

Microstructure of synthetic graphite

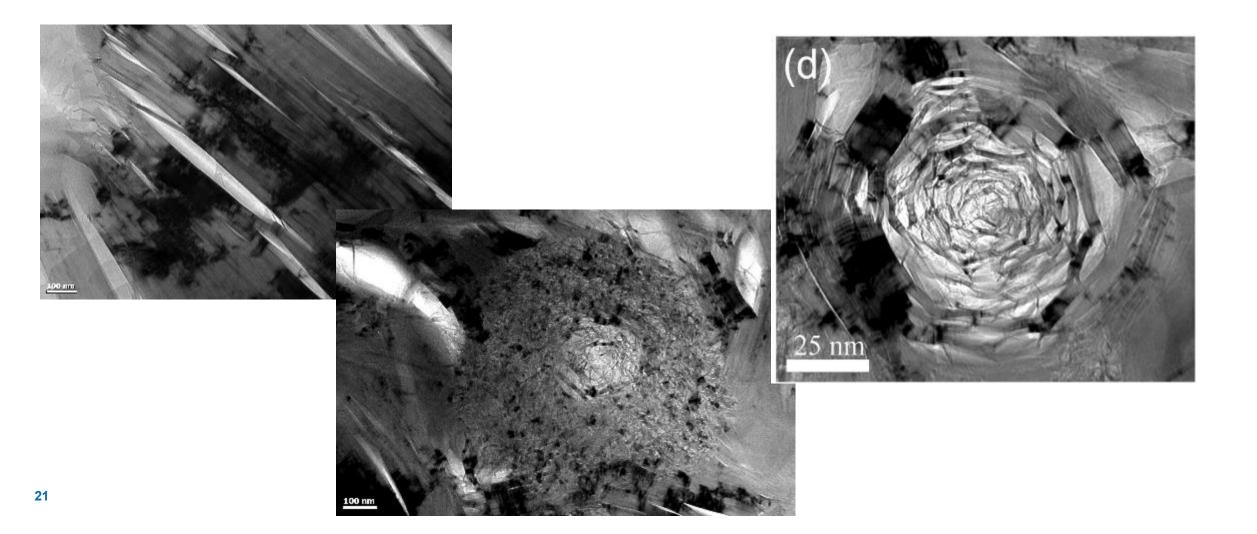


- Three phases in graphite microstructure
 - Filler phase
 - Binder phase
 - And Pore phase
- Nuclear graphite grades ~20% total porosity
 - Pores and pore structure define graphite behavior
 - Pore size range : nm to mm

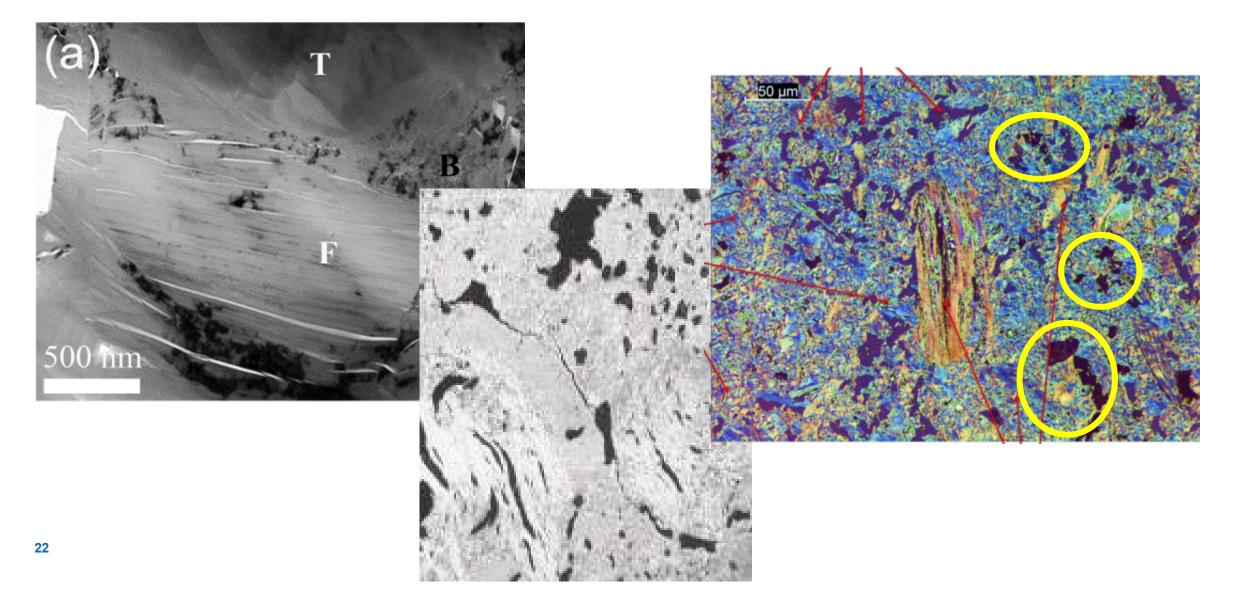
Must. Have. Pores

- More accurately : must have accommodating porosity
- Pores occur at all length-scales

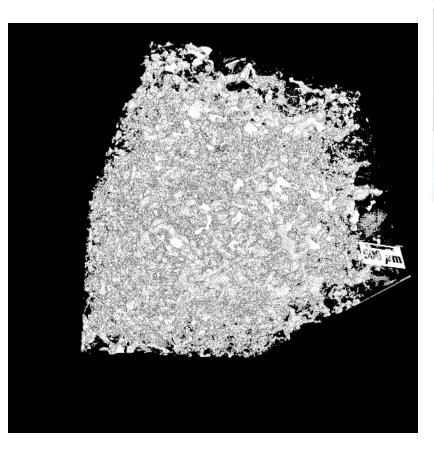
Crystal length-scale



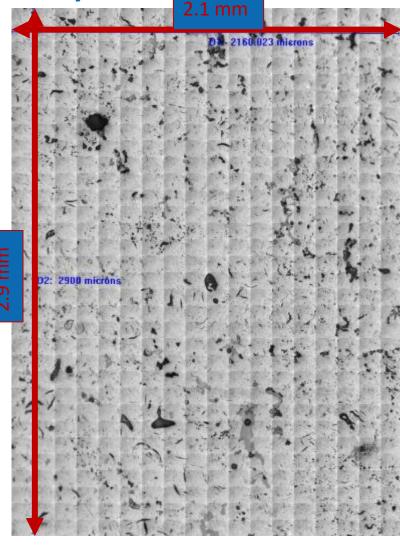
Grain-size length-scale



To the sample and <u>component level</u> (100X)







A "near" isotropic material

Material Properties

- Near-isotropic material response
- High thermal stability > 3000°C
 - Well above any accident temperatures
- High heat capacity (thermal sink)
- High thermal conductivity (better than metal)
- Density: 15% 20% porosity
- Purified graphite: Low activation (Medium waste)
- Chemically inert (Molten salt)
- Neutron moderator (thermal designs)
- Easy machinability / cheap material
- High compressive / Low tensile strength
 - Ceramic composites for tensile
- Ceramic like material response
 - Low fracture toughness (~ 1-2 MPa Vm)
 - Quasi-brittle cracking

Component Behavior

Decent irradiation response

- Smooth dimensional change
 - Life-limiting mechanism
 - Multiple decades of safe operation
 - And even longer at lower temperatures
- Generally gets stronger with irradiation
- Isotropy stays relatively constant
- Thermal stability and capacity are unaffected

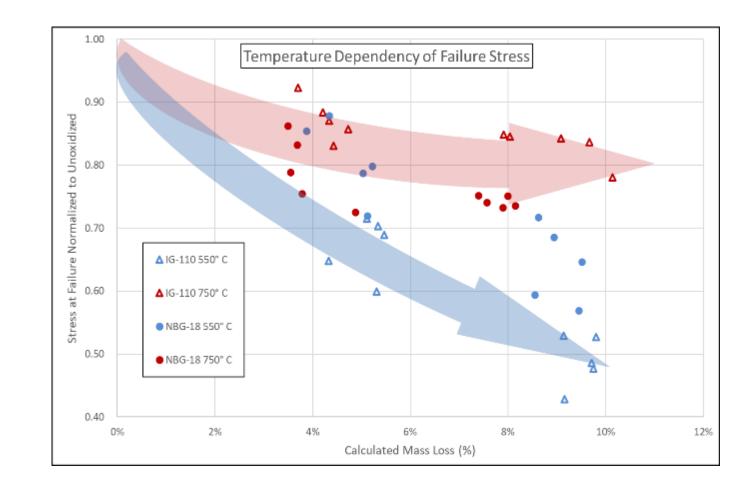
Oxidation and molten salt intrusion

- Graphite does oxidize at all temperatures But it does not burn!
- Oxidation and molten salt behavior depends on pore structure

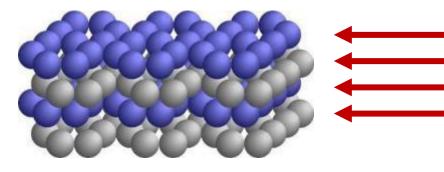
There is **no** "nuclear graphite" fabrication standard All grades are proprietary

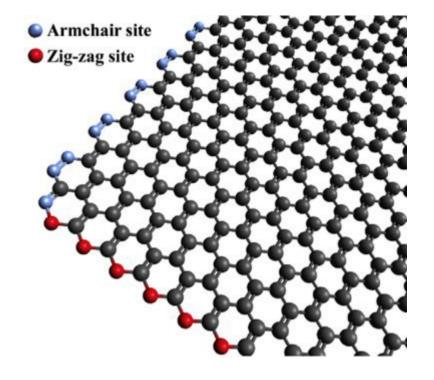
Degradation: Oxidation Behavior

- Internal Oxidation:
 - Air/oxygen into the pore structure
 - Oxidation rate is temperature dependent
 - Oxidation occurs at all temperatures
- Oxidation Effects
 - Internal oxidation has larger effect on residual strength
 - High temperature oxidation only attacks outside of component



Oxidation mechanisms





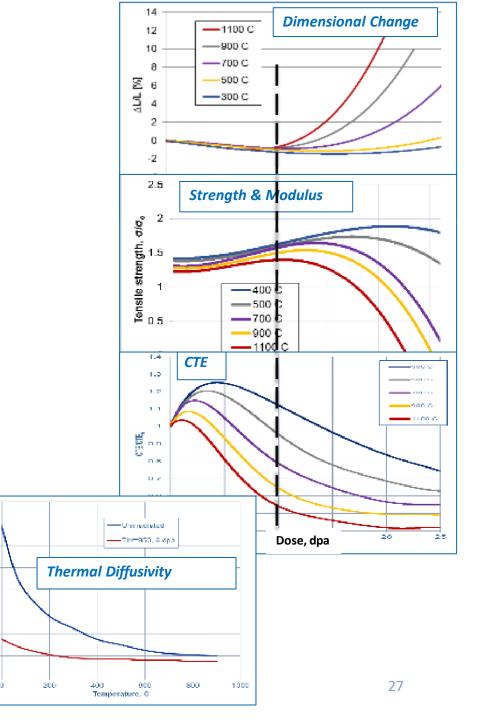
• Only on the Edges

- Oxygen chemical reaction occurs only on the edge of basal planes
- Covalent bonds within planes are too energetic
- RSA sites
 - Reactive Surface Area
 - Most chemically reactive
 - Fission product and other impurities end up here
- Can not sustain reaction
 - No fires!

Irradiation Behavior

Significant changes occur during normal operation:

- Density Densification
 - Graphite gets denser with irradiation until Turnaround dose
 - After Turnaround density decreases (volumetric expansion)
 - Formation of microcracks (molten salt consideration)
- Dimensional change
 - Turnaround dose is key parameter
 - Highly temperature dependent
- Strength and modulus
 - Graphite gets stronger with irradiation ...
 - Until Turnaround dose is achieved. It then decreases
- Coefficient of thermal expansion
 - Initial increase but then reduces before Turnaround
 - CTE is why properties are so temperature dependent
- Thermal diffusivity
 - Decreases immediately to ~30% of unirradiated values
 - At high temperatures it is same as unirradiated conductivity



144

120

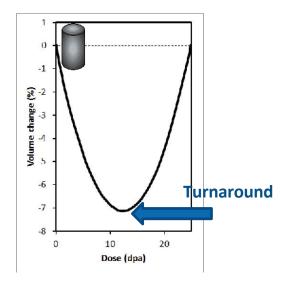
2 100

Diffusivity 8 8

Irr. Dimensional change \rightarrow life limiting mechanism

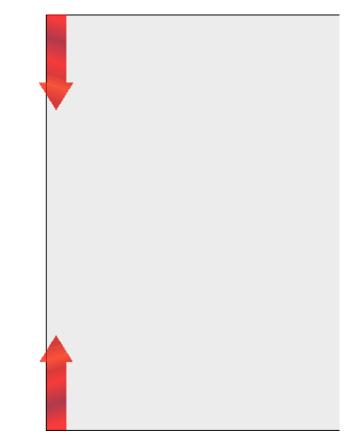
Internal stresses for crack formation develop from irradiation dimensional change

- Densification to volumetric expansion
 - As dose increases, dimensional change creates internal stresses
 - Temperature accelerates dimensional change
 - Internal stress buildup at interface of dimensional change
 - Changes after turnaround most limiting



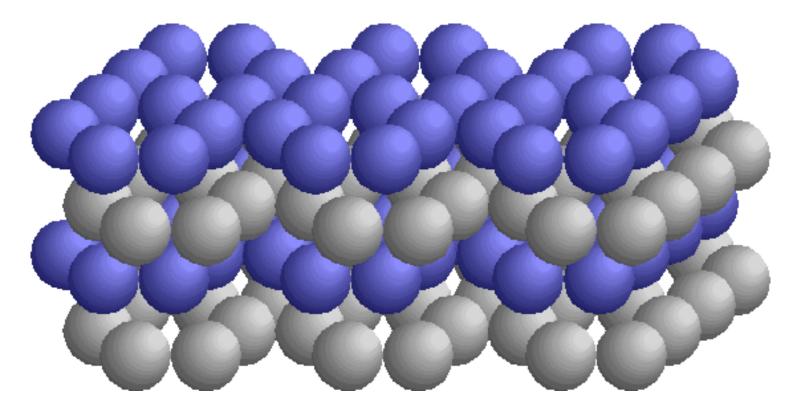


G. Haag, "*Properties of ATR-2E Graphite and Property Changes due to Fast Neutron Irradiation*", *Juel-4183, 2005*



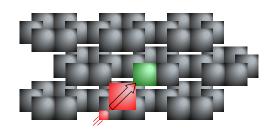
Mechanisms underlying irradiation damage

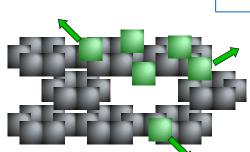
- Ballistic event physically displaces atoms for lattice position
- Sub-plane formation, vacancy clusters



Model describing irradiation damage

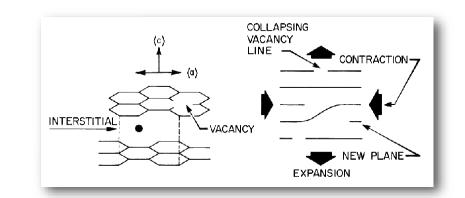
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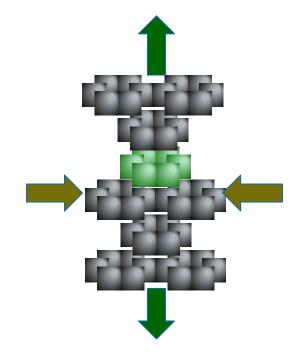


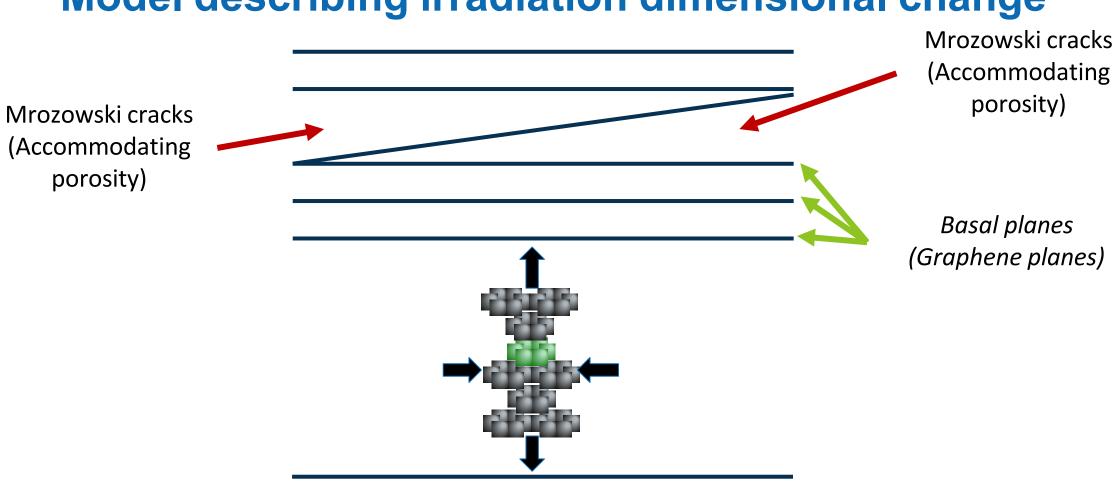


- Sub-plane formation between basal planes
- Vacancy cluster/loop collapse

- Crystallites shrink parallel to basal planes
- They expand perpendicular to planes

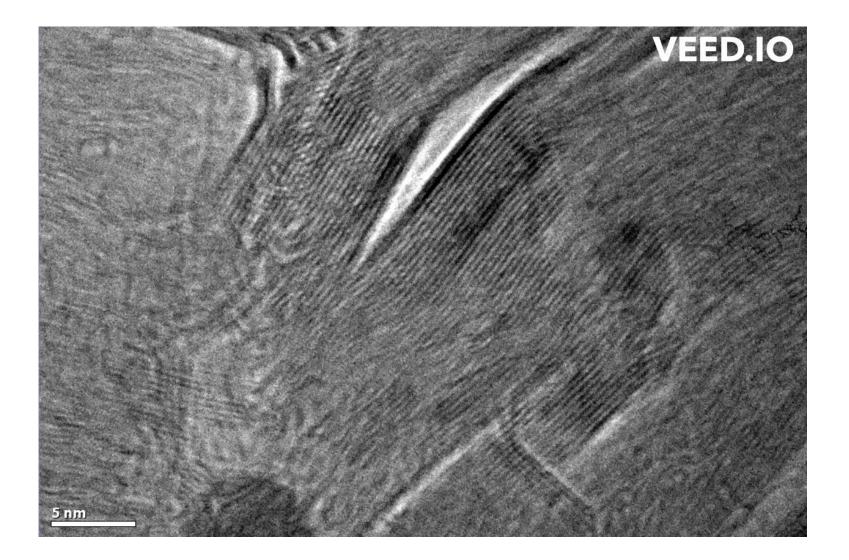




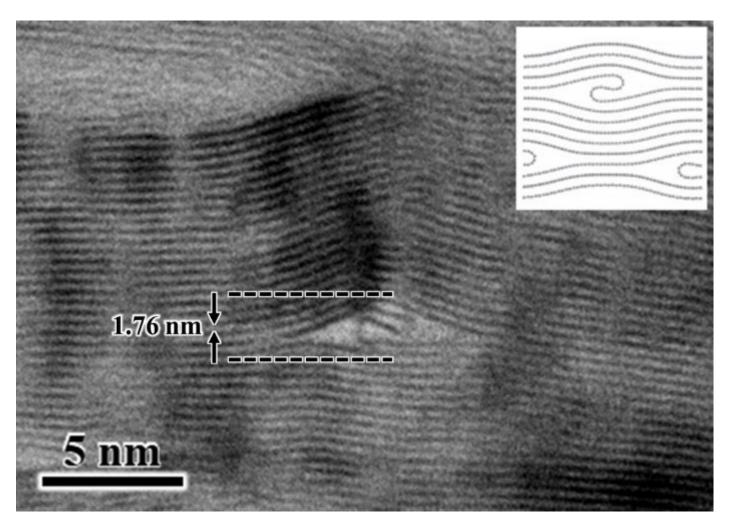


Model describing irradiation dimensional change

Observations: Crack closure



Observations: c-axis expansion



• Buckle, ruck & tuck

- Buckling of basal planes.
- Not sub-plane formation.

New techniques offer new data for mechanisms

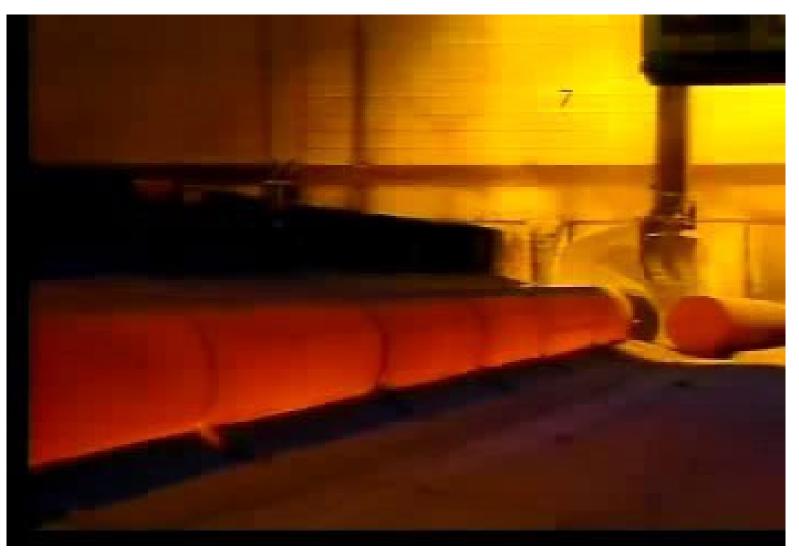
- Defect formations
- Dimensional change
- Property changes

 But we must understand limitations of new techniques

Graphite does not, can not burn



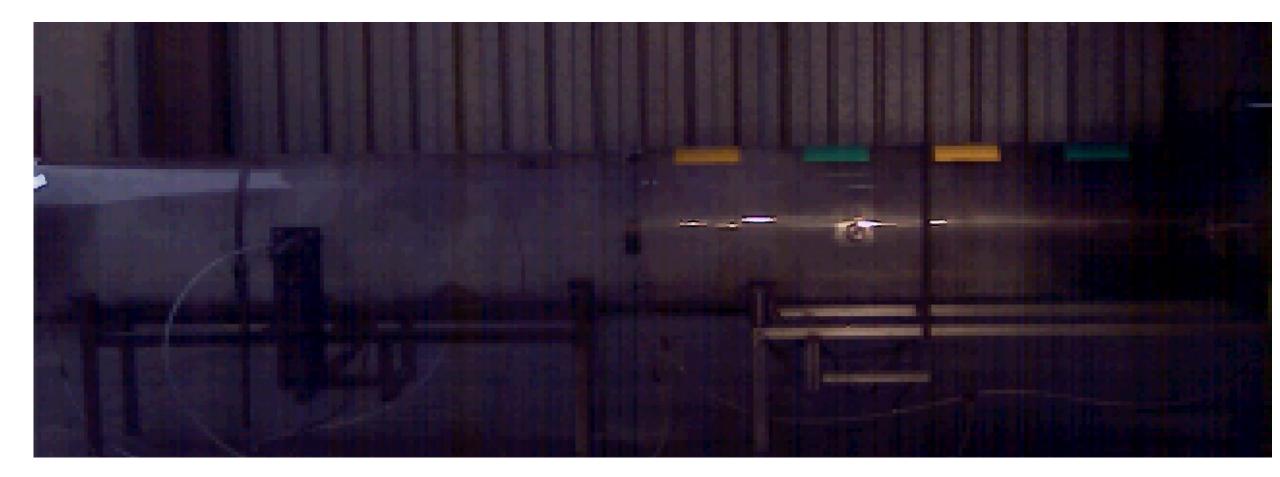
Graphite does not, can not burn



It oxidizes but can not sustain the reaction

- Only outer edges of crystals can react
 - Analogy: burning a thick paper book
- Reactive Surface Area (RSA) sites on the edges

Graphite "Fires" and "Explosions"



Maize Dust

Graphite "Fires" and "Explosions"





(Potential) Molten Salt Issues

Large molten salt tests are being initiated

- Salt impregnation into graphite pores
 - Physical damage/cracks
 - "Hot spots" from fueled molten salt
- Wear/abrasion/erosion
 - Molten salt has higher density than graphite
 - Liquid flow over soft graphite has potential
- Chemical coupling with metallic systems
 - Graphite MS is inert
 - Fluorination questions remain.
 - There are questions when a metallic component is added to the MS system



Dr. Nidia Gallego on 24 May 2023





Salt residue



Before

in FLiNaK

After immersion in FLiNaK

In summary

- Graphite is an important material for future GEN IV reactor designs
 - Carbon-based materials
 - Ceramic composites
 - Ceramics
- Much work still remains in characterizing nuclear graphite
 - Chronic oxidation behavior
 - Irradiation behavior
 - Molten salt interactions
 - New coolant and fuel systems for advanced reactor concepts

Idaho National Laboratory

Upcoming Webinars

| Date | Title | Presenter |
|--------------|---------------------------------------------------------------------------------|--------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| 24 May 2023 | Graphite-Molten Salt Interactions | Dr. Nidia Gallego, Oak Ridge National Laboratory, USA |
| 21 June 2023 | Panel Session: International Knowledge Management and Preservation of SFR | Joel Guidez, CEA (retired), France; Hiroki Hayafune, JAEA, Japan; Ron Omberg, PNNL, USA; Cal Doucette, ARC Energy, Canada; and Patrick Alexander, Terra Power, USA |
| 26 July 2023 | Off-Gas Xenon Detection and Management in Support of MSRs | Dr. Hunter Andrews, ORNL Dr. Praveen Thallapally, PNNL, USA |





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